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LIST OF ABBREVIATIONS

Ac	acetyl
Bn	benzyl
CI	chemical ionization
CuAAC	Cu catalyst azide-alkyne cycloaddition
DFT	density fuctional theory
DIAD	diisopropyl azodicarboxylate
DPPA	diphenylphosphoryl azide
DMF	<i>N</i> , <i>N</i> -dimethylformamide
EI	electron impact ionization
ESI	electrospray ionization
Et	ethyl
IR	infrared absorption spectrometry
i	iso
n	normal
t	tertiary
LHMDS	lithium hexamethyldisilazide
Me	methyl
MIP	2-methoxyisopropyl
MOM	methoxymethyl
Ms	methane sulfonyl
NMR	nuclear magnetic resonance
NOE	nuclear Overhauser effect
Nu	nucleophile
Ph	phenyl
Ру	pyridine
$\mathrm{R}f$	retention factor
SPAAC	strain-promoted azide-alkyne cycloaddition
T	temperature
TBAF	tetrabutylammonium fluoride
TBAI	tetrabutylammonium iodide
TBS	<i>t</i> -butyldimethylsilyl
Tf	trifluoromethanesulfonyl
TFA	trifluoroacetic acid

(011D)	tetrahydrofuran
THF'	thin layer chromatography
TLC	trimethylsilyl
TMS	<i>p</i> -toluene sulfonyl
Ts	1 5

Chapter 1 Introduction

1.1 Synthesis of 1,2,3-triazoles

1,2,3-triazoles have been studied for over a century as the important heterocyclic moiety and still attracted considerable attention due to their numerous applications. The conventional methodology to produce 1,2,3-triazole is [3+2] Azide-Alkyne Cycloaddition (AAC).

The first synthesis of 1,2,3-triazoles was demonstrated with diethyl acetylenedicarboxylate and phenyl azide reported by A. Michael in 1893.¹ In 1960s, Huisgen significantly pioneered the thermal conditions (Scheme 1). However, due to the high activation energy ($\Delta G = +26$ kcal/mol), these cycloadditions require elevated temperature and long reaction time to accelerate the reaction (80~120°C for 12~24h), affording a mixture of 1,4 and 1,5 regioisomers (Scheme 1).²







In the past decade, this cycloaddition reaction has been extensively developed to lower the activation barrier, since the copper-catalyzed conditions were reported. In 2002, the groups of Sharpless³ and Meldal⁴ independently developed the Cu(I)

catalysts dramatically accelerate the reaction of terminal alkynes with azides, to give the 1,4 disubstitutied triazoles regioselectively under mild conditions (Scheme 2).



Scheme 2. Highly regioselective Cu catalyzed azide-alkyne cycloaddition reaction.

The proposed mechanism of CuAAC reaction has been investigated by many groups over the past decade, but still has not yet been completely proven, especially the complexation of the Cu(I)-species and the origin selectivity of the cycloaddition are still unknown. The copper catalysts in these transformation are Cu(I), in addition, Cu(II) salts can also be used as pre-catalysts which can be easily reduced to Cu(I) in the present of reducing agents (sodium ascorbate). Moreover, addition of base can significantly accelerate the transformation under Cu(I) catalyst conditions, comparing with Cu(II) case that no effect of base addition was observed.



Scheme 3. Proposed mechanism for CuAAC reaction by kinetic studies.

The mechanism of the CuAAC reaction has been investigated by kinetic studies and DFT calculation.^{2,5-8} At first, in the present of base, the important intermediate Cu(I) acetylide **1A** (Scheme 3) was formed from Cu(I) and alkyne **1A**. The step 2 is the coordination of azide **1A** and to the Cu(I) acetylide to form the complex **6A**. It is suggested that complex **5A** can be presented as **5AA** or **5AB** that alkynyl **2A** and azide **1A** were coordinated to one Cu center or not. Structure **5AA** indicates that the Cu(I) acetylide **4A** and azide **1A** coordinate to just one copper atom. On the other hand, structure **5AB** demonstrates that the coordination of the substrate to different copper atoms is possible. Through C-N bond formation to give the triazolide **7A**, finally 1,4-disubstituted 1,2,3-triazole **3A** is produced by protonolysis with the recovery of the copper catalyst.



Scheme 4. Proposed mechanism for CuAAC reaction with reactive σ -acetylides.

In the case of dipolar cycloadditions of 1-halo and 1-metalloalkynes with organic azides suggested that the copper catalyst effects the cycloaddition reaction through π -interactions with the formally internal alkyne (Scheme 4).^{8,9} At first, the copper may

activate the reactive σ -acetylides through formation of π -complex intermediate **4B** to which organic azide coordinates leading complex **5B**. Then nucleophilic attack at N of the azide by β carbon of the acetylide forms the C-N bond, producing intermediate **6** which results the ring closure product **3B**.

Fokin, Jia and Sharpless 2005



Scheme 5 Ru-catalyzed Azide-Alkyne cycloaddition reaction.

In 2005, Fokin, Jia and Sharpless demostrated the synthesis of 1,5-disubstituted 1,2,3-triazoles (Scheme 5).¹⁰ These cycloadditions of terminal alkynes with organic azides were carried out in the present of Cp*Ru-catalyst forming 1,5-disubstituted triazoles in good yield with high regioselectivity, complementary to CuAAC producting 1,4-disubstituted triazoles.

Due to the high efficiency, this CuAAC reaction has been extensively developed and applied by many groups. Thus, Yamamoto's group reported the cycloaddition reaction with trimethylsilyl azide (TMSN₃) to prepare the *NH*-triazoles in good yields (Scheme 6).¹¹ The reaction proceed through the formation of Cu-acetylides followed by a cycloaddtion reaction with the in situ-formed hydrazoic acid. This protocol allows the preparation of a lot of substituted triazoles in good yields. Yang's group developed the Cu(I)-mediated cycloaddition reaction of terminal alkynes and sodium azide (NaN₃) to give the corresponding triazoles in good to excellent yields (Scheme 7).¹² More recently, Kuang and co-workers demonstrated that this CuAAC reaction of terminal alkyne and sodium azide could proceeded in the present of catalytic amount of Cu(I) and base (Scheme 8).¹³



Scheme 6. CuAAC reaction with TMSN₃ by Yamamoto's group.



Scheme 7. CuAAC reaction with TMSN₃ by Yang's group.



Scheme 8. CuAAC reaction with TMSN₃ by Kuang's group.

These metal-catalyst variant of the Huisgen [3+2] cycloaddition have made an

enormous impact in chemical discipline, ranging from drug discovery, chemical biology, materials science, development of sensors, polymer chemistry to nanotechnology. In 2004, Fréchet and co-workers reported the application of this CuAAC reaction using poly(vinylacetylene) and dendritic benzyl azide for (Scheme 9).¹⁴



Scheme 9. Application of CuAAC reaction for polymer chemistry.

CuAAC also has been utilized to modifications of activity and selectivity of existing pharmaceuticals and natural products, and compatible with the complexity of these compound classes. Based on replacement of the central part of known HIV inhibitors, Brik et al. synthesized the novel compounds through CuAAC reaction, in which the triazole actively participated in the crucial binding to the water nucleophile (Figure 2).^{15,16} In vitamin D synthesis,^{17,18} advantage has been taken of the exquisite orthogonality and specificity of the of the triazole coupling and the triazole has been formed at last step of synthesis in the presence of unprotected functional groups. Under the standard CuAAC condition, the desired cycloaddition products were

isolated in good yields (70~100%) (Figure 3).



Figure 2. Chemical structure of HIV-1 protease inhibitor.



Figure 3. Chemical structure of Vitamin D dimer.



Figure 4. Reagent development for Copper-free AAC reaction.

Despite the power and versatility of CuAAC reaction, the requirement for the toxic

copper severely limits its application in cellular systems. And the scope of reaction materials is also limited due to the potential for residual traces of copper.¹⁹ Thus, many groups have developed the mild, rapid and Cu-free triazolations.

As an catalyst-free alternative approach to lower the activation barrier for cycloaddition, Bertozzi, Boons and Alabugin, harnessing the reactivity of activated cyclooctynes (OCT), have developed strain-promoted azide-alkyne cycloadditions which has been well-utilized in chemical biology (Figure 4). In 2008, Bertozzi reported the synthesis of 6,7-dimethoxyazacyclooct-4-yne (DIMAC) which is biocompatible and feasible for detecting azide-labeled biomolecules via copper-free azide-alkyne cycloaddition reaction (Scheme 10).²⁰ During the past years, by Boons group the dibenzocyclooctynes (DIBO) has been tested as high potential compounds for copper-free azide-alkyne bioconjugations (Figure 4).



Scheme 10. DIMAC for biolabeling study via copper-free AAC reaction.

Moreover, Moses, Larock, Reddy, and Feringa reported separately that the [3+2] cycloadditions of benzynes and organic azides under mild conditions to form

benzotriazoles rapidly. Benzynes which can be formed from, for example, trimethylsilylphenyl *O*-triflate through fluoride-promoted *o*-elimination, reacted with various azides to afford the functional benzotriazoles (Scheme 11).²¹



Scheme 11. Synthesis of benzotriazoles from benzynes and azides.

More recently, Garg and co-workers demonstrated that the cycloadditions of cyclohexynes and the more elusive intermediate, cyclopentyne, with organic azide to construct the new heterocyclic compounds (Scheme 12).²²



Scheme 12. Cycloaddition of Cyclohexyne and Cyclopentyne.

In additional, Curran, Maas, and others also reported that C-C triple bond activated by a carbonyl can accelerate the reaction,²³ affording the corresponding triazoles under mild conditions compared with general alkynes (Scheme 13).

$$R-N_3 + \bigvee_{R'}^{O} = R'' \xrightarrow{r.t. \text{ or heating}} \bigvee_{R''}^{N'N'} R''$$

. .

Scheme 13. AAC activation by carbonyl furnishes.

However, metal-catalyst AAC are limited to mostly terminal alkynes (Scheme 14) and the toxicity of copper salts limits the utility for the in vivo applications. With the increasing of the reactivity, cyclooctyne became unstable that should be stored as a solid at 0 °C protected from light and oxygen (BARAC, Figure 4).²⁴ The lack of regioselectivity in the cycloadditions was also a disadvantage of strain-promoted azide-alkyne cycloaddtion reactions, preventing its application in drug design and peptidomimetics. Moreover, most of the reported methods require ambient or elevated temperature with long reaction time.



Scheme 14. Strategies of Azide-Alkyne cycloadditions.

Therefore, I was interested in development of the novel rapid triazole synthesis which can allow both internal- and terminal-alkyne to afford fully substituted 1,2,3-triazoles and multicomponent coupling reactions between ambient and low temperature.

1.2 Organic azides

Organic azides are popular functionalities in organic chemistry, and in 1864 the first organic azide, phenyl azide, has been synthesized by Perter Grie β .^{25,26} After that, in more than 150 years, numerous applications of these energy-rich molecules have been investigated,²⁷ such as Curtius rearrangement that provide the isocyanates from appropriate acyl azides,²⁸ Staudinger reaction,²⁹ Schmidt reaction,^{30,31} Huisgen reaction,^{32,33} and so on.



Figure 5 Reactivity of organic azide.

Organic azides involve three zwitterionic nitrogens and prove very different chemical reactivities (Figure 1). In principle, organic azide can work as a nucleophile at the N1 atom and as an electrophile the N3 atom. With the three zwitterionic nitrogens, organic aizdes also work as 1,3-dipolar, reacting with dipolarophile for cycloaddition reaction. In addition, nitrenes can be produced from organic azides under thermal conditions or photoirradiations, and be utilized to aziridinations and C-H aminations.

1.2.1 Reactions as an electrophile

In 1919, Staudinger and Meyer reported the Staudinger reduction using phenyl azide and triphenylphosphine as procedure. This reaction involves the construction of iminophosphoranes which are important reagents and intermediates for variety of organic nitrogen compounds (Scheme 15).³⁴





In the present of water, this iminophosphorane can be hydrolyzed to the corresponding primary amine with formation of phosphine oxide; with carboxylic acid the aza-ylide can be converted to *N*-substituted amindes; and also condensation with

acyl halides can generate imydoyl halides. Moreover, the reaction of iminophosphoranes with various carbonyl compounds, called aza-Wittig reaction, has been frequently used for produce imines.³⁵⁻³⁸ The product of the reaction is Schiff base. Due to its high synthetic potential, this methodology has been received considerable attention in the past decade for the generation of C=N bond containing heterocycles compounds.¹⁵ In particular, the intramolecular aza-Wittig reaction is one of powerful tools for the prepartation of five-, six-, seven-, and eight-membered heterocycles.³⁹⁻⁴³



Scheme 16. Total synthesis of Mycosporin-Gly by Staudinger reaction.

In 1989, Staundinger reaction had been utilized to the enatioselective total synthesis of mycosporins (Scheme 16).⁴⁴ To elaborate the side chain, the cyclic vinyl azide was first converted to a stable vinyl iminophosphorane followed by reaction with benzyl glyoxylate to afford the Schiff base which was promptly reduced by sodium

cyanoborohydride.

The first total synthesis of (-)-benzomalvin A was achieved by Eguchi's group (Scheme 17). Both of the heterocycle skeletons were efficiently prepared by intramolecular aza-Wittig reaction.^{45,46} Starting with L-phenylalanine, the azide derivative was reacted with tributylphosphine to generate the iminophosphorane followed by the intramolecular aza-Wittig cyclization to give the seven-member ring. The endgame of the synthetic route was accomplished by another intramolecular aza-Wittig cyclization reaction.



Scheme 17. Total synthesis of (-)-benzomalvin A with aza-Wittig reaction.

1.2.2 Nitrogen elimination-initiated reactions

Under thermolysis or photolysis conditions, azides can easily release nitrogen gas, and this promotes various rearrangements and addition reactions.^{47,48} Nitrenes are one of chemical species produced under these conditions. Although nitrenes are related to carbenes, they still have different properties.⁴⁹ The reactions of nitrenes were ranged from cycloaddition, rearrangement to insertion reactions.

The Curtius rearrangement reaction is decomposition of acyl azides to afford the corresponding isocyanates (Scheme 18). If the generated isocyanate is treated with water, primary amine can be formed. When the reactions are carried out in the present of amines, the corresponding urea derivatives can be obtained. With alcohol, the isocyanates are converted to carbamate compounds.



Scheme 18. Curtius rearrangement reaction.

The total synthesis of *anti*-cancer alkaloid *trans*-dihydronarciclasine isolated from the Chinese medicinal plant *Zephyranthes candida* (Scheme 19) was achieved using Curtius rearrangement reaction.⁵⁰ Acylazide was generated from carboxylic acid with diphenylphosphoryl azide (DPPA) and the producing isocyanate was generated with *tert*-butanol to give Boc-amine product. Through this reaction, the generated benzylic carbocation was trapped with the amide to constract the tricyclic compound, which was then transformed to the desired (+)-*trans*-dihydronarciclasine.



Scheme 19. Total synthesis of (+)-trans-dihydronarciclasine.

1.2.3 Reactions as a nucleophile

Normally, organoazides can react with suitable electron-deficient compounds (carbon electrophiles, protons, and boranes) to produce amine-substituted diazonium ions, which easily lose nitrogen. In 1923, Schmidt reported that under thermal conditions the reaction of hydrazoic acid with benzophenone produced the benzanilide.⁵¹ This reaction was found to generality for ketones, aldehydes, and carboxylic acid, that undergo with alkylazides followed by rearrangement and loss of

nitrogen in a concerted manner to give amides (Scheme 20, eq 1), nitriles (eq 2), imines

(eq 3), and iminium ions (eq 4) in the presence of acids, respectively.^{27,52}



Scheme 20. Schmidt reactions.

Since the intermolecular reactions end in low yields, Schmidt reaction has usually been demonstrated as intramolecular reactions with aliphatic azides and carbonyl compounds to provide lactams (Scheme 9).⁵³ Aubé⁵⁴ and Pearson^{52,55,56} develop this intramolecular reaction and reported that Lewis acids can accelerate the Schmidt reaction and organic azides can react with ketones, which is also named Boyer reaction.⁵⁷ Besides ketones, azidoalkyl-substituted epoxides or carbenium ions can be transformed in an intramolecular reaction (Scheme 21).⁵⁸ In the presence of Lewis acid, aliphatic azides efficiently reacted with ketones in good yields to obtained *N*-alkynlated amides or lactams, but it is limited to aliphatic ketones.



Scheme 21. Boyer reaction with epoxide.

Despite Schmidt reactions require generation of carbocation, utilization of more reactive allyl cations was limited due to the difficulty of reaction control. Our group had developed the application of allyl cation generated from allylic alcohols and derivatives reacted with organic azide to provide α,β -unsaturated imines which have been difficult to prepare because of their instabilities (Scheme 22).⁵⁹ Using this novel strategy, total synthesis of Costa Rican ant venom alkaloids **A** and **B** possessing conjugated imine moieties were achieved.^{59,60} The allyl/pentadienyl cation-mediated Schmidt reactions were set as the key steps at the last stag of syntheses, affording α,β - and $\alpha,\beta,\gamma,\delta$ -unsaturated imine alkaloids.



Scheme 22. Total synthesis of Costa Rican ant venom A and B.

1.3 Design of synthetic route

Recently, our group has developed the novel method to contract the unsaturated cyclic imines with allyl cations from allyl alcohols under acid condition with organic azides (Scheme 23).⁵⁹



Scheme 23. Allylic cation-mediated intermolecular cyclization reaction.

Base on the same conjugated carbocation chemistry, I design the cyclization reaction of the propargyl cations and organic azides that the carbon-carbon triple bond is employed instead of double bond to afford the functional fully substituted 1,2,3-triazoles without metal catalysts under mild conditions (Scheme 24).

In the presence of acid, propargyl cations prepared from corresponding alcohols reacted with azides to generate the allenylaminodiazonium intermediates which have high potential enough to form triazole rings immediately. To the best of my knowledge, these diazonium compounds have not been reported, despite it is known that the triazoles can be generated from allenyl azides cyclization.⁶¹ I expect that since these high reactive species such as propargyl cations and diazonium intermediates can achieve rapid transformations even at ambient temperature. The trisubstituted triazoles functionalized with additional nucleophiles can be obtained.



Scheme 24. Propargyl cation-mediated cyclization reaction.

Chapter 2 Carbocation-Mediated Azide-Alkyne Cycloaddition

2.1 Intramolecular [3+2] azide-alkyne cyclization reaction

To realize my strategy to make fully substituted triazoles under mild condition, intramolecular [3+2] cyclization reaction can be investigated at first (Scheme 25). Under the acid conditions after performing the propargyl cation, the azide anion attached to the sp carbon C3 affords the unstable intermediate allenylaminodiazonium through pathway A. Through this intramolecular cyclization reaction, the bicycle triazoles can be obtained smoothly.



(undesired)

Scheme 25. Preparation of propargyl alcohol.

As the other pathway, enone products by Meyer-Schuster rearrangement can also be obtained through pathway B. Attack of a water molecule on the carbocation followed by tautomerization prior to the attack of azides would give the α,β -unsaturated carbonyl compounds (Scheme 25).

As alkyl propargyl alcohols required strong acid, phenyl propargyl alcohol **4** was chosen in order to use weaker acid. General procedure of preparation is shown in Scheme 26.



Scheme 26. Preparation of propargyl alcohol.

With 1.2 equiv of TsOH·H₂O,⁵⁹ triazolyl alcohol compound **5a** was generated from propargyl alcohol **4a** at ambient temperature in 32% (entry 1, Table 1). On the other hand, TMSOTf could afford the desired triazolyl ketone **5ab** in 13% with an unexpected vinyl triflate compound **5a**' in 7%, which was confirmed by X-ray crystallographic analysis. The O-triflate group seemed to be derived from TMSOTf. Using TsOH·H₂O, no vinyl tosylate product was observed. TfOH and BF₃·OEt₂ only gave complex mixture.

Table 1. Investigation of intramolecular [3+2] with acids



With this optimized condition in hand, I turned attention to the scope of this intramolecular [3+2] cyclization with substrates (Table 2). Considering the stability of carbocation, electron donating groups were mainly introduced to improve the yield. With electron donative substrates, the corresponding triazoles were obtained in good yields (entry 1-3). On the other hand, reducing the electron density of phenyl group, the yield of triazoles went down (entry 4-5). Because of the stability of carbocation, the alkyl propargyl alcohol did not afford any products, and only starting material was recovered (entry 6).



 Table 2. Scope of intramolecular [3+2] cyclization

^{*a*} With dichloroacetic acid. ^{*b*} With chloroacetic acid.

Moreover, I found that introducing nucleophilic function groups on appropriate position of phenyl moiety could provide bicyclic compounds through intramolecular substitution at α -position of imines. With prepared ortho-methoxy or ortho-hydroxy compounds, benzofurans **6h**, **6k** and **6l** were generated along with triazole compounds **5h**, **5k** and **5l** by TsOH·H₂O (entry 7, 10 and 11). With more electron donative substrates, only corresponding triazole products were observed.

Furthermore, to improve the yield of triazoles, tertiary propargyl carbocations which are more stable and active than secondary carbocations were investigated for the cyclization reaction (Table 3). With methyl phenyl propargyl alcohol (7a), under the standard condition the corresponding triazole product 8a was obtained in 54% with dehydroxyl triazole product 8a' in 19% (entry 1). Due to the decreasing activity and stability of carbocation, with alkyl propargyl alcohol the dehydrated triazole product 8b' was obtained in a moderate yield with dehydrated starting material 10 (entry 2). Then I continued to increase the reactivity, herein diphenyl propargyl alcohol was tested. Using diphenyl substrate 7c, corresponding triazole product 8c was given in a good to excellent yield. Even with weaker acid TFA, triazole 8c was generated in 98% at room temperature in 30 min. Moreover, under -90 °C, the corresponding triazole was successfully produced in 91% with TMSOTf (entry 3). The same transformation toward 6-member ring were successfully completed with TFA and TsOH·H₂O at room temperature, and even -90 °C within 5min.

		Reagents (1 CH ₂ Cl ₂ then sat. Na	.2 eq) HO− HCO ₃ aq.	N ^N N M R			
7a-d , n = 1,2 8a-d , n = 1,2							
Entry	Ar	Reagents	Conditions	Y	<i>'ields</i>		
				8	Byproduct		
1	Me OH 55 7a, n=1	TsOH·H ₂ O	rt, 30 min	54%	8a' (19%)		
2	OH 55° 7b, n=1	TsOH·H ₂ O	Reflux, 30 min	0	8b' (50%) 10 (38%)		
3	Ph Ph S 7c , n=1	TFA TsOH·H2O TMSOTf	rt, 30 min 0 °C, 2 h -90 °C, 5 min	98% 99% 91%	no		
4	Ph Ph S S 7d, n=2	TFA TsOH·H2O TMSOTf	rt, 30 min rt, 20 min -90 °C, 5 min	92% 90% 96%	no		
	CH ₂ N=N 8a'	38	N=N (N ₃ 10			

Table 3. Intramolecular [3+2] cyclization with tertiary propargyl alcohols

On the other hand, further development of this azide-alkyne [3+2] cyclization reaction, double intramolecular cyclization was investigated with dialkyne substrates. At first the dialkyne substrates were prepared as following procedure. The starting material **12a-c** could be easily prepared from the coupling reaction of tosylated pentynol with corresponding carbonyl compound followed by azidation using NaN₃

(Scheme 27).



Scheme 27. Preparation of bialkyne propargyl alcohols.

Table 4 shows the investigation results of double cyclization reactions. Due to instability of the carbocation, **12a** gave **13a** only in the yield of 13% (entry 1). With alkyl bialkyne substrate **12b**, at ambient temperature triazole compound was obtained in 37% with dehydrated starting material **14** in 11% using TsOH·H₂O. Employing stronger Lewis acid TMOTf, the yield of desired triazole was increasing to a moderated yield and the byproduct was observed in the yield of 10% (entry 2). In entry 3, phenyl dialkyne compound **12c** was studied with TsOH·H₂O at room temperature, and the double cyclization was completed in a good yield in 10 min, because of increased activity of carbocation.



Table 4. Intramolecular double [3+2] cyclization with bialkyne substrates

These intramolecular [3+2] cyclization reactions of azide with internal alkyne were highly accelerated by the propargyl cation effiently. According to the above results, the reaction possible mechanism is described in Scheme 28. Formation of the generated propargyl cations from corresponding alcohols under acid condition, after azide anions attached to the sp carbon, the unstable diazonium intermediates were produced. Then this active allenylaminodiazonium intermediates could be immediately transformed to the desired triazoles. On the other hand, diazo moiety in

diazoallenamines is a good leaving group and the center carbon atom in allene could be a nucleophilic position. Thus, nucleophiles came to eliminate dinitrogen by S_N2 , substitution. TsOH·H₂O was effective in most cases and when using TMSOTf, vinyl triflate compound **5a**' was obtained via umpolung. With ortho-substituted phenyl group, benzofuryl imines **6** were produced through intramolecular cyclizations.



Scheme28. Preparation of bialkyne propargyl alcohols.

2.2 Intermolecular [3+2] azide-alkyne cyclization reaction

After investigation of intramolecular [3+2] cyclization, I then turn my attention to intermolecular transformation which could be more widely used in organic chemistry, chemical biology and pharmaceuticals to produce synthetic precursors. By established conditions of intermolecular reactions, highly substituted 1H-1,2,3-triazoles could be obtained (Scheme 29).



Scheme 29. Generation of 1,2,3-triazoles via allenylaminodiazonium intermediates.

Comparing with typical synthetic method, through this strategy both terminal and internal alkyne could be accepted under low temperature. Formation of the active allenylaminodiazonium intermediates followed by nucleophile addition to the Methylene-triazolium ions, the fully substituted 1,2,3-triazole could be demonstrated rapidly. Although concerted [3+2] reactions would deliver both 1H- and 3H-triazoles,
deactivation of the C2 position by a delocalized carbocation can avoid this pathway and yield products selectively.

This strategy is challenging from the follow points: (1) with azides, a sp^2 carbocation (C1) is more reactive than the desired sp carbocation (C3) to produce unsaturated imines by a Schmidt reaction or propargyl azides (Scheme 30);⁷ (2) Meyer-Schuster rearrangement producing the enones would be competitive.



Scheme 30. Intermolecular Schmidt reaction of azide with carbocation.

2.2.1 Optimization of intermolecular azide-alkyne cyclization reaction

To avoid the reaction at C1 position by steric and electronic influence,⁶² I designed diphenyl propargyl alcohol for the initial study of the reaction condition, and the reaction were quenched with a saturated sodium bicarbonate aqueous solution in order to produce triazolylalkanols, recently reported as new synthetic precursors (Table 5).⁶³

Ph Ph HO	└───── <i>n</i> -Bu	BnN_3 , Reagents H_2Cl_2 (0.1 M) Pho- en sat. NaHCO ₃ aq.	$h \rightarrow h$ $h \rightarrow h$ h	Ph O Ph 15a' Mever-Schu	n-Bu	
	15a		16a	rearrangement	ketone	
Entry	BnN ₃	Reagents	Temp	Time	Yield	
Lmry	(equiv)	(equiv)	(°C)	(min)	(%)	
l^a	2.5	$TsOH \cdot H_2O(1.2)$	rt	20	11	
2	1.5	MsOH (1.2)	rt	10	79	
$\mathcal{3}^b$	1.5	TMSCl (1.2)	rt	120	0	
4^c	1.5	<i>FeCl</i> ₃ (1.2)	rt	5	0	
5	1.5	$Sc(OTf)_{3}(1.2)$	rt	5	52	
6	1.5	Cu(OTf) ₂ (1.2)	rt	10	47	
7	1.5	$BF_{3} \cdot OEt_{2} (1.2)$	rt	1	90	
8	1.5	$BF_{3} \cdot OEt_{2} (1.2)$	-20	5	92	
9	1.5	$BF_{3} \cdot OEt_{2} (1.2)$	-60	5	80	
10	1.5	TMSOTf(1.2)	-78	5	97	
11	1.5	TMSOTf(1.2)	-90	5	99	
12	1.2	TMSOTf(1.2)	-90	5	90	
13	1.5	TMSOTf (1.05)	-90	5	94	
14	1.5	TMSOTf(0.2)	-90	120	16	
15	1.5	TBSOTf(1.2)	-90	5	60	
16^d	1.5	TMSOTf(1.2)	-90	5	88	
17 ^e	1.5	TMSOTf (1.2)	-90	5	97	

Table 5. Optimization of intermolecular [3+2] cyclization reaction

^{*a*} **15a'** was obtained in 30%. ^{*b*} **15a'** was obtained in 90%. ^{*b*} **15a'** was obtained in 24%. ^{*d*} Performed in toluene. ^{*e*} High dilution conditions (0.005M).

Tosylic acid gave a desired triazole **16a**, but the Meyer-Schuster rearrangement product was major probably due to its solubility and the presence of water of hydrates (entry 1). On the other hand, mesylic acid could produce **16a** in 10 min in good yield (entry 2). Although the conditions are effective at rt, further investigations on reagents

were continued to improve the reaction speed and availability under low temperatures. TMSCl and FeCl₃ only afforded the Meyer-Schuster rearrangement ketone 15a' (entries 3-4). $Sc(OTf)_3$ and $Cu(OTf)_2$ worked to yield **16a** in moderate yield (entries 5-6), and $BF_3 \cdot OEt_2$ worked well resulting in an excellent yield (entry 7). This reagent could complete the reaction in 1 min and was powerful enough to perform the reaction at -60 °C (entries 8-9). Further cooling conditions were achieved with TMSOTf, and the desired transformation was successfully demonstrated even at -90 °C, close to the melting point of the solvent (entries 10-11). It should be noted that these reaction conditions could afford 16 in almost quantitative yield in only 5 min at -90 °C. Reducing the equivalence of benzyl azide and an acid reagent could also give similar results (entries 12-13). Unfortunately, catalytic conditions were ineffective probably due to the basicity of the resulting triazoles (entry 14). The use of TBSOTf also worked, but not as well as TMSOTf (entry 15). Instead of dichloromethane, toluene could work as an efficient solvent (entry 16). It is noteworthy that high dilution conditions did not reduce the efficiency of the reaction (entry 17). TfOH, MgBr₂, Ti(OiPr)₄, TiCl₄, or Yb(OTf)₃ were not effective. When using TfOH, the starting material directly converted into the unidentified polymeric materials. Whether MgBr2 or Ti(OiPr)4 could not afford the desired triazole product, only starting material recovered. TiCl₄ only afford the Meyer-Schuster rearrangement ketone. Yb(OTf)₃ also could not gave the desired product, starting materials was recovered along with trace amount of Meyer-Schuster rearrangement product.

2.2.2 Scope of intermolecular azide-alkyne cyclization reaction

With optimized condition in hand, I pay attention to scope of substituents on the alkynes and azides. Herein, the cyclization reaction was carried out with TMSOTf in dichloromethane at -90 °C.





^a 2.5 equiv of BnN₃ and 2.1 equiv of TMSOTf were used at room temperature.

Firstly, investigation of substrates was performed with R^1 and R^2 group (Table 6). Electron deficient aryl **16b** was found to be effective as a phenyl group, despite the electron-donative aryl **16c** was obtained in low yield under this standard reaction condition. Methylphenyl propargyl alcohol **15d** gave **16d'** in a moderated yield as dehydrated product. In the case of benzyl alcohol **15g**, aldehyde **16g'** was obtained probably through [3+2] followed by Schmidt reaction-hydrolysis (Scheme 31).



Scheme 31. Cyclization of Propargyl alcohol 15g.

Diethyl propargyl alcohol **15f** didn't afford the desired triazole product **15f** in this intermolecular cyclization reaction, only starting material was recovered after the reaction. And the fluorenyl substrate **15e** was labile under acid conditions. With electron withdrawing methoxycarbonyl group, propargyl alcohol **15h** was converted into azido triazole **16h'** at ambient temperature. In this case, the azidation was seemed to proceed on the C1 position. Because these types of azido triazoles were not found in other cases,

I described the possible mechanism in Scheme 32. The azide group in compound **16h**' may be introduced by the [3+2] cyclization of an additional benzyl azide to the unsaturated ester moiety of methylenetriazolium ions **15h**', and the ring-opening of spirocyclic bistriazoline **15h**'' would trigger the production of **16h**'. Generation of product **16h**' could be one of the evidence of formation of triazolium intermediate **15h**'.



Scheme 32. Cyclization of Propargyl alcohol 15h.

After that, I began to investigate the reaction with R³ group on alkyne moiety (Table 7). Secondary cyclohexyl alkyne **15i**, alkoxyl methyl **15l-m** and methoxy carbonyl **15o** produced the corresponding triazoles **16i**, **16l-m**, and **16o** in good to excellent yields. Interestingly, transformation of terminal alkyne **15k** was also achieved in giving the desired disubstituted triazole **16k** in 90%. The propargylic alkyne in **15q** could selectively react with benzyl azide to afford the triazole **16q** in good to excellent yield. Even sterically bulky substrate **15r** could produce the cyclization product **16r** rapidly.

On the other hand, I also found that conjugated alkyne **15j** and propargyl acetate alkyne **15n** gave corresponding triazoles in fair yield. It was observed that the desired product **15j** was generated with unexpected byproduct benzazepine **15j'**. The plausible mechanism of **15j'** was described in Scheme 33. In the case of trityl compound **15s** (eq 2), triazolation followed by Fridel-Crafts reaction occurred to produce tricyclic product **16s'**. In addition, the cyclohexenyl alkyne was converted into cyclopentyl aldehyde **15t'** (eq 3). Probably, the benzyl azide selectively reacted with alkene group instead of alkyne to form triazoline followed by ring-contraction-hydrolysis to obtain the aldehyde **15t'** (Scheme 34).

Table 7. Scope of R^3 group study





^{*a*} 2.1 equiv of TMSOTf were used. ^{*b*} Along with 7% deTBS triazole. ^{*c*} Performed at -60 °C. ^{*d*} 10 min reaction.



Scheme 33. Paulisable mechanism of compound 15j'.



Scheme 34. Paulisable mechanism of compound 15t'.

Furthermore, I investigated the substitutents of \mathbb{R}^4 on organic azides (Table 8). In contrast to unreactive phenyl azide, the sterically hindered 2,6-diisopropylphenylazide successfully converted to desired triazole **16v** in good yield, due to the stronger nucleophilicity of its azide moiety comparing with phenyl azide as reported by Hosoya *et al.*⁶⁴ They reported that calculation of the transition state structure of the cyclization, the activation energy for the cyclization of cyclooctyne with 2,6-diisopropylphenylazide was estimated to be 2.5 kcal mol⁻¹ lower than that with phenyl azide, providing a good agreement with the experimental result. It is revealed that although the less nucleophilic sulfonylazide could not give product **16x**, the primary azide (3-azidopropan-1-Boc protected amine, cinnamyl azide) and bulky tertiary azide (adamantyl azide) could generated the desired trisubstituted triazoles **16w** and **16y-z** in good yield. It should be noted that all triazoles were obtained as single isomer even in the case of ester **16p**.

Table 8. Scope of R^4 group study



^a 2.5 equiv of TMSOTf was used.







^a 2.1 equiv of MsOH was used.

Moreover, to conduct this cyclization reaction at ambient temperature, the reactions were investigated with methanesulfonic acid (MsOH) (Table 9). The desired transformation was successfully demonstrated at room temperature in 10 min in moderate yields.

Chapter 3 Multicomponent Coupling Reaction

In the studies described, I mainly disclosed the introduction of hydroxyl group into the triazole products by quenching with aqueous media as nucleophiles. However, considering the proposed reaction mechanism (Scheme 29), the products could be functionalized using other additional nucleophiles instead of a hydroxyl group. Before testing the generality of multicomponent coupling reaction, I investigated the origin of the hydroxyl group to make sure the hydroxyl group's source (Scheme 35).

The treatment of methylation propargyl alcohol **17** with stoichiometric amount of trimethylsilyl trifluoromethanesulfonate (TMSOTf), which were mainly used in previous work, only afforded the triazole possessing methoxy group **18** at both -90 and -60 °C, even after quenching with an aqueous medium.



Scheme 35. Investigation of the origin of hydroxyl group.

However, $BF_3 \cdot OEt_2$ gave hydroxy compound **16a** as the major product. This indicates that the origin of the hydroxy group depends on the acid used, and the resulting silanols or silyl ethers may be the source of hydroxy groups in the case of TMSOTf. With $BF_3 \cdot OEt_2$, the benzylic position was successfully substituted by external water.

This results also indicate that $BF_3 \cdot OEt_2$ is a suitable acid for the substitution of triazolium intermediates with additional nucleophiles probably because hydroxy group donation activity of the resulting boronates or borinates are limited. For introduction of hydroxy groups, TMSOTf seems to be better reagent. Methyl ether **17** was not obtained from **16a** by acid treatments in the presence of methanol. Thus, the functionalization of the benzylic position should be performed as one-pot reactions.

Based on these results, three-component reaction coupling reactions with various nucleophiles were investigated to functionalize the benzylic position with BF₃·OEt₂ (Table 10, Condition A). In order to avoid the quenching reaction by moisture, the reaction was carried out under nitrogen gas atmosphere. As nucleophiles, primary alcohols could produce the desired ether triazoles **19a-e** in good yields along with triazolyalkanol **16a** (entries 1-5). Although the reactions with secondary alcohols was failed to afford the coupling products, naphthalenethiol, diethylamine and allyl amine

were successfully introduced to the benzylic position to afford coupling product **19f-h** in good yields (entries 6-8). When azidotrimethylsilane (TMSN₃) was used as the nucleophile, azido compound **19i** was generated in good yield.





Yield (%)

Entry	Nucleophiles and p	products	Cond	ition A	Cond	lition B
			19	16 a	19	16a
1	5 () Me 7	19a	80	10	69	trace
2	stort Br	19b	73	11	58	11
3	s ^S OOMe	19c	76	12	63	0
4	5000	19d	74	12	53	trace
5	st off	19e	77	10	61	0
6	st s	19f	81	7	62	0
7	s ^S N Et Et	19g	78	11	55	0



^{*a*} Azidotrimethylsilane was used as the nucleophile. ^{*b*} Allyltributyltin was used as the nucleophile. ^{*c*} Ethyl vinyl ether was used as the nucleophile. ^{*d*} 1-Ethoxy-1-trimethylsilyloxyethylene was used as the nucleophile.



Interestingly, through these three-component coupling reactions, not only heteroatom nucleophiles, but also carbon nucleophiles could achieve to form quaternary carbon centers. Indole selectively gave the corresponding coupling product **19j** in good yield (entry 9). The allyl group was successfully introduced with allyltributyltin to afford the **19k** (entry 10), while the allylsilanes was failed. These carbon-carbon bond formations also could be achieved by silyl enol ethers⁶⁵ to give obtain the desired

aldehyde **191** and ethyl ester **19m** (entries 12 and 13). Addition of molecular sieves did not improve the results.

In addition, to achieve these three-component coupling reactions at ambient temperature, the same reactions were investigated with MsOH (Table 10, Condition B), and the desired coupling products were successfully obtained in 40 min. Interestingly, although the product yields were slightly lower than those obtained under condition A, lesser amount of byproduct **16a** were obtained. Considering the generation of **16a** in Scheme 46, boronic acids or borates were still active as nucleophiles, and water molecule generated by MsOH was relatively weaker nucleophiles than those acids.

Generation of three-component coupling reactions also indicated the presence of methylenetriaolium intermediates or carbocations of triazole (Scheme 29). In chapter 2 of scope of intermolecular cyclization, generation of azido triazole **16h** could be other evidence of the methylenetriazolium intermediates **15h**' from **15h** (Scheme 32).

To further develop the functionalization of triazoles by multicomponent coupling reactions, double [3+2] triazolation reactions were investigated. Based on the chapter 2 results, at first we investigated the cyclization reaction with dihexynyl alcohol **20** in the present of TMSOTf at -90 °C (Table 11). After 5 min, the reactions were quenched by saturated sodium bicarbonate aqueous solution to produce hydroxylated compound **21**.

With 2.5 equiv BnN₃, the corresponding double triazolation product **21** was obtained in 62% along with mono-cyclization product **22** in 7% (entry 1). Increasing the amount of BnN₃ to 3.0 equiv, the best yield of desired bistriazole product **21** was raised to 72% with trace amount of mono triazole **22** (entry 2). On the other hand, this cyclization reaction could be controlled to afford monotriazolation product **22** in moderate yield by reducing the amount of organic azide (entry 3).





Based on these successful results, I then conducted the four-component coupling reaction with two kinds of organic azide. Firstly, mono triazolation compound **22** was used to demonstrate the desired four-component coupling product (Scheme 36). Using cinnamyl azide, the mono substrate could convert into the corresponding double triazolation product **21** in 15% yield, but Meyer-Schuster rearrangement byproduct **22**'

was obtained in 58%. Thus, to effectively generate the four component coupling reaction product, this transformation should be performed as one-pot reactions.



Scheme 36. Cyclization reaction of mono triazolated compound.

Controlling the reagent (Condition A), double [3+2] cyclization reaction was sequentially achieved to produce the desired hydroxylated triazole compound **23** in moderate yield (Scheme 37). In additional, not only hydroxyl group but also other functional group could be introduced by changing quenching method with other nucleophiles. For further functionalization, allylamine was used as a nucleophile, and the dialkyne **21** was successfully converted in to the corresponding four-component product **24** (Condition B).

According to the successful results, I also tested $BF_3 \cdot OEt_2$ to improve the yields of four-component coupling reactions (Scheme 38). Under condition C and D, double cyclization successfully gave the corresponding products **23** and **24** in moderate yields. It was noticed that, not only TMSOTf but also $BF_3 \cdot OEt_2$, the desired triazoles was obtained along with trace amount of Meyer-Schuster rearrangement product, hydroxy-subsituted compound and polymeric materials. In these cases, introduction of carbon nucleophiles like indole were unsuccessful, and hydroxylated compunds were produced.



Scheme 37. Investigation of four-component coupling reaction with TMSOTf.



Scheme 38. Investigation of four-component coupling reaction with BF₃·OEt₂.

Chapter 4 Synthesis of Serine Hydrolases Inhibitor

and Its 5-Substituted Derivatives

Serine hydrolases are considered as one of the largest known and most diverse enzyme families. One characteristic defining feature of this family is the presence of an active site nucleophilic serine that is used for hydrolysis of the substrates.⁶⁶



Scheme 39. Hydrolization of activated serine hydrolase.

The nucleophilic serine of these hydrolases is typically activated by a proton relay involving an acidic residue (aspartate or glutamate) and a basic residue (usually histidine) (Scheme 39). *N*-acetyl-L-phenylalanine *p*-nitrophenyl amide enables to be used as substrate analogs for enzyme assays. This powerful nucleophilic serine residue attacked the unreactive carbonyl group forming an enzyme substrate intermediate. Then the peptide bond was cleaved followed by hydrolysis again. Along with histidine and aspartic acid, this serine residue constitutes the catalytic triad of the active site.

Due to the biological importance, serine hydrolases have been used as the targets of clinical drugs to treat various diseases such as diabetes and Alzheimer's disease.⁶⁷ However, the biochemical activities of these enzymes are yet to be understood. For this reason, it is important to produce efficient synthetic methods to prepare selective inhibitors of these enzymes and preparation of its derivatives as candidates of more active drug molecules. To develop the efficiency of this triazole synthesis for bioactive molecule synthesis, I carried out the synthesis of triazole urea 26a reported as a serine hydrolase inhibitor by Cravatt et al.⁶⁸ and its 5-substituted derivatives. Cravatt's reported the serine hydrolase inhibitors can selectively inhibit enzymes from diverse branches of the serine hydrolase family, such as peptidases (aceyl-peptide hydrolase), lipases (platelet-activating factor acetylhydrolase-2), uncharacterized hydrolases $(\alpha,\beta$ -hydrolase-11, ABHD11). Here, my triazole products have the same core structure of AA32-1 and AA44-2 which selectively inhibit ABHD11.



Scheme 40. Synthesis of Serine hydrolases inhibitor 26a.

1,4-disubstituted **26a** were prepared from appropriate propargyl alcohols with TMSOTf followed by quenching with aqueous media (Scheme 40). The obtained *N*-benzyltriazoles were deprotected by hydrogenolysis, and the obtained unprotected triazoles were coupled with carbamoyl chloride **26** to afford serine hydrolase inhibitor triazole urea **26a** regioselectively.

To confirm the structure of synthesized triazole urea compound **26a**, same as the reported data, I compared the ¹H and ¹³C NMR analysis in CDCl₃ along with other analytical data. However, the spectroscopic data of compound **26a** didn't match the reported data (Table 12). Then I noticed that the peak of benzylic carbon C1 (δ 77.2) was overlapped with the solvent peak (CDCl₃, δ 77.0) in ¹³C NMR analysis. However, the split patterns of δ 53.8 and 25.2 (Cx and Cy) were quite different from mine and

C1 carbon peak at 91.1 was not recognized.

	1 1 0 1	1	
Reported data	26a	Reported	26a
¹ H-NMR	¹ H-NMR	¹³ C-NMR	¹³ C-NMR
(400MHz, CDCl ₃)	(500MHz, CDCl ₃)	(100MHz, CDCl ₃)	(126MHz, CDCl ₃)
7.49 (s, 1H)	7.56 (s, 1H)	152.7	155.6
7.30-7.18 (m, 10H)	7.33-7.29 (m, 10H)	144.5	147.7
3.97 (t, <i>J</i> = 6.4Hz, 1H)	3.76 (t, <i>J</i> = 6.5Hz, 2H)	131.2	144.9
3.78 (t, <i>J</i> = 6.4Hz, 1H)	3.70 (t, <i>J</i> = 6.5Hz, 2H)	128.8	135.3
3.65 (m, 2H)	1.96-1.92 (m, 4H)	127.0	128.2
1.91 (m, 4H)		126.5	127.8
		123.3	127.1
		91.1	77.2
		53.8	50.1
		25.2	48.7
			26.4
			24.0

Table 12. ¹*H* and ¹³*C* spectroscopic data of compound **26a** (in CDCl₃).

Therefore, I changed solvent the CDCl₃ to CD₃OD and CD₂Cl₂. Then after discussion with Professor Adibekian one of the authors, he kindly gave me the NMR analysis data of CD₃OD. My ¹H NMR data almost matched to their corrected data (Figure 5, 6). For ¹³C NMR, at



first, they still have two incorrect assignments around $\delta 100.0$ and $\delta 55.0$ (Figure 7, yellow). They also miss assign two peaks (Figure 7, red). They think the peak of δ 78.0 from chloroform and δ 49.8 is another impurity. Finally, they agree with my NMR assignment and the incorrect NMR assignments were found in the reference article. The C1 was found at 78.0 and Cx were found at δ 51.5 and δ 49.8 which is

near with the solvent peak (CD₃OD, δ 49.0). Finally, the NMR data of compound **26a** were identical to those in CD₃OD, which were newly provided by author (Table 13, Figure 6, 8).



Figure 5. ¹*H NMR spectroscopic data of* **26***a provided by author (in CD*₃*OD).*



Figure 6. ¹*H NMR spectroscopic data of* **26***a provided by author (in CD*₃*OD).*



Figure 7. ¹*H NMR spectroscopic data of* **26***a provided by author (in CD*₃*OD).*



Figure 8. ¹*H NMR spectroscopic data of* **26***a provided by author (in CD*₃*OD).*

26b	Corrected data	26b
¹ H-NMR	¹³ C-NMR	¹³ C-NMR
(500M, CD ₃ OD)	(100M, CD ₃ OD)	(125M, CD ₃ OD)
7.84(s, 1H)	158.0	158.0
7.23-7.37(m, 10H)	149.6	149.6
3.73(t, J = 6.0Hz, 2H)	147.1	147.1
3.64(t, J = 6.0Hz, 2H)	136.7	136.7
1.94-1.90(m, 4H)	128.9	128.9
	128.5	128.5
	128.4	128.4
	78.0	78.0
	51.5	51.5
	49.8	49.8
	27.4	27.4
	25.0	25.0

Table 13. ¹*H* and ¹³*C* spectroscopic data of compound **26b** (in CD₃OD).

Then after comfirmation of the compound **26a**, I began to synthesize the its 5-subsitituted derivetives **26b-f** (Scheme 41). According to the sucessful result, 26b-f were obtained from corresponding propargyl alkyne under starding cyclization condition, followed by debenzylation and coupling with compound **25**. In all cases, desired 2*H*-triazole ureas **26a–f** were obtained regioselectively. Description of structures of **26d,e** were unambiguously confirmed by X-ray crystallographic analysis. Since it is difficult to prepare triazole ureas **26a,c–f** obtained from internal alkynes by CuAAC, this method could prove to be an efficient way to explore the property and activity of fully substituted triazole molecules.



Scheme 41. Synthesis of serine hydrolases inhibitor and its 5-substituted derivatives.

Chapter 5 Supporting Information

5.1 Experimental section

General information

¹H and ¹³C NMR were recorded on a JEOL JNM-ECP500 spectrometer (500MHz for ¹H NMR, 126 MHz for ¹³C NMR). Chemical shifts are reported as δ values in ppm and calibrated by residual solvent peak (CDCl₃, δ 7.26 for 1H NMR, δ 77.00 for ¹³C NMR; CD₃OD, δ 3.31 for ¹H NMR, δ 49.00 for ¹³C NMR; CD₂Cl₂, δ 5.32 for ¹H NMR, δ 53.8 for ¹³C NMR) or tetramethylsilane (δ 0 for ¹H NMR). Abbreviations are following: s (singlet), d (doublet), t (triplet), q (quartet), br (broad peak), m (complex multiplet). Inflared spectra were measured on a JASCO FT/IR-4200 spectrometer. Mass spectra were recorded on a JEOL JMS-700 MStaion [EI (70 eV), CI, FAB and ESI]. X-ray crystallography was performed on Rigaku R-AXIS RAPID/S imaging plate diffractometer. Flush column chromatography was performed by MERCK Silica gel 60. The progress of reactions was monitored by silica gel thin layer chromatographyplates (MERCKTLC Silicagel 60 F_{254}). Phosphomolybdic acid ethanol solution, ninhydrin-acetic acid butanol solution and anisaldehyde-acetic acid-sulfuric acid ethanol solution were used as TLC stain. All reagents were purchased from Sigma-Aldrich, Wako pure chemical industries, Ltd, TCI (Tokyo Chemical Industry, Co. Ltd), Kanto Chemical Co. Inc., and Nakalai Tesque. Used Dehydrated solvents -tetrahydrofuran, dichloromethane and toluene- were purchased from Kanto Chemical, Wako pure chemical industries, Ltd, and Nakalai Tesque. Sodium azide purchased from Nakalai Tesque was carefully handled, and transferred with plastic spatulas.

pent-4-yn-1-yl 4-methylbenzenesulfonate (2)



To a stirred solution of 4-pentyn-1-ol **1** (100.0 mg, 1.19 mmol) and TsCl (249.3 mg, 1.31 mmol) in dichloromethane (12 ml) was added dropwise triethylamine (0.2 mL, 1.43 mmol) at 0 °C, then reaction mixture was allowed to warm up to ambient temperature. After 2 h, the mixture was diluted with ethyl acetate and was washed with water and brine. Drying collected organic layer over MgSO₄ followed by silica gel column chromatography (ethyl acetate / hexane = 1/10) gave tosylate **2** (220 mg, 77.7%) as a colorless oil.

Colorless oil; R_f value 0.53(ethyl acetate / hexane = 1/3); IR (NaCl, neat) $v_{max} = 3291$, 2962, 1598, 1360, 1176 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.79 (d, 2H, J = 8.5 Hz), 7.35 (d, 2H, J = 8.5 Hz), 4.14 (t, 2H, J = 6.0 Hz), 2.45 (s, 3H), 2.25 (td, 2H, J = 6.5, 2.5 Hz), 1.88 (t, 1H, J = 2.5 Hz), 1.86 (tt, 2H, J = 6.5, 6.0 Hz); ¹³C NMR (126 MHz,

CDCl₃) δ 144.8, 132.9, 129.8, 127.9, 82.1, 69.4, 68.7, 27.7, 21.6, 14.7; HRMS (ESI) calcd for C₁₂H₁₄O₃SNa [M+Na]⁺ 261.0561, found 261.0561.

General Experimental Procedure of propargyl alcohols

To a stirred solution of tosylate 2 in dry THF under an atmosphere of nitrogen was added dropwise *n*-BuLi at -78 °C and the mixture was stirred for 10 min. Benzaldehyde was then added at same temperature. After 4h, the reaction was quenched with water. The mixture was diluted with ethyl acetate and was washed with water and brine. Drying collected organic layer over MgSO₄. The crude product can be used to the next step without further purification.

To a stirred solution of benzyl alcohols in DMF was added sodium azide at room temperature and the reaction mixture was heated to 50 °C. After 20 min, the reaction mixture was diluted with ether and was washed with water and brined. Drying collected organic layer over MgSO₄ followed by silica gel column chromatography gave azide as colorless oil.

6-hydroxy-6-phenylhex-4-yn-1-yl 4-methylbenzenesulfonate (3)



The reaction with tosylate **2** (146 mg, 0.612 mmol), *n*-BuLi (1.6 M in hexane, 0.44 mL, 0.706 mmol) and benzaldehyde (47.6 μ L, 0.471 mmol) in TH F (4.8 mL) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1 / 5) gave benzyl alcohol **3** (148 mg, 91% based on benzaldehyde) as a colorless oil.

Colorless oil; R_f value 0.15(ethyl acetate / hexane = 1/3);IR (NaCl, neat) $v_{max} = 3524$, 2960, 1598, 1189, 1175, 930 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.77 (d, 2H, J = 8.0 Hz), 7.48 (d, 2H, J = 8.0 Hz), 7.30 – 7.38 (m, 3H), 5.36 (br, 1H), 4.14 (t, 2H, J = 6.5 Hz), 2.42 (s, 3H), 2.35 (td, 2H, J = 7.0, 2.0 Hz), 1.87 (tt, 2H, J = 7.0, 6.5 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 144.8, 140.9, 132.8, 129.8, 128.5, 128.2, 127.8, 126.5, 84.2, 81.3, 68.9, 64.5, 27.6, 21.6, 15.1; HRMS (ESI) calcd for C₁₉H₂₀O₄SNa [M+Na]⁺ 367.0980, found 367.09800.

6-azido-1-phenylhex-2-yn-1-yl acetate (4a)



To a stirred solution of benzyl alcohol **3** (780 mg, 2.02 mmol) in DMF (20 mL) was added sodium azide (170.6 mg, 2.62 mmol) at room temperature and the reaction mixture was heated to 50 °C. After 20 min, the reaction mixture was diluted with ether and was washed with water and brined. Drying collected organic layer over MgSO₄ followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 3) gave azide **4a** (487.9 mg, 94%) as a colorless oil.

Colorless oil; R_f value 0.45(ethyl acetate / hexane = 1/2); IR (NaCl, neat) $v_{max} = 2931$, 2098, 1739, 1227 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.52 (d, 2H, *J* = 7.0 Hz), 7.38 (m, 3H), 6.45 (br, 1H), 3.40 (t, 2H, *J* = 7.0 Hz), 2.40 (td, 2H, *J* = 2.0, 6.5 Hz), 2.10 (s, 3H), 1.80 (tt, 2H, *J* = 6.5, 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 169.6, 137.2, 128.7, 128.5, 127.5, 86.1, 77.9, 65.7, 49.9, 27.4, 20.9, 16.0; HRMS (ESI) calcd for C₁₄H₁₅N₃O₂Na [M+Na]⁺ 280.1062, found 280.1061.

6-azido-1-(3,4-dimethoxyphenyl)hex-2-yn-1-ol(4c)



The reaction with tosylate 2 (2.15 g, 9.03 mmol), *n*-BuLi (1.55 M in hexane, 6.41 mL, 9.93 mmol) and benzaldehyde (1 g, 6.02 mmol) in THF (60 mL) followed by collected the organic layer under vacuum affording the crude product 2.97g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol (2.87g, 7.15 mmol) with sodium azide (0.511g, 7.86mmol) in DMF (25ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 5 to 1/2) gave azide **4c** (1.81 g, 92%) as a colorless oil.

Colorless oil; R_f value 0.34 (ethyl acetate / hexane = 1/2); IR (NaCl, neat) v_{max} 3433, 2938, 2098, 1593, 1125 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.07(m, 2H), 6.86(d, 1H, *J* = 8.5 Hz), 5.41(m, 1H), 3.91(s, 3H), 3.89(s, 3H), 3.42(t, 2H, *J* = 7.0 Hz), 2.42(td, 2H, *J* = 6.5, 2.0 Hz), 2.18(d, 1H, OH, *J* = 5.5 Hz), 1.82(tt, 2H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 149.1, 133.7, 118.9, 110.9, 109.8, 85.4, 81.2, 64.6, 55.93, 55.86, 50.2, 27.7, 16.1; HRMS (ESI) calcd for C₁₄H₁₇N₃O₃Na [M+Na]⁺ 298.1168, found 298.1166. **6-azido-1-(3,4,5-trimethoxyphenyl)hex-2-yn-1-ol(4d)**



The reaction with tosylate 2 (1.09 g, 4.59 mmol), *n*-BuLi (1.55 M in hexane, 3.26 mL, 5.05 mmol) and benzaldehyde (1 g, 3.06 mmol) in THF (31 mL) followed by collected the organic layer under vacuum affording the crude product 1.32g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol (1.32 g, 3.04 mmol) with sodium azide (0.217 g, 3.34 mmol) in DMF (15 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1/1) gave azide **4d** (0.897 g, 96%) as a colorless oil.

Colorless oil; R_f value 0.48(ethyl acetate/hexane = 1/1); IR (NaCl, neat) v_{max} 3433, 2999, 2099, 1594, 1125 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.73(s, 2H), 5.34(t, 1H, *J* = 2.0 Hz), 3.84(s, 6H), 3.80(s, 3H), 3.39(t, 2H, *J* = 6.5 Hz), 2.37(td, 2H, *J* = 7.0, 2.0 Hz), 1.78(tt, 2H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 153.1, 137.6, 136.7, 103.4, 85.2, 81.1, 64.6, 60.7, 56.0, 50.0, 27.6, 16.0; HRMS (ESI) calcd for C₁₅H₁₉N₃O₄Na [M+Na]⁺ 328.1273, found 328.1273.

6-azido-1-(4-chlorophenyl)hex-2-yn-1-ol(4e)



The reaction with tosylate 2 (1.53 g, 6.40 mmol), *n*-BuLi (1.55 M in hexane, 4.54 mL, 7.04 mmol) and benzaldehyde (0.6 g, 4.27 mmol) in THF (42 mL) followed by collected the organic layer under vacuum affording the crude product 1.6g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol (1.6 g, 6.40 mmol) with sodium azide (0.499 g, 7.68mmol) in DMF (32 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1/3) gave azide **4e** (1.08 g, 99%) as a colorless oil.

Colorless oil; R_f value 0.52(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3388, 2936, 2099, 1488, 1255, 1089 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.45(m, 2H), 7.34(m, 2H), 5.41(s, 1H), 3.40(t, 2H, J = 6.0 Hz), 2.43(br, 1H, OH), 2.39(td, 2H, J = 7.0, 2.0 Hz), 1.80(tt, 2H, J = 7.0, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 139.5, 134.0, 128.7, 127.9, 85.9, 80.7, 63.9, 50.1, 27.6, 16.1; HRMS (ESI) calcd for C₁₂H₁₂ClN₃ONa [M+Na]⁺ 272.0567, found 272.0566.

6-azido-1-(2,4-dichlorophenyl)hex-2-yn-1-ol(4f)



The reaction with tosylate 2 (787.0 g, 1.7 mmol), *n*-BuLi (1.55 M in hexane, 2.27 mL, 1.85 mmol) and benzaldehyde (0.34 g, 1.94 mmol) in THF (20 mL) followed by collected the organic layer under vacuum affording the crude product 0.880 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol (0.880 g, 2.13 mmol) with sodium azide (0.166 g, 2.55 mmol) in DMF (10 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 5) gave azide **4f** (0.461 g, 87%) as a colorless oil.

Colorless oil; R_f value 0.38(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3346, 2932, 2098, 1469, 1254 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.67(d, 1H, *J* = 8.5 Hz), 7.39(d, 1H, *J* = 1.5 Hz), 7.30(dd, 1H, *J* = 8.5, 1.5 Hz), 5.75(br, 1H), 3.40(t, 2H, *J* = 6.5 Hz), 2.40(br, 1H), 2.39(td, 2H, *J* = 7.0, 2.0 Hz), 1.80(tt, 2H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 136.9, 134.8, 133.2, 129.5, 129.0, 127.5, 85.9, 79.6, 61.5, 50.1, 27.6, 16.1; LRMS (EI) 255([M-N₂]⁺, 5%), 220(86), 175(64), 110(100); HRMS (EI) calcd for C₁₂H₁₁Cl₂NO [M-N₂]⁺ 255.0218, found 255.0188.

6-azido-1-(2-methoxyphenyl)hex-2-yn-1-ol(4h)



The reaction with tosylate 2 (3.94 g, 16.5 mmol), *n*-BuLi (1.55 M in hexane, 12.77 mL, 19.8 mmol) and benzaldehyde (1.5 g, 11.02 mmol) in THF (110 mL) followed by collected the organic layer under vacuum affording the crude product 3.12 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol (3.12 g, 8.33 mmol) with sodium azide (0.65 g, 10 mmol) in DMF (8 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 2) gave azide **4h** (1.98 g, 97%) as a colorless oil.

Colorless oil; R_f value 0.36(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3417, 2938, 2098, 1244 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.56(dd, 1H, *J* = 7.5, 2.0 Hz), 7.31(ddd, 1H, *J* = 7.5, 7.5, 2.0 Hz), 6.99(dd, 1H, *J* = 7.5, 7.5 Hz), 6.92(d, 1H, *J* = 7.5 Hz), 5.70(td, 1H, *J* = 6.0, 2.0 Hz), 3.90(s, 3H), 3.43(t, 2H, *J* = 6.5 Hz), 3.00(d, 1H, OH, *J* = 6.0 Hz),

2.41(td, 2H, J = 7.0, 2.0 Hz), 1.82(tt, 2H, J = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 156.7, 129.6, 128.9, 127.7, 120.8, 110.7, 85.0, 80.4, 61.2, 55.5, 50.2, 27.7, 16.2; HRMS (ESI) calcd for C₁₃H₁₅N₃O₂Na [M+Na]⁺ 268.1062, found 268.1061.

6-azido-1-(2,4-dimethoxyphenyl)hex-2-yn-1-ol(4i)



To a stirred solution of 5-azidopent-1-yne (945.7 mg, 8.67 mmol) in dry THF (48 mL) under an atmosphere of nitrogen was added dropwise *n*-BuLi (1.58 M in hexane, 5.79 mL, 9.15 mmol) at -78 °C and the mixture was stirred for 60 min. Benzaldehyde (700.0 mg, 4.81 mmol) was then added at same temperature. After 8 h, the reaction was quenched with water. The mixture was diluted with ethyl acetate and was washed with water and brine. Drying collected organic layer over MgSO₄ followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1 / 5 to 1/3) gave desired benzyl alcohol **4i** (0.774 g, 67%) as a colorless oil.

Colorless oil; R_f value 0.33 (ethyl acetate / hexane = 1 / 2); IR (NaCl, neat) v_{max} 3426, 2938, 2099, 1613, 1506, 1208 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.46(d, 1H, *J* = 9.0 Hz), 6.48(d, 1H, *J* = 2.5 Hz), 6.46(dd, 1H, *J* = 7.0, 2.5 Hz), 5.65(m, 1H), 3.85(s, 3H), 3.80(s, 3H), 3.42(t, 2H, *J* = 7.0 Hz), 2.83(d, 1H, OH, *J* = 6.0 Hz), 2.40(td, 2H, *J* = 7.0, 2.5 Hz), 1.81(tt, 2H, *J* = 7.0, 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 161.0, 157.1, 128.5, 121.8, 104.1, 98.7, 84.6, 80.7, 60.6, 55.5, 55.4, 50.2, 27.7, 16.1; LRMS (EI) 275(M⁺, 100%), 216(71); HRMS (EI) calcd for C₁₄H₁₇N₃O₃ (M⁺) 275.1270, found 275.1262.

6-azido-1-(2,4,6-trimethoxyphenyl)hex-2-yn-1-ol(4j)



To a stirred solution of 5-azidopent-1-yne (1.4 mg, 12.84 mmol) in dry THF (71 mL) under an atmosphere of nitrogen was added dropwise *n*-BuLi (1.58 M in hexane, 8.58 mL, 913.56 mmol) at -78 °C and the mixture was stirred for 60 min. Benzaldehyde (1.4 g, 7.14 mmol) was then added at same temperature. After 8 h, the reaction was quenched with water. The mixture was diluted with ethyl acetate and was washed with water and brine. Drying collected organic layer over MgSO₄ followed by silica gel

column chromatography (ethyl acetate / hexane = 1/10 to 1/6 to 1/3) gave desired benzyl alcohol **4j** (1.7 g, 80%) as a colorless oil.

Colorless oil; R_f value 0.33(ethyl acetate / hexane = 1/1); IR (NaCl, neat) v_{max} 3427, 2938, 2098, 1593, 1233, 1125 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.13(s, 2H), 5.78(m, 1H), 3.85(s, 6H), 3.80(s, 3H), 3.37(t, 2H, J = 7.0 Hz), 2.30(t, 2H, J = 7.0 Hz), 1.74(tt, 2H, J = 7.0, 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 160.9, 158.0, 110.7, 91.1, 82.5, 81.2, 56.3, 55.8, 55.3, 50.1, 27.7, 16.1; HRMS (ESI) calcd for C₁₅H₁₉N₃O₄Na [M+Na]⁺ 328.1273, found 328.1273.

2-(6-azido-1-hydroxyhex-2-yn-1-yl)phenol(4k)



The reaction with tosylate 2 (1.9 g, 7.98 mmol), *n*-BuLi (1.55 M in hexane, 5.67 mL, 19.8 mmol) and benzaldehyde (0.65 g, 5.32 mmol) in THF (50 mL) followed by collected the organic layer under vacuum affording the crude product 1.86 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol (1.86 g, 5.16 mmol) with sodium azide (0.402 g, 6.19 mmol) in DMF (10 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1/8) gave desired azide **4k** (0.672 g, 57 %) as a colorless oil.

Colorless oil; R_f value 0.4(ethyl acetate / hexane = 1/1); IR (NaCl, neat) v_{max} 3481, 2938, 2099, 1490, 1245, 755 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.32–7.33(m, 2H including OH), 7.24(ddd, 1H, J = 7.5, 7.5, 1.5 Hz), 6.89–6.91(m, 2H), 5.66(d, 1H, J = 5.5 Hz), 3.42(t, 2H, J = 7.0 Hz), 3.04(d, 1H, OH, J = 5.5 Hz), 2.42(td, 1H, J = 7.0, 2.0 Hz), 1.81(tt, 2H, J = 7.0, 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 155.1, 130.0, 127.5, 124.9, 120.2, 117.0, 87.2, 79.1, 64.0, 50.1, 27.6, 16.1; HRMS (ESI) calcd for C₁₂H₁₃N₃O₂Na [M+Na]⁺ 254.0906, found 254.0905.

6-azido-1-(4-chloro-2-methoxyphenyl)hex-2-yn-1-ol(4l)



The reaction with tosylate 2 (2.10 g, 8.79 mmol), *n*-BuLi (1.55 M in hexane, 6.74 mL, 9.67 mmol) and benzaldehyde (1.0 g, 5.86 mmol) in THF (59 mL) followed by collected the organic layer under vacuum affording the product 2.74 g. [silica gel

purification (ethyl acetate / hexane = 1 / 10 to 1/5 to 1/1)].

Colorless oil; R_f value 0.24(ethyl acetate / hexane = 1/3); IR (NaCl, neat) v_{max} 3433, 2963, 2116, 1637, 1248, 1175 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.78(d, 2H, *J* = 8.5 Hz), 7.44(d, 1H, *J* = 8.0 Hz), 7.32(d, 2H, *J* = 8.5 Hz), 6.95(dd, 1H, *J* = 8.0, 1.5 Hz), 6.88(d, 1H, *J* = 2.0 Hz), 5.58(t, 1H, *J* = 2.0 Hz), 4.15(t, 2H, *J* = 6.0 Hz), 3.87(s, 3H), 2.43(s, 3H), 2.35(td, 2H, *J* = 6.5, 2.0 Hz), 1.87(tt, 2H, *J* = 6.5, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 157.1, 144.8, 134.9, 132.9, 129.8, 128.6, 127.8, 127.6, 120.7, 111.5, 84.7, 80.2, 68.9, 60.2, 55.8, 27.7, 21.6, 15.1; HRMS (ESI) calcd for C₂₀H₂₁ClO₅SNa [M+Na]⁺ 431.0696, found 431.0699.

Then the crude benzyl alcohol (2.74 g, 6.7 mmol) with sodium azide (478.9 mg, 7.37 mmol) in DMF (39 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1/8) gave azide **4l** (1.41 g, 86 %) as a colorless oil.

Colorless oil; R_f value 0.34(ethyl acetate / hexane = 1 / 3); IR (NaCl, neat) v_{max} 3419, 2940, 2099, 1639, 1596, 1489, 1248 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.50(d, 1H, *J* = 8.0 Hz), 6.97(dd, 1H, *J* = 8.0, 1.5 Hz), 6.89(d, 1H, *J* = 1.5 Hz), 5.66(t, 1H, *J* = 1.5 Hz), 3.88(s, 3H), 3.41(t, 2H, *J* = 6.5 Hz), 2.40(td, 2H, *J* = 7.0, 1.5 Hz), 1.81(tt, 2H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 157.2, 135.0, 128.7, 127.7, 120.8, 111.5, 85.2, 80.1, 60.4, 55.8, 50.1, 27.7, 16.1; HRMS (ESI) calcd for C₁₃H₁₄ClN₃O₂Na [M+Na]⁺ 302.0672, found 302.0672.

6-azido-1-(o-tolyl)hex-2-yn-1-ol(4m)



The reaction with tosylate **2** (1.04 g, 4.37 mmol), *n*-BuLi (1.55 M in hexane, 2.95 mL, 4.66 mmol) and benzaldehyde (0.35 g, 2.91 mmol) in THF (59 mL) followed by collected the organic layer under vacuum affording the product 0.95 g. [silica gel purification (ethyl acetate / hexane = 1 / 8 to 1/2)].

Colorless oil; R_f value 0.30(ethyl acetate / hexane = 1 / 2); IR (NaCl, neat) v_{max} 3433, 1644, 1360, 1175 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.78(d, 2H, *J* = 8.0 Hz), 7.59(m, 1H), 7.32(d, 2H, *J* = 8.0 Hz), 7.23(m, 2H), 7.17(m, 1H), 5.53(d, 1H, *J* = 2.0 Hz), 4.15(t, 2H, *J* = 6.0 Hz), 2.43(s, 3H), 2.41(s, 3H), 2.35(td, 2H, *J* = 6.5, 2.0 Hz), 1.87(tt, 2H, *J* = 6.5, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 144.8, 138.6, 135.7, 132.9, 130.7, 129.8, 128.3, 127.9, 126.2, 126.1, 84.8, 81.0, 68.9, 62.2, 27.7, 21.6, 18.9, 15.2; HRMS (ESI)

calcd for C₂₀H₂₂O₄SNa [M+Na]⁺ 381.1137, found 381.1136.

Then the crude benzyl alcohol (0.95 g, 2.65 mmol) with sodium azide (0.189 g, 2.92mmol) in DMF (27 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 10 to 1/8) gave azide **4m** (0.579 g, 95 %) as a colorless oil. Colorless oil; R_f value 0.45(ethyl acetate / hexane = 1 / 2); IR (NaCl, neat) v_{max} 3420, 2098, 1646 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.63(m, 1H), 7.22–7.25(m, 2H), 7.18(m, 1H), 5.60(br, 1H), 3.40(t, 2H, *J* = 6.5 Hz), 2.44(s, 3H), 2.39(dt, 2H, *J* = 7.0, 2.0 Hz), 2.17(br, 1H, OH), 1.80(tt, 2H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 138.7, 135.7, 130.7, 128.3, 126.29, 126.23, 85.3, 80.8, 62.5, 50.2, 27.7, 18.9, 16.1; LRMS (EI) 200([M-N₂-H]⁺, 47%), 186(51), 184(71), 115(100); HRMS (EI) calcd for C₁₃H₁₆N₃O [M+H]⁺ 230.12934, found 230.1299.

7-azido-2-phenylhept-3-yn-2-ol(7a)



The reaction with tosylate **2** (1.40 g, 6.24 mmol), *n*-BuLi (1.55 M in hexane, 4.43 mL, 6.87 mmol) and ketone (0. 5 g, 4.16 mmol) in THF (42 mL) followed by followed by collected the organic layer under vacuum affording the crude product 0.95 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol with sodium azide (446.2 mg, 6.86mmol) in DMF (20 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1 / 20 to 1/10 to 1/5) gave azide **7a** (0.872 g, 91 %) as a colorless oil.

Colorless oil; R_f value 0.19(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3393, 2982, 2094, 1446 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.65(d, 2H, *J* = 6.5 Hz), 7.37(dd, 2H, *J* = 7.5, 7.5 Hz), 7.29(dd, 1H, *J* = 7.5, 7.5 Hz), 3.43(t, 2H, *J* = 6.5 Hz), 2.42(t, 2H, *J* = 6.5 Hz), 2.31(s, 1H, OH), 1.83(tt, 2H, *J* = 6.5, 6.5 Hz), 1.75(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 145.9, 128.3, 127.6, 124.8, 85.0, 83.7, 70.0, 50.2, 33.4, 27.8, 16.1; HRMS (ESI) calcd for C₁₃H₁₅N₃ONa [M+Na]⁺ 252.11128, found 252.11129.

1-(5-azidopent-1-yn-1-yl)cyclohexan-1-ol(7b)


The reaction with tosylate **2** (1.37 g, 6.11 mmol), *n*-BuLi (1.55 M in hexane, 3.62 mL, 5.60 mmol) and ketone (0.5 g, 5.09 mmol) in THF (50 mL) followed by followed by collected the organic layer under vacuum affording the crude product 1.71 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol with sodium azide (1.71 g, 5.08 mmol) in DMF (20 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1/10 to 1/5) gave azide **7b** (0.97 g, 85%) as a colorless oil.

Colorless oil; R_f value 0.58(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3383, 2933, 2856, 2097 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 3.42(t, 2H, *J* = 6.0 Hz), 2.35(t, 2H, *J* = 6.5 Hz), 1.80–1.89(m, 3H including OH), 1.78(tt, 2H, *J* = 6.5, 6.0 Hz), 1.68(m, 2H), 1.47–1.59(m, 6H); ¹³C NMR(126 MHz, CDCl₃) δ 85.1, 82.7, 68.7, 50.2, 40.1, 27.9, 25.1, 23.4, 16.0; HRMS (ESI) calcd for C₁₁H₁₇N₃ONa [M+Na]⁺ 230.1269, found 230.1273.

6-azido-1,1-diphenylhex-2-yn-1-ol (7c)



The reaction with tosylate **2** (0.543 g, 2.15 mmol), *n*-BuLi (1.58 M in hexane, 1.46 mL, 2.30 mmol) and ketone (0.28 g, 1.54 mmol) in THF (15 mL) followed by followed by collected the organic layer under vacuum affording the product 2.2 g. [silica gel purification (ethyl acetate / hexane = 1 / 8 to 1/4)].

Colorless oil; R_f value 0.25(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3502, 3060, 3029, 2958, 1598, 1491, 1449, 1360, 1175 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.75(d, 2H, J = 7.5 Hz), 7.53(m, 4H), 7.23–7.32(m, 8H), 4.15(t, 2H, J = 6.0 Hz), 2.76(s, 1H), 2.43(t, 2H, J = 6.5 Hz), 2.39(s, 3H), 1.90(tt, 2H, J = 6.5, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.1, 144.8, 132.8, 129.9, 128.2, 127.9, 127.6, 125.9, 85.5, 84.4, 74.3, 68.8, 27.7, 21.6, 15.2; LRMS (EI) 420(M⁺, 0.8%), 403(5), 343(100), 220(53), 105(71); HRMS (EI) calcd for C₂₅H₂₄O₄S (M⁺) 420.1395, found 420.1397.

Then the crude benzyl alcohol with sodium azide (0.124 g, 1.9 mmol) in DMF (5 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1/10 to 1/5) gave azide **7c** (0.32 g, 70%) as a colorless oil.

Colorless oil; R_f value 0.50(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3426, 2933, 2099, 1490, 1449 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.59(m, 4H), 7.31(m, 4H), 7.27(m, 2H), 3.43(t, 2H, *J* = 6.5 Hz), 2.75(s, 1H), 2.48(t, 2H, *J* = 7.0 Hz), 1.85(tt, 2H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.2, 128.2, 127.6, 125.9, 86.2, 84.2, 74.4, 50.2, 27.7, 16.2; HRMS (CI) calcd for C₁₈H₁₈N₃O [M+H]⁺ 292.1450, found 292.1455.

7-azido-1,1-diphenylhept-2-yn-1-ol(7d)



The reaction with tosylate (0.255 g, 1.07 mmol), *n*-BuLi (1.55 M in hexane, 0.729 mL, 1.15 mmol) and ketone (0.15 g, 0.823 mmol) in THF (8 mL) followed by followed by collected the organic layer under vacuum affording the product 0.45 g. [silica gel purification (ethyl acetate / hexane = 1 / 8 to 1/3)].

Colorless solid; R_f value 0.28(ethyl acetate/hexane = 1/3); m.p. 102.8–103.2 °C;IR (NaCl, neat) v_{max} 3477, 1644, 1174, 933 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.78(d, 2H, J = 8.0 Hz), 7.60–7.57(m, 4H), 7.38–7.31(m, 6H), 7.28–7.25(m, 2H), 4.06(t, 2H, J = 6.5 Hz), 2.92(s, 1H, OH), 2.44(s, 3H), 2.34(t, 2H, J = 7.0 Hz), 1.82(tt, 2H, J = 6.5, 6.5 Hz), 1.63(tt, 2H, J = 6.5, 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.4, 144.8, 132.9, 129.9, 128.2, 127.8, 127.5, 125.9, 86.9, 83.9, 74.4, 69.9, 28.0, 24.3, 21.6, 18.3; HRMS (ESI) calcd for C₂₆H₂₆O₄SNa [M+Na]⁺457.14495, found 457.14440.

Then the crude benzyl alcohol with sodium azide (83.47 mg, 1.28 mmol) in DMF (20 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1/10 to 1/5) gave azide **7d** (0.108 g, 43%) as a colorless oil.

Colorless oil; R_f value 0.47(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3421, 2935, 2096, 1489, 1449 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.60(m, 4H), 7.33(dd, 4H, *J* = 6.5, 6.5 Hz), 7.27(m, 2H), 3.31(t, 2H, *J* = 6.5 Hz), 2.76(s, 1H, OH), 2.40(t, 2H, *J* = 7.0 Hz), 1.66–1.76(m, 4H); ¹³C NMR(126 MHz, CDCl₃) δ 145.3, 128.2, 127.6, 125.9, 87.2, 83.7, 74.4, 50.9, 28.0, 25.6, 18.5,; HRMS (ESI) calcd for C₁₉H₁₉N₃ONa [M+Na]⁺ 328.1426, found 328.1424.

1,11-diazidoundeca-4,7-diyn-6-ol(12a)



The reaction with tosylate (1.54 g, 6.58 mmol), *n*-BuLi (1.55 M in hexane, 4.39 mL, 6.58 mmol) and aldehyde (0.28 g, 2.74 mmol) in THF (30 mL) followed by followed by collected the organic layer under vacuum affording the crude product 0.33 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol with sodium azide (94.38 mg, 1.45 mmol) in DMF (15 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1/2) gave azide **12a** (0.128 g, 79%) as a colorless oil.

Colorless oil; R_f value 0.36(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3390, 2935, 2098, 1255 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 5.10(tt, 1H, *J* = 1.5, 1.5 Hz), 3.42(t, 4H, *J* = 7.0 Hz), 2.36(td, 4H, *J* = 6.5, 1.5 Hz), 2.19(m, 1H, OH), 1.80(tt, 4H, *J* = 7.0, 6.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 83.3, 78.9, 52.3, 50.1, 27.4, 16.0; HRMS (ESI) calcd for C₁₁H₁₄N₆ONa [M+Na]⁺ 269.1127, found 269.1123.

1,11-diazido-6-pentylundeca-4,7-diyn-6-ol(12b)



The reaction with tosylate (0.915 g, 3.84 mmol), *n*-BuLi (1.55 M in hexane, 2.8 mL, 4.22 mmol) and ester (0.25 g, 1.92 mmol) in THF (20 mL) followed by followed by collected the organic layer under vacuum affording the crude product 1.10 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol with sodium azide (299.6 mg, 4.61 mmol) in DMF (38 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1/15 to 1/5 to 1/1) gave azide **12b** (0.521 g, 86%) as a colorless oil.

Colorless oil; R_f value 0.45(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3420, 2930, 2098, 1254 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 3.40(t, 4H, *J* = 7.0 Hz), 2.45(s, 1H, OH), 2.35(t, 4H, *J* = 7.0 Hz), 1.83(m, 2H), 1.78(tt, 4H, *J* = 7.0, 7.0 Hz), 1.54(m, 2H), 1.31–1.35(m, 4H), 0.90(t, 3H, *J* = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 82.2, 81.9, 63.7, 50.1, 44.1, 31.4, 27.5, 24.3, 22.5, 16.0, 14.0; HRMS (ESI) calcd for C₁₆H₂₄N₆ONa [M+Na]⁺ 339.1909, found 339.1901.

1,11-diazido-6-phenylundeca-4,7-diyn-6-ol(12c)



The reaction with tosylate (1.19 g, 4.98 mmol), *n*-BuLi (1.55 M in hexane, 3.49 mL, 5.23 mmol) and chloride (0.35 g, 2.49 mmol) in THF (25 mL) followed by followed by collected the organic layer under vacuum affording the crude product 2.89 g. The crude product can be used to the next step without further purification.

Then the crude benzyl alcohol with sodium azide (388.5 mg, 5.98 mmol) in DMF (25 ml) followed by silica gel column chromatography (ethyl acetate / hexane = 1/20 to 1/10 to 1/2) gave azide **12c** (0.724 g, 90%) as a colorless oil.

Colorless oil; R_f value 0.51(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3398, 2935, 2099, 1449, 1255 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.76(m, 2H), 7.35–7.41(m, 3H), 3.41(t, 4H, *J* = 7.0 Hz), 2.86(m, 1H, OH), 2.42(t, 4H, *J* = 7.0 Hz), 1.82(tt, 4H, *J* = 7.0, 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 142.4, 128.5, 128.4, 125.7, 83.9, 82.0, 65.0, 50.1, 27.5, 16.1; HRMS (ESI) calcd for C₁₇H₁₈N₆ONa [M+Na]⁺ 345.1440, found 345.1436.

General Experimental Procedure of triazolations



To the mixture of propargyl alcohol (1.0 equiv) in dichloromethane (0.1 M to alcohols) under nitrogen atmosphere, $TsOH \cdot H_2O$ (1.2 equiv) was added at ambient temperature. After 30 minutes, the reaction was quenched with saturated sodium bicarbonate aqueous solution, and was washed with brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography afforded triazole.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(phenyl)methanol(5a)



The reaction with propargyl alcohol **4a** (50 mg, 0.232 mmol) and TsOH·H₂O (53.0 mg, 0.278 mmol) in dichloromethane (2.3 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/2 to ethyl acetate) afforded triazole **5a** (16.1 mg, 32%) along with tosylated byproduct in 6%.

Colorless oil; R_f value 0.15(dichloromethane/mechanol = 20/1); IR (NaCl, neat) v_{max} 3250, 1228, 1487, 1087, 1013, 805 cm⁻¹; cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.43(d, 2H, J = 7.0 Hz), 7.34(dd, 1H, J = 7.0, 7.5 Hz), 7.28(t, 2H, J = 7.5 Hz), 6.01(sd, 1H, J = 2.5 Hz), 4.20(t, 2H, J = 7.0 Hz), 4.12(sd, 1H, J = 3.0 Hz), 2.65-2.55(m, 2H), 2.48-2.42(m, 1H), 2.28-2.22(m, 1H); ¹³C NMR(126 MHz, CDCl₃) δ 143.0, 141.8, 139.4, 128.3, 127.6, 126.1, 68.6, 46.2, 28.0, 20.6; HRMS (ESI) calcd for C₁₂H₁₁NONa [M +Na]⁺ 238.0956, found 238.0955.

(Z)-1-(3, 4-dihydro-2*H*-pyrrol-5-yl)-2-phenylvinyl trifluoromethanesulfonate (5a')



To a stirred solution of propargyl alcohol **1** (68.0 mg, 0.316 mmol) in dichloromethane (3 mL) under an atmosphere nitrogen was added dropwise TMSOTf (68.5 μ L, 0.379 mmol) at 0 °C. After 20 min, the reaction was quenched with saturated NaHCO₃ aqueous solution at 0 °C. The mixture was diluted with ethyl acetate and was washed with water and brine. Drying collected organic layer over MgSO₄ followed by silica gel column chromatography (ethyl acetate / hexane = 1/10 to 1/5 to 1/4) gave imine product **5a'** (7.1 mg, 7%) as a colorless solid along with triazole ketone **5ab** in 13%.

White crystal; R_f value 0.28(ethyl acetate/hexane = 1/3); m.p. 109.1–112.7 °C; IR (NaCl, neat) v_{max} 3385, 3035, 2971, 1258, 1049, 740 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.56(d, 2H, J = 6.5 Hz), 7.41(m, 3H), 6.78(s, 1H), 4.15(t, 2H, J = 7.0 Hz), 2.85(t, 2H, J = 8.5 Hz), 2.09(tt, 2H, J = 7.0, 8.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 167.5, 142.5, 130.8, 129.9, 129.8, 128.7, 126.9, 118.2(q, J = 322 Hz), 61.9, 34.5, 23.2; HRMS (ESI) calcd for C₁₃H₁₂F₃NONaS [M+Na]⁺ 342.03877, found 342.03882.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(phenyl)methanone(5ab)



White crystal; R_f value 0.13(ethyl acetate/hexane = 1/3); m.p. 141.7–146.7 °C; IR (NaCl, neat) v_{max} 2958, 2871, 1694, 1263, 893, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 8.45(d, 2H, J = 8.0 Hz), 7.60(t, 1H, J = 7.0 Hz), 7.52(dt, 2H, J = 8.0, 7.0 Hz), 4.43(t, 2H, J = 7.0 Hz), 3.23(t, 2H, J = 7.5 Hz), 2.91(tt, 2H, J = 7.0, 7.5 Hz); ¹³C NMR(126 MHz,

CDCl₃) δ 186.1, 148.6, 140.4, 136.6, 133.0, 130.5, 128.3, 46.7, 28.2, 22.4; HRMS (ESI) calcd for C₁₂H₁₁NONa [M-N₂+Na]⁺ 208.07383, found 208.07385.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(3,4-dimethoxyphenyl)methanol(5 c)



The reaction with propargyl alcohol **4c** (150 mg, 0.545 mmol) and TsOH·H₂O (124.4 mg, 0.654 mmol) in dichloromethane (5 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/2 to ethyl acetate) afforded triazole **5c** (106.7 mg, 71%). White solid; R_f value 0.10(ethyl acetate/hexane = 1/1); m.p. 125.4–125.7 °C; IR (NaCl, neat) v_{max} 3433, 1644, 1514, 1234, 1137 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.01(d, 1H, J = 2.0 Hz), 6.94(dd, 1H, J = 8.0, 2.0 Hz), 6.84(d, 1H, J = 8.0 Hz), 5.95(s, 1H), 4.24(t, 2H, J = 7.5 Hz), 2.77(br, 1H, OH), 2.63(m, 2H), 2.47(ddd, 1H, J = 15.1, 8.5, 6.0 Hz), 2.33(ddd, 1H, J = 15.1, 8.0, 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 149.0, 148.5, 143.0, 139.2, 134.4, 118.4, 110.8, 109.3, 68.6, 55.9, 46.2, 28.0, 20.7; HRMS (ESI) calcd for C₁₄H₁₇N₃O₃Na [M+Na]⁺ 298.1168, found 298.1170.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(3,4,5-trimethoxyphenyl)methano l(5d)



The reaction with propargyl alcohol 4d (200 mg, 0.655 mmol) and TsOH·H₂O (149.5 mg, 0.786 mmol) in dichloromethane (6.6 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/21/1 ethyl acetate to to then methanol/dichloromethane = 30/1 to 20/1) afforded triazole **5d** (106.7 mg, 71%). White crystal; R_f value 0.22(dichloromethane/mechanol = 20/1); m.p. 137.8–138.2 °C; IR (NaCl, neat) v_{max} 3433, 1233, 1124 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.66(s, 2H), 5.91(s, 1H), 4.17(t, 2H, J = 7.5 Hz), 3.78 (s, 6H), 3.77(s, 3H), 2.48–2.64(m, 3H), 2.29(ddd, 1H, J = 16.0, 9.5, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 153.0, 143.0, 139.2, 137.7, 137.0, 102.8, 68.3, 60.7, 56.0, 46.1, 27.9, 20.7; HRMS (ESI) calcd for C₁₅H₁₉N₃O₄Na [M+Na]⁺ 328.1273, found 328.1276.

(4-chlorophenyl)(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)methanol(5e)



The reaction with propargyl alcohol **4e** (200 mg, 0.801 mmol) and TsOH·H₂O (182.8 mg, 0.961 mmol) in dichloromethane (8 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/5 to 1/2 to 1/1 to 2/1) afforded triazole **5e** (121.7 mg, 61%).

White crystal; R_f value 0.2(ethyl acetate/hexane = 1/2); m.p. 142.8–142.9 °C; IR (NaCl, neat) v_{max} 3250, 1228, 1487, 1087, 1013, 805 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.38(d, 2H, J = 8.5 Hz), 7.32(d, 2H, J = 8.5 Hz), 6.00(s, 1H), 4.23(t, 2H, J = 7.5 Hz), 2.64(m, 2H), 2.47(ddd, 1H, J = 15.0, 8.5, 6.5 Hz), 2.29(ddd, 1H, J = 15.0, 9.0, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 142.6, 140.3, 139.3, 133.4, 128.5, 127.6, 68.0, 46.3, 28.0, 20.7; HRMS (ESI) calcd for C₁₂H₁₂ClN₃ONa [M+Na]⁺ 272.0567, found 272.0570.

(2,4-dichlorophenyl)(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)methanol(5f)



The reaction with propargyl alcohol **4f** (150 mg, 0.528 mmol) and TsOH·H₂O (120.5 mg, 0.633 mmol) in dichloromethane (5 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/5 to 1/2 to 1/1 to 2/1 to ethyl acetate then methanol / dichloromethane = 20/1) afforded triazole **5f** (44 mg, 44%).

White crystal; R_f value 0.1(ethyl acetate/hexane = 1/2); m.p. 176.8–170.0 °C; IR (NaCl, neat) v_{max} 3224, 2879, 1589, 1035, 858 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.76(d, 1H, J = 8.5 Hz), 7.36(d, 1H, J = 1.0 Hz), 7.33(dd, 1H, J = 8.5, 1.0 Hz), 6.30(s, 1H), 4.43(br, 1H, OH), 4.25(t, 2H, J = 7.0 Hz), 2.65(tdd, 2H, J = 8.0, 7.5, 7.0 Hz), 2.37(td, 1H, J = 16.0, 7.5 Hz), 2.30(td, 1H, J = 16.0, 8.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 141.0, 139.3, 137.9, 133.9, 132.6, 129.0, 128.9, 127.4, 64.9, 46.3, 28.0, 20.5; HRMS (ESI) calcd for C₁₂H₁₁Cl₂N₃ONa [M+Na]⁺ 306.0177, found 306.0177.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(2-methoxyphenyl)methanol(5h)



The reaction with propargyl alcohol **4h** (80 mg, 0.326 mmol) and TsOH·H₂O (74.5 mg, 0.19 mmol) in dichloromethane (3 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/2 to 1/1 to 2/1) afforded triazole **5h** (56.1 mg, 70%), along with imine **6h** (3.5mg, 6%).

White solid; R_f value 0.43(methaol/dichloromethane = 1/10); m.p. 119.1–119.4 °C; IR (NaCl, neat) v_{max} 3340, 1490, 1240, 1028 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.47(dd, 1H, J = 7.5, 2.0 Hz), 7.27(ddd, 1H, J = 8.0, 7.5, 2.0 Hz), 6.98(dd, 1H, J = 8.0, 8.0 Hz), 6.87(d, 1H, J = 8.0 Hz), 6.23(d, 1H, J = 5.0 Hz), 4.23(t, 2H, J = 7.0 Hz), 3.80(s, 3H), 3.69(d, 1H, OH, J = 5.0 Hz), 2.63(tt, 2H, J = 7.5, 7.5 Hz), 2.38(td, 1H, J = 15.0, 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 156.2, 142.1, 139.2, 129.9, 128.7, 127.2, 120.7, 110.3, 64.7, 55.4, 46.1, 28.0, 20.6; HRMS (ESI) calcd for C₁₃H₁₅N₃O₂ [M+Na]⁺ 268.1062, found 268.1062.

5-(benzofuran-2-yl)-3,4-dihydro-2H-pyrrole(6h)



Colorless oil; R_f value 0.48 (methanol/dichloromethane = 1/10); IR (NaCl, neat) v_{max} 3423, 1630, 824, 599 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.62(d, 1H, J = 8.0 Hz), 7.56(d, 1H, J = 7.5 Hz), 7.35(dd, 1H, J = 8.0, 7.5 Hz), 7.26(dd, 1H, J = 7.5, 7.5 Hz), 7.10(s, 1H), 4.14(t, 2H, J = 8.0 Hz), 2.97(t, 2H, J = 8.0 Hz), 2.07(tt, 2H, J = 8.0, 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 164.7, 155.4, 151.3, 127.8, 126.2, 123.2, 121.9, 111.9, 109.2, 62.0, 35.0, 22.5; HRMS (ESI) calcd for C₁₂H₁₂NO [M+H]⁺ 186.0919, found 186.0918.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(2,4-dimethoxyphenyl)methanol(5 i)



The reaction with propargyl alcohol 4i (127.3 mg, 0.462 mmol) and dichloroacetic acid

(0.045 mL, 0.555 mmol) in dichloromethane (4.6 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/6 to 1/2 to 1/1 to 2/1 then methanol / dichloromethane = 20/1 to 10/1) afforded triazole **5i** (56.1 mg, 70%).

White solid; R_f value 0.23(methaol/dichloromethane = 1/10); m.p. 124.4–124.6 °C; IR (NaCl, neat) v_{max} 3433, 2097, 1646, 1030 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.33(d, 1H, J = 8.0 Hz), 6.48(dd, 1H, J = 8.0, 2.0 Hz), 6.44(d, 1H, J = 2.5 Hz), 6.14(s, 1H), 4.23(t, 2H, J = 7.5 Hz), 3.79(s, 3H), 3.76(s, 3H), 2.64(tt, 2H, J = 7.5, 7.5 Hz), 2.52(dt, 1H, J = 16.5, 7.5 Hz), 2.43(dt, 1H, J = 16.5, 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 160.4, 157.4, 142.4, 139.1, 128.0, 122.6, 104.1, 98.4, 64.6, 55.4, 55.3, 46.1, 28.0, 20.7; HRMS (ESI) calcd for C₁₄H₁₇N₃O₃Na [M+Na]⁺ 298.1168, found 298.1177.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(2,4,6-trimethoxyphenyl)methano l(5j)



The reaction with propargyl alcohol **4j** (117.7 mg, 0.386 mmol) and chloroacetic acid (43.7 mg, 0.463 mmol) in dichloromethane (3.8 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/5 to 1/1 to ethyl acetate then methanol / dichloromethane = 10/1) afforded triazole **5j** (67.7 mg, 57%).

White solid; R_f value 0.33(methaol/dichloromethane = 1/10); m.p. 139.3–139.9 °C; IR (NaCl, neat) v_{max} 3445, 1607, 1205, 1150 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.30(s, 1H), 6.14(s, 2H), 4.22(t, 2H, J = 7.5 Hz), 3.74(s, 9H), 2.50–2.68(m, 4H), 2.06(br, 1H); ¹³C NMR(126 MHz, CDCl₃) δ 160.8, 158.4, 142.9, 138.6, 110.2, 90.9, 62.3, 55.8, 55.3, 46.0, 28.1, 20.7; HRMS (ESI) calcd for C₁₅H₁₉N₃O₄Na [M+Na]⁺ 328.1273, found 328.1272.

2-((5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(hydroxy)methyl)phenol(5k)



The reaction with propargyl alcohol **4k** (65.0 mg, 0.281 mmol) and TsOH·H₂O (64.2 mg, 0.259 mmol) in dichloromethane (2 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/2 to ethyl acetate then methanol/dichloromethane = 10/1)

afforded triazole 5k (26.5 mg, 41%).

Colorless oil; R_f value 0.2(methaol/dichloromethane = 1/10); IR (NaCl, neat) v_{max} 3196, 1456, 1240, 911, 755 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 8.92(br, 1H, OH), 7.21(dd, 1H, J = 8.0, 7.5 Hz), 7.02(d, 1H, J = 7.0 Hz), 6.94(d, 1H, J = 8.0 Hz), 6.84(dd, 1H, J = 7.5, 7.5 Hz), 6.17(s, 1H), 5.96(br, 1H, OH), 4.24(t, 2H, J = 8.0 Hz), 2.51–2.66(m, 3H), 2.14(m, 1H); ¹³C NMR(126 MHz, CDCl₃) δ 156.2, 141.9, 140.0, 129.4, 127.6, 124.4, 119.8, 117.7, 69.3, 46.5, 27.9, 20.3; HRMS (ESI) calcd for C₁₂H₁₃N₃O₂ [M+Na]⁺ 254.0906, found 254.0909.

(4-chloro-2-methoxyphenyl)(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)meth anol(5l)



The reaction with propargyl alcohol **4l** (74.1 mg, 0.265 mmol) and TsOH·H₂O (60.5 mg, 0.318 mmol) in dichloromethane (2.7 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/5 to 2/1 ethyl acetate) afforded triazole **5l** (51.8 mg, 70%) along with benzofuran (6%).

White solid; R_f value 0.25(methaol/dichloromethane = 1/10); m.p. 154.9–155.1 °C; IR (NaCl, neat) v_{max} 3422, 2065, 1645, 1244 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.45(d, 1H, J = 8.0 Hz), 6.98(dd, 1H, J = 8.0, 2.0 Hz), 6.86(d, 1H, J = 2.0 Hz), 6.20(s, 1H), 4.25(t, 2H, J = 8.0 Hz), 3.78(s, 3H), 2.67(tt, 2H, J = 8.0, 8.0 Hz), 2.50(td, 1H, J = 15.0, 7.5 Hz), 2.39(td, 1H, J = 15.0, 8.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 156.8, 141.8, 139.1, 134.1, 128.7, 128.1, 120.8, 111.1, 63.9, 55.7, 46.2, 28.1, 20.6; HRMS (ESI) calcd for C₁₃H₁₄ClN₃O₂Na [M+Na]⁺ 302.06722, found 302.0676.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(o-tolyl)methanol(5m)



The reaction with propargyl alcohol **4m** (188.7 mg, 0.823 mmol) and TsOH·H₂O (187.9 mg, 0.988 mmol) in dichloromethane (8 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/5 to 1/2 to 1/1 to 2/1) afforded triazole **5m** (117.5 mg, 63%).

White solid; R_f value0.25(methaol/dichloromethane = 1/20); m.p. – °C; IR (NaCl, neat) v_{max} 3420, 2064, 1646, 1031 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.67(d, 1H, *J* = 7.0 Hz), 7.26(dd, 1H, *J* = 8.0, 7.0 Hz), 7.20(ddd, 1H, *J* = 8.0, 7.0, 1.5 Hz), 7.13(d, 1H, *J* = 7.0 Hz), 6.20(s, 1H), 4.22(t, 2H, *J* = 7.0 Hz), 3.30(br, 1H, OH), 2.58(m, 2H), 2.35(ddd, 1H, *J* = 16.5, 9.0, 5.5 Hz), 2.21(s, 3H), 2.07(ddd, 1H, *J* = 16.5, 9.0, 6.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 142.1, 139.7, 139.5, 134.7, 130.3, 127.5, 126.1, 125.5, 65.7, 46.2, 28.0, 20.4, 19.1; LRMS (EI) 229(M⁺, 100%), 184(57); HRMS (EI) calcd for C₁₃H₁₅N₃O (M⁺) 229.1215, found 229.1212.

1-(5,6-dihydro-4H-pyrrolo[1,2-c][1,2,3]triazol-3-yl)-1-phenylethan-1-ol(8a)



The reaction with propargyl alcohol **7a** (154 mg, 0.671 mmol) and TsOH·H₂O (153 mg, 0.806 mmol) in dichloromethane (6.7 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/5 to 1/2 to 1/1 to 2/1) afforded triazole **8a** (82.9 mg, 54%) along with byproduct (26.7 mg, 19%).

Colorless oil; R_f value 0.28(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3362, 2980, 1446, 1066, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.48(d, 2H, *J* = 8.0 Hz), 7.33(dd, 2H, *J* = 8.0, 7.5 Hz), 7.25(dd, 1H, *J* = 7.0, 6.5 Hz), 4.25(t, 2H, *J* = 7.5 Hz), 3.26(s, 1H, OH), 2.66(m, 2H), 2.49(m, 2H), 1.99(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 146.4, 146.2, 138.6, 128.1, 127.0, 125.2, 71.9, 46.2, 30.0, 28.1, 21.2,; HRMS (ESI) calcd for C₁₃H₁₅N₃O [M+Na]⁺ 252.1113, found 252.1118.

3-(cyclohex-1-en-1-yl)-5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazole(8b')





The reaction with propargyl alcohol (100 mg, 0.482 mmol) and TsOH·H₂O (110 mg, 0.579 mmol) in dichloromethane (4.8 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/4 to 1/2 to 1/1) afforded triazole **8b'** (51.7 mg, 50%) along with byproduct (30.1 mg, 30%).

Colorless oil; R_f value 0.13(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 2926, 1558 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.19(m, 1H), 4.30(t, 2H, *J* = 7.5 Hz), 2.96(t, 2H, *J* =

8.0 Hz), 2.80 (tt, 2H, J = 8.0, 7.5 Hz), 2.21-2.18(m, 2H), 1.78-1.73(m, 2H), 1.68-1.63(m, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 141.4, 137.2, 128.6, 124.2, 46.0, 28.3, 26.0, 25.3, 22.5, 22.2, 22.0; HRMS (ESI) calcd for C₁₁H₁₆N₃ [M+H]⁺ 190.1344, found 190.1338.

1-(5-azidopent-1-yn-1-yl)cyclohex-1-ene(10)



Colorless oil; R_f value 0.67(ethyl acetate/hexane = 1/2; IR (NaCl, neat) v_{max} 2930, 2858, 2097, 1255, 917 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.02(m, 1H), 3.41(t, 2H, *J* = 7.0 Hz), 2.41(t, 2H, *J* = 7.0 Hz), 2.05–2.09(m, 4H), 1.78(tt, 2H, *J* = 7.0, 7.0 Hz), 1.53–1.64(m, 4H); ¹³C NMR(126 MHz, CDCl₃) δ 133.8, 120.6, 85.1, 83.4, 50.2, 29.4, 28.0, 25.5, 22.3, 21.5, 16.5; HRMS (ESI) calcd for C₁₁H₁₆N₃ [M+H]⁺ 190.1344, found 190.1348.

(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)dip henylmethanol(8c)



The reaction with propargyl alcohol **7c** (42.0 mg, 0.144 mmol) and TsOH·H₂O (38.5 mg, 0.173 mmol) in dichloromethane (2 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/4 to 1/1) afforded triazole **8c** (41.8 mg, 99%).

White crystal; R_f value 0.24(ethyl acetate/hexane = 1/1); m.p. 113–114 °C; IR (NaCl, neat) v_{max} 3378, 3059, 1491, 1448, 1316, 1168, 1021 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.25–7.34(m, 10H), 4.24(t, 2H, J = 7.5 Hz), 4.19(br-s, 1H, OH), 2.56(tt, 2H, J = 7.5, 7.5 Hz), 2.07(t, 2H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.5, 145.1, 140.2, 127.8, 127.34, 127.29, 76.6, 46.2, 27.9, 20.8; HRMS (ESI) calcd for C₁₈H₁₇N₃ONa [M+Na]⁺ 314.1269, found 314.1267.

diphenyl(4,5,6,7-tetrahydro-[1,2,3]triazolo[1,5-a]pyridin-3-yl)methanol(8d)



The reaction with propargyl alcohol 7d (62.0 mg, 0.203 mmol) and TsOH·H₂O (54.2

mg, 0.244 mmol) in dichloromethane (2 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/2 to 1/1) afforded triazole **8d** (62.0 mg, 99%).

Colorless oil; R_f value 0.11(ethyl acetate/hexane = 1/3); IR (NaCl, neat) v_{max} 3376, 2953, 1490, 1447, 1016, 759, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.28–7.31(m, 10H), 4.34(t, 2H, *J* = 6.0 Hz), 4.28(br, 1H, OH), 2.00(t, 2H, *J* = 7.0 Hz), 1.93(m, 2H), 1.67(m, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 147.7, 145.2, 131.0, 127.9, 127.7, 127.5, 77.4, 46.6, 22.2, 20.6, 19.9; HRMS (ESI) calcd for C₁₉H₁₉N₃ONa [M+Na]⁺ 328.1426, found 328.1426.

bis(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)methanone (13a)



The reaction with propargyl alcohol (150 mg, 0.610 mmol) and TMSOTf (0.132 mL, 0.731 mmol) in dichloromethane (6 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/5 to 1/2 to 1/1 to ethyl acetate then methaol/dichloromethane = 1/10) afforded triazole **13a** (20.1 mg, 13%).

Colorless oil; R_f value 0.1(methaol/dichloromethane = 1/20); IR (NaCl, neat) v_{max} 2961, 2098, 1645, 1572, 1195, 1644, 1571, 930 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 4.43(t, 4H, J = 7.5 Hz), 3.33(t, 4H, J = 7.5 Hz), 2.88(tt, 4H, J = 7.5, 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 177.5, 147.5, 139.0, 46.8, 28.0, 23.4; HRMS (ESI) calcd for C₁₁H₁₂N₆ONa [M+Na]⁺ 267.0970, found 267.0971.

1,1-bis(5,6-dihydro-4*H*-pyrrolo[1,2-c][1,2,3]triazol-3-yl)hexan-1-ol(13b)



The reaction with propargyl alcohol **12b** (117.6 mg, 0.372 mmol) and TMSOTf (0.081 mL, 0.446 mmol) in dichloromethane (4 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/5 to 1/2 to 1/1 to ethyl acetate then methaol/dichloromethane = 1/30 to 1/20 to 1/10) afforded triazole **13b** (67.6mg, 58%).

White crystal; R_f value 0.1(methaol/dichloromethane = 1/20); m.p. 108.8–109.7 °C; IR (NaCl, neat) v_{max} 3349, 2953, 2868, 1455, 1313, 911, 729 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 4.31(s, 1H, OH), 4.24(t, 4H, *J* = 7.5 Hz), 2.93(m, 4H), 2.71(tt, 4H, *J* = 7.5, 7.5

Hz), 2.19(m, 2H), 1.24(m, 6H), 0.81(m, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 144.7, 138.9, 71.3, 46.3, 41.4, 31.8, 28.1, 22.9, 22.5, 21.6, 14.0; HRMS (ESI) calcd for C₁₆H₂₄N₆ONa [M+Na]⁺ 339.1909, found 339.1903.

1-azido-6-(5-azidopent-1-yn-1-yl)undec-6-en-4-yne(14)



Colorless oil; R_f value 0.24(methaol/dichloromethane = 1/20); IR (NaCl, neat) v_{max} 2931, 2097, 1255 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 6.19(t, 1H, J = 8.0 Hz), 3.41–3.45(m, 4H), 2.49(t, 2H, J = 6.5 Hz), 2.42(t, 2H, J = 6.5 Hz), 2.30(td, 2H, J = 8.0, 8.0 Hz), 1.78–1.85(m, 4H), 1.30–1.42(m, 4H), 0.90(t, 3H, J = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 148.0, 105.0, 91.5, 84.9, 80.0, 50.21, 50.19, 30.7, 30.3, 27.8, 27.7, 22.3, 16.8, 16.5, 13.9; HRMS (CI) calcd for C₁₆H₂₃N₆ [M+H]⁺ 299.1984, found 299.1985.

bis(5,6-dihydro-4H-pyrrolo[1,2-c][1,2,3]triazol-3-yl)(phenyl)methanol(13c)



The reaction with propargyl alcohol (117.6 mg, 0.372 mmol) and TMSOTf (0.081 mL, 0.446 mmol) in dichloromethane (4 ml) followed by silica gel column purification (ethyl acetate/hexane = 1/5 to 1/2 to 1/1 to ethyl acetate then methaol/dichloromethane = 1/30 to 1/20 to 1/10) afforded triazole **13c** (67.6mg, 58%).

White crystal; R_f value 0.14(dichloromethane/methanol = 10/1); m.p. 157.7–157.8 °C; IR (NaCl, neat) v_{max} 3284, 2965, 1448, 1171, 751 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.56(m, 2H), 7.29(m, 2H), 7.25(m, 1H), 5.72(s, 1H, OH), 4.18–4.30(m, 4H), 2.43–2.64(m, 8H); ¹³C NMR(126 MHz, CDCl₃) δ 144.8, 143.9, 140.2, 127.9, 127.4, 126.3, 71.9, 46.3, 27.9, 21.3; HRMS (ESI) calcd for C₁₇H₁₈N₆ONa [M+Na]⁺ 345.1440, found 345.1430.

General Procedure of Preparation of Starting Materials (15a-k, 15m, 15o-s)

$$\begin{array}{c} R^{1} & \xrightarrow{\qquad} R^{3} & R^{1} \\ R^{2} & \xrightarrow{n-\text{BuLi or LHMDS}} & R^{2} \xrightarrow{\qquad} R^{3} \\ & \text{THF, RT} \\ & -78 \ ^{\circ}\text{C to RT} \end{array}$$

To the mixture of corresponding terminal alkyne (1.3-2.0 equiv) in THF (0.2–0.1 M for ketone) was added *n*-butyllithium in hexane or lithium hexamethyldisilazide in THF (same to alkynes) was added dropwise at -78 °C. After 30 minutes, ketone or aldehyde (1 equiv) was added at the same temperature and the mixture was gradually warmed up to room temperature. Checking the consumption of ketone, saturated ammonium chloride aqueous solution was added. The mixture was extracted with ethyl acetate and washed with saturated ammonium chloride aqueous solution and brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography afforded propargyl alcohols for triazolations.

1,1-Diphenylhept-2-yn-1-ol (15a)



The reaction with 1-hexyne (0.53 mL, 4.6 mmol), *n*-butyllithium (1.58 M in hexane, 3.13 mL, 4.94 mmol) and benzophenone (600 mg, 3.3 mmol) in THF (33 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/8) afforded **15a** (855 mg, 3.23 mmol, 98%).

Colorless oil; R_f value 0.45(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3446, 2957, 2931, 1449, 1002, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.55(d, 4H, J = 8.0 Hz), 7.26(m, 4H), 7.19(m, 2H), 2.65(m, 1H, OH), 2.29(t, 2H, J = 7.0 Hz), 1.52(m, 2H), 1.40(m, 2H), 0.87(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.5, 128.1, 127.4, 125.9, 88.3, 82.9, 74.4, 30.6, 22.0, 18.6, 13.6; LRMS (EI) 264(M⁺, 28%), 221(95), 208(46), 207(100), 187(79); HRMS (EI) calcd for C₁₉H₂₀O (M⁺) 264.1514, found 264.1516.

1,1'-Bis(4-chlorophenyl)hept-2-yn-1-ol (15b)



The reaction with 1-hexyne (0.3 mL, 2.59 mmol), LHMDS (1.0 M in THF, 2.6 mL, 2.6 mmol) and 4,4'-dichlorobenzophenone (500 mg, 2.06 mmol) in THF (10 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/100 to 1/50 to 1/25) afforded **15b** (654.4 mg, 1.964 mmol, 99%).

Light yellow oil; R_f value 0.25(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 3411, 2958, 2932, 2871, 2229, 1905, 1488, 1092, 1013, 815 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.50(d, 4H, J = 9.0 Hz), 7.28(d, 4H, J = 9.0 Hz), 2.70(s, 1H, OH), 2.33(t, 2H, J = 7.0 Hz), 1.56(tt, 2H, J = 7.5, 7.0 Hz), 1.43(qt, 2H, J = 7.5, 7.5 Hz), 0.93(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 143.7, 133.6, 128.4, 127.4, 89.1, 82.1, 73.5, 30.5, 22.0, 18.5, 13.6; LRMS (EI) 332(M⁺, 13%), 299(29), 298(24), 297(94), 275(59), 221(86), 139(100); HRMS (EI) calcd for C₁₉H₁₈Cl₂O (M⁺) 332.0735, found 332.0734.

1,1'-Bis(4-methoxyphenyl)hept-2-yn-1-ol (15c):



The reaction with 1-hexyne (0.47 mL, 4.13 mmol), LHMDS (1.0 M in THF, 4.2 mL, 4.2 mmol) and 4,4'-dimethoxybenzophenone (500 mg, 2.06 mmol) in THF (10 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/20 to 1/15 to 1/10) afforded **15c** (669.5 mg, 2.064 mmol, 100%).

Light yellow oil; R_f value 0.36(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3462, 2956, 2932, 2836, 1507, 1248, 1173, 1033, 827 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.50(d, 2H, J = 9.0 Hz), 6.84(d, 2H, J = 9.0 Hz), 3.79(s, 6H), 2.70(s, 1H, OH), 2.33(t, 2H, J = 7.0 Hz), 1.57(tt, 2H, J = 7.5, 7.0 Hz), 1.45(qt, 2H, J = 7.5, 7.5 Hz), 0.94(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 158.8, 138.1, 127.2, 113.3, 87.8, 83.3, 73.8, 55.2, 30.7, 22.0, 18.6, 13.6; LRMS (EI) 324(22%, M⁺), 281(14), 267(27), 135(21),

83(100); HRMS (EI) calcd for C₂₁H₂₄O₃ (M⁺) 324.1725, found 324.1724.

2-Phenyloct-3-yn-2-ol (15d):

$$Ph \xrightarrow{Me} nBu$$

HO 15d

The reaction with 1-hexyne (0.67 mL, 5.8 mmol), LHMDS (1.0 M in THF, 6.2 mL, 6.2 mmol) and acetophenone (486 mg, 4.16 mmol) in THF (42 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/20) afforded **15d** (549 mg, 2.71 mmol, 65%).

Colorless oil; R_f value 0.46(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3397, 2958, 1447, 921, 763, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.66(d, 2H, *J* = 7.5 Hz), 7.36(dd, 2H, *J* = 7.5, 8.0 Hz), 7.29(t, 1H, *J* = 8.0 Hz), 2.33(br, 1H), 2.29(t, 2H, *J* = 7.5 Hz), 1.75(s, 3H), 1.53(tt, 2H, *J* = 7.5, 7.5 Hz), 1.44(tq, 2H, *J* = 7.5, 7.5 Hz), 0.93(t, 3H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 146.2, 128.2, 127.5, 124.9, 85.7, 83.7, 70.1, 33.5, 30.7, 22.0, 18.4, 13.6; HRMS (CI) calcd for C₁₄H₁₉O [M+H]⁺ 203.1436, found 203.1438.

9-(Hex-yn-1-yl)-9H-fluoren-9-ol (1e):



The reaction with 1-hexyne (0.27 mL, 2.31 mmol), *n*-butyllithium (1.05 M in hexane, 2.4 mL, 2.50 mmol) and 9-fluorenone (300 mg, 1.67 mmol) in THF (9 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/80) afforded **15e** (265.6 mg, 1.012 mmol, 61%).

Yellow oil; R_f value 0.45(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3398, 3064, 3042, 2956, 2931, 1450, 1009, 768, 732 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.69(dd, 2H, J = 7.5, 1.0 Hz), 7.61(dd, 2H, J = 7.5, 1.0 Hz), 7.39(ddd, 2H, J = 7.5, 7.5, 1.0 Hz), 7.35(ddd, 2H, J = 7.5, 7.5, 1.0 Hz), 2.44(s, 1H, OH), 2.21(t, 2H, J = 7.0 Hz), 1.48(m, 2H), 1.37(m, 2H), 0.88(t, 3H, J = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 147.6, 138.9, 129.4, 128.5, 124.1, 120.1, 84.3, 79.8, 74.9, 30.5, 21.9, 18.5, 13.6; LRMS (EI) 262(M⁺, 100%), 215(35); HRMS (EI) calcd for C₁₉H₁₈O (M⁺) 262.1358, found 262.1358.

3-Ethylnon-4-yn-3-ol (1f):

$$Et \xrightarrow{Et} nBu$$

HO 15f

The reaction with 1-hexyne (0.97 mL, 8.7 mmol), *n*-butyllithium (1.64 M in hexane, 5.3 mL, 8.7 mmol) and 3-pentanone (0.61 mL, 5.8 mmol) in THF (58 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/10) afforded **15f** (787 mg, 4.68 mmol, 81%).

Colorless oil; R_f value 0.49(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3389, 2965, 2935, 1460, 1143, 965 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 2.2(t, 2H, J = 7.0 Hz), 1.81(m, 1H, OH), 1.63(m, 4H), 1.48(tt, 2H, J = 7.5, 8.0 Hz), 1.40(tq, 2H, J = 7.5, 8.0 Hz), 1.01(t, 6H, J = 7.5 Hz), 0.90(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 84.9, 82.5, 72.2, 34.6, 30.9, 21.9, 18.3, 15.6, 8.6; HRMS (CI) calcd for C₁₁H₂₀O [M+H]⁺ 169.1592, found 169.1592.

1-Phenylhept-2-yn-1-ol (15g):

Ph HO HO 15g

The reaction with 1-hexyne (0.31 mL, 2.59 mmol), *n*-butyllithium (1.64 M in hexane, 1.7 mL, 2.8 mmol) and benzaldehyde (0.19 mL, 1.9 mmol) in THF (18 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/20 to 1/10) afforded **15g** (271 mg, 1.44 mmol, 77%).

Colorless oil; R_f value 0.48(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3332, 2957, 1454, 1001, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.55(d, 2H, *J* = 6.5 Hz), 7.38(dd, 2H, *J* = 7.0, 6.5 Hz), 7.32(t, 1H, *J* = 7.0 Hz), 5.45(br-d, 1H, *J* = 4.0), 2.36(s, 1H, OH), 2.29(td, 2H, *J* = 7.0, 1.5 Hz), 1.54(tt, 2H, *J* = 7.0, 8.0 Hz), 1.44(tq, 2H, *J* = 7.0, 8.0 Hz), 0.93(t, 3H, *J* = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 141.2, 128.5, 128.1, 126.6, 87.6, 79.8, 64.7, 30.6, 21.9, 18.4, 13.5; LRMS (EI) 188(M⁺, 100%), 145(63), 105(50), 77(62); HRMS (EI) calcd for C₁₃H₁₆O (M⁺) 188.1201, found 188.1202.

Methyl 2-hydroxy-2-phenyloct-3-ynoate (15h)



To a stirred solution of 1-hexyne (0.487 mL, 4.26 mmol) in THF (31 mL) at 0 °C under nitrogen atmosphere was added lithium bis(trimethylsilyl)amide (1.0 M in THF, 4.57 mL, 4.57 mmol) dropwise. After 30 min, methyl benzoylformate (500.0 mg, 3.05 mmol) was then added at the same temperature and the mixture was warmed up to room temperature. After 12 h, reaction mixture was quenched with saturated ammonium chloride aqueous solution. The mixture was diluted with ether and washed with water and brine. Then the collected organic layer was dried over magnesium sulfate and concentration *in vacuo* followed by silica gel column chromatography (ethyl acetate/hexane = 1/20 to 1/15) to give **15h** (344.8 mg, 46%).

Colorless oil; R_f value 0.29 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3496, 3029, 1736, 1256, 1063, 698 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.68 (d, 2H, *J* = 6.5 Hz), 7.39–7.32 (m, 3H), 4.18 (s, 1H, OH), 3.76 (s, 3H), 2.33 (t, 2H, *J* = 7.5 Hz), 1.58 (tt, 2H, *J* = 7.5, 8.0 Hz), 1.46 (tq, 2H, *J* = 8.0, 7.5 Hz), 0.93 (t, 3H, *J* = 7.5 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 172.7, 139.6, 128.5, 128.2, 126.2, 87.6, 78.1 72.8, 54.0, 30.3, 21.9, 18.4, 13.5; HRMS (CI) calcd for C₁₅H₁₉O₃ [M+H]⁺ 247.1334, found 247.1335.

3-Cyclohexyl-1,1-diphenylprop-2-yn-1-ol (15i)



The reaction with cyclohexylacetylene (0.2 mL, 1.531 mmol), LHMDS (1.0 M in THF, 1.9 mL, 1.9 mmol) and benzophenone (232.4 mg, 1.276 mmol) in THF(7 mL) afforded **15i** (368.4 mg, 1.269 mmol, 99%) by silica gel column purification (ethyl acetate/hexane = 1/80 to 1/50).

Light yellow oil; R_f value 0.31(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 3455, 3060, 2929, 2853, 2231, 1490, 1448, 1031, 991, 765, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.71(m, 4H), 7.38(dd, 4H, J = 7.5, 7.5 Hz), 7.31(t, 2H, J = 7.5 Hz), 2.98(m, 1H, OH), 2.61(m, 1H), 1.94(m, 2H), 1.82(m, 2H), 1.59–1.65(m, 3H), 1.38–1.45(m, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 145.5, 128.0, 127.3, 125.8, 92.0, 83.0, 74.3, 32.4, 29.0, 25.8, 24.7; LRMS (EI) 290(M⁺, 15%), 208(80), 207(100); HRMS (EI) calcd for C₂₁H₂₂O (M⁺) 290.1671, found 290.1670.

1,1,3-Triphenyl-2-propyn-1-ol (15j)



The reaction with phenylacetylene (0.6 mL, 5.488 mmol), *n*-butyllithium (1.57 M in hexane 3.5 mL, 5.488 mmol) and benzophenone (500 mg, 2.744 mmol) in THF (14 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/20) afforded **15j** (763.5 mg, 2.685 mmol, 98%).

Light yellow oil; R_f value 0.54 (ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3545, 3435, 3060, 3029, 2224, 1598, 1489, 1448, 1335, 1164, 1045, 984, 916, 890, 755, 697 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.69(m, 4H), 7.52(m, 2H), 7.34–7.37(m, 7H), 7.29(m, 2H), 2.88(s, 1H); ¹³C NMR (126 MHz, CDCl₃) δ 144.9, 131.8, 128.7, 128.3, 127.7, 128.3, 127.7, 126.0, 122.3, 91.8, 87.2, 74.8; LRMS (EI) 284(M⁺, 83%), 267(12), 207(30), 206(55), 153(77), 136(54), 107(79), 89(60), 78(58), 77(100); HRMS (EI) calcd for C₂₁H₁₆O (M⁺) 284.1201, found 284.1201.

4-(*Tert*-butyldimethylsilyl)oxy-1,1-diphenylbut-2-yn-1-ol (15m)



The reaction with 1-(tert-butyldimethylsilyl)oxyprop-2-yne¹ (291.4 mg, 1.711 mmol), *n*-butyllithium (1.57 M in hexane, 1.5 mL, 0.2.355 mmol) and benzophenone (222.7 mg, 1.222 mmol) in THF (12 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/100 to 1/80 to 1/50) afforded **15m** (185.7 mg, 0.5267 mmol, 43%).

1) Karjalainen, O. K.; Koskinen, A. M. P. *Org. Biomol. Chem.* **2011**, *9*, 1231–1236. White solid ; R_f value 0.30(ethyl acetate/hexane = 1/10); m.p. 83.0–84.5 °C; IR (NaCl, neat) v_{max} 3387, 3060, 2930, 2857, 1449, 1352, 1254, 1044 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.61(d, 4H, *J* = 8.0 Hz), 7.33(dd, 4H, *J* = 8.0, 8.0 Hz), 7.27(t, 2H, *J* = 8.0 Hz), 4.48(s, 2H), 2.90(s, 1H), 0.93(s, 9H), 0.13(s, 3H), 0.12(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 144.7, 128.1, 127.6, 126.0, 87.1, 86.1, 74.3, 51.8, 25.7, 18.2, -5.2; HRMS (ESI) calcd for C₂₂H₂₈O₂SiNa [M+Na]⁺ 375.1756, found 375.1751.

6-Benzoyloxy-1,1-diphenylhex-2-yn-1-ol (150)



The reaction with 4-pentynyl benzoate¹ (568.1 mg, 0.3018 mmol), LHMDS (1.0 M in THF, 3.3 mL, 3.29 mmol) and benzophenone (500 mg, 2.744 mmol) in THF (12 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/20 to 1/10 to 1/6)

afforded **15o** (974.2 mg, 2.630 mmol, 84%).

1) Stevens, B. D.; Nelson, S. G. *J. Org. Chem.* **2005**, *70*, 4375–4379. Light white crystal; R_f value 0.33(ethyl acetate/hexane = 1/4); m.p. 65.5–67.0 °C; IR (NaCl, neat) v_{max} 3462, 3060, 2959, 2235, 1717, 1450, 1276, 1119, 752, 701 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 8.04(dd, 2H, *J* = 8.0, 1.0 Hz), 7.55–7.60(m, 5H), 7.44(dd, 2H, *J* = 8.0, 8.0 Hz), 7.31(m, 4H), 7.24(m, 2H), 4.46(t, 2H, *J* = 7.0 Hz), 2.82(m, 1H, OH), 2.55(t, 2H, *J* = 7.0 Hz), 2.07(tt, 2H, *J* = 7.0, 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 166.5, 145.2, 133.0, 130.1, 129.6, 128.4, 128.2, 127.6, 125.9, 86.5, 83.9, 74.4, 63.5, 27.7, 15.8; LRMS (EI) 370(M⁺, 0.3%), 293(20), 105(100), 77(51); HRMS (EI) calcd for C₂₅H₂₂O₃ (M⁺) 370.1569, found 370.1573.

Methyl 4-hydroxy-4-diphenylbut-2-ynoate (15p)

$$Ph$$

 $Ph \longrightarrow CO_2Me$
 HO **15p**

The reaction with methyl propiolate (0.64 mL, 7.7 mmol), LHMDS (1.0 M in THF, 8.2 mL, 8.2 mmol) and benzophenone (1.0 g, 5.5 mmol) in THF (33 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/15 to 1/10) afforded **15p** (201 mg, 0.755 mmol, 14%).

Colorless oil; R_f value 0.26(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 3393, 2954, 1701, 1439, 1267, 750, 696 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.57(d, 4H, J = 7.0 Hz), 7.36(dd, 4H, J = 7.0, 8.0 Hz), 7.31(t, 2H, J = 8.0 Hz), 3.80(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 153.8, 143.0, 128.5, 128.3, 126.0, 88.8, 78.3, 74.3, 52.9; LRMS (EI) 266(M⁺, 12%), 234(68), 206(78), 178(85), 105(100); HRMS (EI) calcd for C₁₇H₁₄O₃ (M⁺) 266.0943, found 266.0944.

1,1-Diphenyldeca-2,9-diyn-1-ol (15q)



The reaction with 1,8-nonadiyne (0.6 mL, 4.116 mmol), LHMDS (1.0 M in THF, 4.1 mL, 4.116 mmol) and benzophenone (500 mg, 2.744 mmol) in THF (14 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/40 to 1/20 to 1/15) afforded **15q** (703.3 mg, 2.326 mmol, 85%).

Colorless oil; R_f value 0.19(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 3517, 3293, 2938, 2861, 2233, 2115, 1598, 1490, 1449, 1332, 1004, 767 cm⁻¹; ¹H NMR(500

MHz, CDCl₃) δ 7.60(d, 4H, *J* = 7.5 Hz), 7.32(dd, 4H, *J* = 7.5, 7.5 Hz), 7.25(t, 2H, *J* = 7.5 Hz), 2.72(m, 1H, OH), 2.36(t, 2H, *J* = 7.0 Hz), 2.20(m, 2H), 1.93(t, 1H, *J* = 2.5 Hz), 1.54(m, 6H); ¹³C NMR(126 MHz, CDCl₃) δ 145.4, 128.1, 127.5, 125.9, 87.9, 84.4, 83.2, 74.4, 68.4, 27.97, 27.94, 27.86, 18.8, 18.3; HRMS (ESI) calcd for C₂₂H₂₂NaO [M+Na]⁺ 325.1568, found 325.1570.

1,1,-Diphenyl-3-trimethylsilyl-2-propyn-1-ol (15r)

Ph $Ph \longrightarrow TMS$ HO **15r**

The reaction with trimethylsilylacetylene (0.76 mL, 5.488 mmol), *n*-butyllithium (1.57 M in hexane, 3.5 mL, 5.488 mmol) and benzophenone (500 mg, 2.744 mmol) in THF (14 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/30 to 1/20) afforded **15r** (721.4 mg, 2.572 mmol, 94%).

Colorless oil; R_f value 0.27(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 3548, 3454, 3060, 3029, 2959, 2170, 1599, 1491, 1449, 1251, 1057, 845 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.61(m, 4H), 7.33(m, 4H), 7.27(m, 2H), 2.77(s, 1H), 0.24(s, 9H); ¹³C NMR(126 MHz, CDCl₃) δ 144.7, 128.2, 127.6, 125.9, 107.6, 92.0, 74.6, -0.2, ,; LRMS (EI) 280(8%, M⁺), 264(13), 263(15), 203(27), 191(22), 190(100), 189(46); HRMS (EI) calcd for C₁₈H₂₀OSi (M⁺) 280.1283, found 280.1281.

1,1,4,4,4-Pentaphenyl-2-butyn-1-ol (15s)



The reaction with 3-triphenylprop-1-yne^a (1.10 g, 4.116 mmol), *n*-butyllithium (1.57 M, 2.80 mL, 4.390 mmol) and benzophenone (500 mg, 2.744 mmol) in THF (14 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/100 to 1/20 to 1/10) afforded **15s** (1.24 g, 0.266 mmol, 97%).

a) Karlen, S.D.; Ortiz, R.; Chapman, O. L.; Garcia-Garibay, M. A. J. Am. Chem. Soc. **2005**, *127*, 6554–6555.

White crystal; R_f value 0.25(ethyl acetate/hexane = 1/10); m.p. 128.5–130.0 °C; IR (NaCl, neat) v_{max} 3541, 3437, 3084, 3060, 3027, 1598, 1490, 1447, 1326, 1216, 1033, 758, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.76(m, 4H), 7.35–7.46(m, 21H), 3.10(br, 1H); ¹³C NMR(126 MHz, CDCl₃) δ 145.1, 144.9, 129.1, 128.1, 128.0, 127.5, 126.8, 126.0, 93.3, 87.8, 74.7, 55.7; LRMS (EI) 450(M⁺, 18%), 268(44), 267(43), 183(49),

105(100); HRMS (EI) calcd for C₃₄H₂₆O (M⁺) 450.1984, found 450.1983.

Preparation of terminal alkyne 15k (1,1-diphenylprop-2-yn-1-ol)



To the solution of benzophenone (1.0 g, 5.5 mmol) in THF (27 mL) was added ethynylmagnesium bromide (0.5 M in THF, 26 mL, 13.2 mmol) at -78 °C under nitrogen atmosphere, and the mixture was warmed up to room temperature. After 20 h, the mixture was diluted with ethyl acetate and was treated with saturated ammonium chloride aqueous solution and brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography (ethyl acetate/hexane = 1/25 to 1/20 to 1/15) afforded propargyl alcohols **15** (1.06 g, 93%) as a light yellow oil, which was slowly crystallized to light yellow solid.

Light yellow crystal; R_f value 0.52(ethyl acetate/hexane = 1/4); m.p. 48.5–49.0 °C; IR (NaCl, neat) v_{max} 3543, 3438, 3286, 3060, 2115, 1956, 1599, 1490, 1449, 1335, 1167 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.62(m, 4H), 7.35(m, 4H), 7.28(m, 2H), 2.89(s, 1H, OH), 2.80(s, 1H); ¹³C NMR(126 MHz, CDCl₃) δ 144.3, 128.3, 127.9, 125.9, 86.3, 75.5, 74.3; LRMS (EI) 208(M⁺, 100%), 207(48), 191(17), 131(60), 130(29); HRMS (EI) calcd for C₁₅H₁₂O (M⁺) 208.0888, found 208.0889.

Preparation of phenyl ethers 15l (1,1-diphenyl-4-phenoxybut-2-yn-1-ol)



To the solution of 1-phenoxyprop-2-yne^b (652.7 mg, 4.94 mmol) in benzene (10 mL) was added ethylmagnesium bromide (0.95M in THF, 6.7 mL, 6.37 mmol) dropwise at room temperature under nitrogen atmosphere. After 20 min, benzophenone (500 mg, 2.74 mmol) dissolved in 2 mL of THF was added dropwise to the mixture. After 6 h, the mixture was treated with saturated ammonium chloride aqueous solution and brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography (ethyl acetate / hexane = 1/30 to 1/25 to 1/10 to 1/8) afforded phenyl propargyl ether **15l** (814.5.4 mg, 94%) as viscous light-yellow oil. b) Pastine, S. J.; Youn, S. W.; Sames, D. *Org. Lett.* **2003**, *5*, 1055–1058.

Light yellow oil; R_f value 0.45(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3433, 3060, 3029, 1597, 1492, 1449, 1214, 1032, 754, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃)

δ 7.53(d, 4H, *J* = 8.0 Hz), 7.24–7.34(m, 8H), 7.02–7.03(m, 3H), 4.85(s, 2H), 2.78(s, 1H, OH); ¹³C NMR(126 MHz, CDCl₃) δ 157.5, 144.4, 129.5, 128.2, 127.8, 125.9, 121.6, 115.2, 89.8, 82.3, 74.4, 56.2; LRMS (EI) 314(M⁺, 21%), 221(51), 203(49), 178(47), 143(50), 105(100); HRMS (EI) calcd for C₂₂H₁₈O₂ (M⁺) 314.1307, found 314.1307.

Preparation of acetate 15n (4-acetoxy-1,1-diphenylbut-2-yn-1-ol)



To the dichloromethane solution (5mL) of known 1,1-diphenyl-but-2-yne-1,4-diol (111.1 mg, 0.4663 mmol), prepared by the same procedure for 1k,¹ was added pyridine (188µL, 2.33 mmol) and acetic anhydride (220µL, 2.33 mmol) at room temperature. After 2 h, the mixture was treated with saturated ammonium chloride aqueous solution and brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography (ethyl acetate / hexane = 1/10 to 1/8) afforded acetate **15n** (127.7 mg, 98%) as white crystal.

1) Fiesselmann, H.; Sasse, K. Chem. Ber. 1956, 89, 1775–1791.

White solid; R_f value 0.62(ethyl acetate/hexane = 1/2); m.p. 79.0–80.5 °C; IR (NaCl, neat) v_{max} 3448, 3060, 3029, 1743, 1449, 1229, 1030, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.61(m, 4H), 7.35(m, 4H), 7.29(m, 2H), 4.81(s, 2H), 3.34(br, 1H, OH), 2.09(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 170.4, 144.4, 128.2, 127.7, 125.9, 89.1, 81.0, 74.2, 52.3, 20.6; LRMS (EI) 280(M⁺, 5%), 220(83), 192(100), 115(81); HRMS (EI) calcd for C₁₈H₁₆O₃ (M⁺) 280.1099, found 280.1097.

Preparation of dialkyne 20 (7-phenyltrideca-5,8-diyn-7-ol)



To the solution of 1-hexene (0.987 mL, 8.89 mmol) in THF (36 mL) was added *n*-Butyllithium (2.64 M in hexane, 3.37 mL, 8.89 mmol) at -78 °C under nitrogen atmosphere. After 30 min, benzoyl chloride (500 mg, 3.56 mmol) was added to the mixture and the mixture was warmed up to room temperature. After 3h, the mixture was diluted with ether, and was treated with saturated ammonium chloride aqueous solution

and brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography (Ethyl acetate/hexane= 1/20 to 1/15) afforded dipropargyl alcohol **20** (870 mg, 91%) as a colorless oil.

Colorless oil; R_f value 0.64(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3520, 2871, 1449, 995, 696 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.80(d, 2H, *J* = 7.5 Hz), 7.38(dd, 2H, *J* = 7.5, 7.5 Hz), 7.32(t, 1H, *J* = 7.5 Hz), 2.73(s, 1H, OH), 2.29(t, 4H, *J* = 7.0 Hz), 1.54(tt, 4H, *J* = 7.0, 7.5 Hz), 1.43(tq, 4H, *J* = 7.5, 7.5 Hz), 0.92(t, 6H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 142.9, 128.3, 128.2, 125.7, 85.8, 81.0, 65.2, 30.4, 21.9, 18.5, 13.6; HRMS (ESI) calcd for C₁₉H₂₄NaO [M+Na]⁺ 291.1725, found 291.1715.

Preparation of tert-butyl (3-azidopropyl)carbamate (16yA)

HO NH ₂	2) TsCl, Et ₃ N, DMAP, CH_2Cl_2	
3-aminopropan-1-ol	6) NaN3, DNN , 66 G	16 y A

To a solution of 3-aminopropan-1-ol (300 mg, 0.152 mmol) in DCM (4ml) under an atmosphere of nitrogen was added triethylamine (0.337 mL, 2.4 mmol) at ambient temperature and the resulting solution was stirred for 5 min. Then Boc₂O (0.459 mL, 2.0 mmol) was added at the same temperature. After 10 h, the reaction was quenched with water. The reaction mixture was washed with water and brine. Drying collected organic layer over magnesium sulfate followed by concentration, the crude product was used to the next step without further purification.

To a solution of *N*-Boc-3-aminopropan-1-ol (350 mg) in dichloromethane (3 mL) under nitrogen atmosphere was added triethylamine (0.336 ml, 0.239 mmol) and 4-toluenesulfonyl chloride (0.380 mg, 1.99 mmol) at ambient temperature. After 2.5 h, the reaction was quenched by water. The reaction mixture was washed with water and brine. Drying collected organic layer over magnesium sulfate followed by concentration, the crude product was used to the next step without further purification.

To a solution of crude material (430 mg) in DMF (6.5 mL) under nitrogen atmosphere was added sodium azide (93.4 mg, 1.44 mmol) at ambient temperature. Then the reaction mixture was heated at 50 °C for 2 h. The reaction was quenched with water and was extracted with ether. Drying collected organic layer over magnesium sulfate followed by concentration and silica gel chromatography (ethyl acetate/hexane = 1/10) gave *tert*-butyl (3-azidopropyl)carbamate **16yA** (244.7 mg, 94% for 3 steps) as a colorless oil.

Colorless oil; R_f value 0.11(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3351, 2977, 2097, 1692, 1520, 1251, 1171 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 4.64(br, 1H, NH),

3.36(t, 2H, J = 7.0 Hz), 3.21(td, 2H, J = 6.5, 6.0 Hz), 1.77(tt, 2H, J = 6.5, 7.0 Hz), 1.44(s, 9H); ¹³C NMR(126 MHz, CDCl₃) δ 155.9, 79.4, 49.1, 38.0, 29.2, 28.4; HRMS (CI) calcd for C₈H₁₇N₄O₂ [M+H]⁺ 201.1352, found 201.1349.

General Experimental Procedure of triazolations



To the mixture of propargyl alcohol (1 equiv), organic azide (1.5 equiv) in dichloromethane (0.1 M to alcohols) under nitrogen atmosphere, trimethylsilyl trifluoromethanesulfonate (1.2 equiv) was added at -90 °C dropwise. After five minutes, the reaction was quenched with saturated sodium bicarbonate aqueous solution, and was washed with brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column chromatography afforded triazole.

In the case of three component coupling reactions, for dialkyne **20**, cinnamylazide was added at -90 °C, and then the mixture was treated as shown above after 5 min.

For coupling products **7a–d**, the reactions were performed at -60 °C for 5 min. Then, nucleophiles (3 equiv) were added at the same temperature and the mixture was wormed up to 0 °C. After 30 min, the mixture was treated as same as above to obtain three-component coupling products.

c) For cinnamylazide, see: Lal, S.; McNally, J.; White, A. J. P.; Díez-González, S. *Organometallics*, **2011**, *30*, 6225–6232.

d) For phenylazide, see: Cheng, H.; Wan, J.; Lin, M.-I; Liu, Y.; Lu, X.; Liu, J.;Xu, Y.; Chen, J.; Tu, Z.; Cheng, Y.-S. E.; Ding, K. *J. Med. Chem.* **2012**, *55*, 2144–2153.

e) For 2,6-diisopropylphenylazide, see: Pilyugina, T. S.; Schrock, R. R.; Hock, A. S.; Müller, P. *Organometallics*, **2005**, *24*, 1929–1937. Since this azide was gradually decomposed, it was used soon after purifications.

(1-Benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)diphenylmethanol (2a):



The reaction with 15a (67.0 mg, 0.253 mmol), benzylazide (50.6 mg, 0.380 mmol) and

TMSOTf (55 μ L, 0.304 mmol) in dichloromethane (2.5 mL) followed by silica gel column chromatography (ethyl acetate/hexane = 1/5) afforded **16a** (99.7 mg, 0.251 mmol, 99%).

Colorless oil; R_f value 0.19(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3419, 3060, 2957, 2870, 1494, 1448, 1013, 760 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.31–7.34(m, 3H), 7.24–7.30(m, 10H), 7.13(d, 2H, *J* = 7.5 Hz), 5.45(s, 2H), 4.29(s, 1H, OH), 1.92(m, 2H), 0.86(qt, 2H, *J* = 8.0, 7.5 Hz), 0.71(m, 2H), 0.58(t, 3H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 148.9, 145.4, 135.1, 134.9, 128.9, 128.3, 127.9, 127.5, 127.0, 76.7, 51.9, 30.3, 22.7, 22.6, 13.3; LRMS (EI) 397(M⁺, 78%), 320(41), 105(79), 91(100); HRMS (EI) calcd for C₂₆H₂₇N₃O (M⁺) 397.2154, found 397.2155.

(**1-Benzyl-5-butyl-1***H***-1,2,3-triazol-4-yl**)**bis**(**4-chlorophenyl**)**methanol** (**16b**) (CCDC 950501):



The reaction with **15b** (92.0 mg, 0.276 mmol), benzylazide (55.1 mg, 0.414 mmol), TMSOTf (60 μ L, 0.331 mmol) in dichloromethane (2.8 mL) followed by silica gel column chromatography purification (ethyl acetate / hexane = 1/20 to 1/13 to 1/6)afforded **2b** (113.0 mg, 0.242 mmol, 88%).

White solid; R_f value 0.24(ethyl acetate/hexane = 1/4); m.p. 142.3–143.3 °C; IR (NaCl, neat) v_{max} 3347, 2958, 2931, 2871, 1490, 1092, 1013, 911, 826, 731 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.33–7.37(m, 3H), 7.27(d, 4H, *J* = 8.0 Hz), 7.20(d, 4H, *J* = 8.0 Hz), 7.15(d, 2H, *J* = 7.5 Hz), 5.46(s, 2H), 4.33(m, 1H), 2.01(m, 2H), 0.94(tq, 2H, *J* = 7.5, 7.5 Hz), 0.76(m, 2H), 0.64(t, 3H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, C₆D₆) δ 148.6, 144.8, 135.8, 135.0, 133.8, 129.8, 129.1, 128.5, 128.3, 127.3, 77.4, 51.8, 30.7, 23.0, 22.9, 13.6; HRMS (FAB) calcd for C₂₆H₂₄Cl₂N₃O [M–H]⁺ 465.1375, found 465.1374.

(1-Benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)bis(4-methoxyphenyl)methanol (2c):



The reaction with **15c** (85.1 mg, 0.262 mmol), benzylazide (52.4 mg, 0.393 mmol), TMSOTf (57 μ L, 0.315 mmol) in dichloromethane (2.6 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/20 to 1/4 to 1/3) afforded **16c** (24.7 mg, 0.206 mmol, 21%).

Yellow oil; R_f value 0.27(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3481, 2956, 1509, 1250, 1173, 1034, 830 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.30–7.36(m, 3H), 7.15–7.18(m, 6H), 6.81(dt, 4H, J = 9.0, 3.0 Hz), 5.46(s, 2H), 4.19(s, 1H, OH), 3.78(s, 6H), 1.96(m, 2H), 0.91(qt, 2H, J = 7.5, 7.5 Hz), 0.74(m, 2H), 0.61(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 158.8, 149.3, 137.9, 135.2, 134.7, 129.1, 128.9, 128.2, 127.0, 113.1, 77.1, 55.2, 51.9, 30.4, 22.7(2C), 13.4; HRMS (ESI) calcd for C₂₈H₃₁N₃O₃Na [M+Na]⁺ 480.2263, found 480.2261.

1-Benzyl-5-butyl-4-(1-phenylvinyl)-1*H*-1,2,3-triazole (16d'):



The reaction with **15d** (64.2 mg, 0.317 mmol), benzylazide (63.4 mg, 0.476 mmol), TMSOTf (69 μ L, 0.381 mmol) in dichloromethane (3.2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/5) afforded **16d'** (68.4 mg, 0. 215 mmol, 68%).

Colorless oil; R_f value 0.35(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2957, 1495, 1455, 1240, 1028, 702 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.26–7.33(m, 8H), 7.19 (d, 2H, J = 6.5 Hz), 5.63(dd, 2H, J = 21.0, 2.0 Hz), 5.52(s, 2H), 2.24(t, 2H, J = 8.0 Hz), 1.10(m, 2H), 0.99(tq, 2H, J = 7.0, 7.0 Hz), 0.64(t, 3H, J = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.0, 139.8, 139.8, 135.2, 134.8, 128.9, 128.2, 128.2, 127.9, 127.3, 127.0, 116.9, 51.9, 30.3, 22.5, 22.2, 13.3; HRMS (ESI) calcd for C₂₁H₂₃N₃Na [M+Na]⁺ 340.1790, found 340.1789.

1-benzyl-5-butyl-1*H*-1,2,3-triazole-4-carbaldehyde (16g'):



The reaction with **16g** (100.8 mg, 0.535 mmol), benzylazide (106.9 mg, 0.803 mmol), TMSOTf (116 μ L, 0.642 mmol) in dichloromethane (5.4 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/20 to 1/10) afforded **16g'** (41.9 mg, 0. 172 mmol, 32%).

Colorless oil; R_f value 0.24(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 2958, 1696, 1472, 835, 717 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 10.2(s, 1H,CHO), 7.36–7.34(m, 3H), 7.20(m, 2H), 5.54(s, 2H), 2.85(t, 2H, *J* = 7.5 Hz), 1.32–1.27(m, 4H), 0.82(t, 3H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 186.1, 144.1, 141.5, 134.1, 129.1, 128.7, 127.2, 51.8, 29.9, 22.9, 22.4, 13.5; HRMS (ESI) calcd for C₁₄H₁₇N₃NaO [M+Na]⁺ 266.1269, found 266.1269.

Methyl 2-azido-2-(1-benzyl-5-butyl-1H-1,2,3-triazol-4-yl)-2-phenylacetate (16h', CCDC 950507)



To a mixture of propargyl alcohol **16h** (68.3 mg, 0.277 mmol) and benzyl azide (55.4 mg, 0.416 mmol) in dichloromethane (2.8 mL) under nitrogen atmosphere, TMSOTf (110.2 μ L, 0.610 mmol) was added at room temperature dropwise. After five minutes, the reaction was quenched with saturated sodium bicarbonate aqueous solution, and was washed with brine. Then the collected organic layer was dried over magnesium sulfate followed by concentration *in vacuo* and silica gel column chromatography (ethyl acetate / hexane = 1 / 5) to afford **16h'** (34.3 mg, 31%)

Colorless solid.; R_f value 0.18 (ethyl acetate/hexane = 1/5); m.p. 158.7–159.4 °C; IR (NaCl, neat) v_{max} 2957, 2112, 1746, 1241 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.44–7.42 (m, 2H), 7.35–7.30 (m, 6H), 7.11 (d, 2H, J = 7.0 Hz), 5.54 (d, 1H, J = 16.0 Hz), 5.47 (d, 1H, J = 16.0 Hz), 3.89 (s, 3H), 2.19 (m, 1H), 2.02 (m, 1H), 1.06–0.84 (m, 4H), 0.62 (t, 3H, J = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 169.8, 143.3, 136.6, 136.2, 134.7, 129.0, 128.5, 128.3, 128.1, 127.4, 126.8, 71.1, 53.6, 52.1, 29.7, 22.5, 22.4, 13.2; HRMS (ESI) calcd for C₂₂H₂₄N₆NaO₂ [M+Na]⁺ 427.1858, found 427.1856.

(**1-Benzyl-5-cyclohexyl-1***H***-1,2,3-triazol-4-yl**)**diphenylmethanol** (**16i**) (CCDC 950502)



The reaction with **15i** (55.0 mg, 0.189 mmol), benzylazide (35.8 mg, 0.269 mmol), TMSOTf (39 µL, 0.215 mmol) in dichloromethane (2 mL) afforded **16i** (76.6 mg, 0.181 mmol, 96%) by silica gel column purification (ethyl acetate/hexane = 1/6 to 1/5 to 1/4). White solid; R_f value 0.18(ethyl acetate/hexane = 1/4); m.p. 112.9–114.0 °C; IR (NaCl, neat) v_{max} 3446, 2928, 1448, 758, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.28–7.35(m, 13H), 7.07(d, 2H, *J* = 7.5 Hz), 5.62(s, 2H), 4.51(s, 1H, OH), 2.25(tt, 1H, *J* = 12.5, 2.5 Hz), 1.48–1.51(m, 3H), 1.25(dddd, 2H, *J* = 12.5, 12.5, 12.5, 2.5 Hz), 1.16(m, 2H), 0.95(dtt, 1H, *J* = 13.0, 13.0, 3.5 Hz), 0.72(ddddd, 2H, *J* = 13.0, 13.0, 13.0, 3.5, 3.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 148.8, 145.4, 138.4, 136.0, 128.8, 127.90, 127.89, 127.5, 126.4, 77.9, 53.1, 34.0, 30.3, 26.5, 25.5; LRMS (EI) 423(43%, M⁺), 105(47), 91(100); HRMS (EI) calcd for C₂₈H₂₉N₃O (M⁺) 423.2311, found 423.2306.

(1-Benzyl-5-phenyl-1H-1,2,3-triazol-4-yl)diphenylmethanol (16j)



The reaction with **15j** (60.4 mg, 0.212 mmol), benzylazide (42.4 mg, 0.319 mmol) and TMSOTf (42 μ L, 0.234 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/25 to 1/12 to 1/10 to 1/5) afforded **16j** (19.5 mg, 0.0467 mmol, 22%).

Light yellow solid; R_f value 0.23 (ethyl acetate/hexane = 1/4); m.p. 99.6–101.0 °C; IR (NaCl, neat) v_{max} 3419, 3060, 3030, 1489, 1448, 1009, 757, 697 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.20–7.26(m, 4H), 7.12–7.16(m, 10H), 7.04(dd, 2H, J = 8.0, 8.0 Hz), 6.94(dd, 2H, J = 8.0, 2.0 Hz), 6.55(dd, 2H, J = 8.0, 2.0 Hz), 5.26(s, 2H), 4.15(s, 1H, OH); ¹³C NMR (126 MHz, CDCl₃) δ 149.7, 145.2, 135.1, 135.0, 129.9, 128.7, 128.6,

128.1, 128.0, 127.7, 127.6, 127.3, 126.9, 77.7, 52.1; HRMS (ESI) calcd for $C_{28}H_{23}N_3ONa \ [M+Na]^+ 440.1739$, found 440.1739.

3,4,5-Triphenyl-1*H*-benzo[*c*]azepine(15j')



White solid; R_f value 0.33(ethyl acetate/hexane = 1/4); m.p. 138.6–140.0 °C; IR (NaCl, neat) v_{max} 3057, 2957, 2835, 1489, 1444, 754, 697 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.50–7.53(m, 3H), 7.35(d, 1H, J = 8.0 Hz), 7.32(dd, 1H, J = 7.5, 7.5 Hz), 7.19–7.28(m, 9H), 6.93–7.03(m, 5H), 5.07(d, 1H, J = 10.0 Hz), 4.45(d, 1H, J = 10.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 167.4, 148.1, 140.6, 139.9, 139.4, 138.7, 138.20, 138.16, 130.6, 130.3, 129.5, 128.9, 128.7, 128.6, 127.9, 127.8, 127.5, 127.4, 127.1, 127.0, 126.6, 56.2; HRMS (ESI) calcd for C₂₈H₂₂N [M+H]⁺ 372.1752, found 372.1752.

(1-Benzyl-1H-1,2,3-triazol-4-yl)diphenylmethanol (16k) (CCDC 950503)



The reaction with **15k** (55.0 mg, 0.264 mmol), benzylazide (52.7 mg, 0.396 mmol) and TMSOTf (100 μ L, 0.555 mmol) in dichloromethane (3 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/6 to 1/2 to 1/1) afforded **16k** (80.8 mg, 0.237 mmol, 90%).

White crystal; R_f value 0.40(ethyl acetate/hexane = 1/2); m.p. 140.5–140.9 °C; IR (NaCl, neat) v_{max} 3393, 3056, 3030, 1490, 1446, 1173, 1016 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.40–7.56(m, 16H), 5.65(s, 2H), 4.04(s, 1H); ¹³C NMR(126 MHz, CDCl₃) δ 154.4, 145.6, 134.5 129.1, 128.7, 128.0, 127.8, 127.5, 127.1, 122.4, 76.7, 54.1; HRMS (ESI) calcd for C₂₂H₁₉N₃ONa [M+Na]⁺ 364.1426, found 364.1430.

(1-Benzyl-5-phenoxymethyl-1H-1,2,3-triazol-4-yl)diphenylmethanol (15l)



The reaction with **15l** (64.0 mg, 0.204 mmol), benzylazide (40.7 mg, 0.305 mmol) and TMSOTf (44 μ L, 0.244 mmol) in dichloromethane (2.0 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/10 to 1/9 to 1/8) afforded **16l** (83.1 mg, 0.186 mmol, 91%).

White solid; R_f value 0.20(ethyl acetate/hexane = 1/4) ; m.p. 122.0–123.2 °C; IR (NaCl, neat) v_{max} 3420, 3061, 3033, 2246, 1599, 1495, 1233, 1173, 1031, 909, 754, 731, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.15–7.27(m, 13H), 7.10–7.14(m, 4H), 6.90(t, 1H, J = 7.5 Hz), 6.50(d, 2H, J = 8.5 Hz), 5.58(s, 2H), 4.24(s, 2H), 4.01(br, 1H, OH); ¹³C NMR(126 MHz, CDCl₃) δ 157.3, 151.6, 144.8, 134.4, 129.6, 129.4, 128.9, 128.4, 128.0, 127.7, 127.6, 127.4, 121.6, 114.4, 77.8, 57.5, 52.8; HRMS (ESI) calcd for C₂₉H₂₅N₃NaO₂ [M+Na]⁺ 470.1846, found 470.1851.

(1-Benzyl-5-((*tert*-butyldimethylsilyl)oxy)methyl-1*H*-1,2,3-triazol-4-yl)diphenyl-me thanol (16m):



The reaction with **15m** (62.6 mg, 0.178 mmol), benzylazide (35.5 mg, 0.266 mmol), TMSOTf (39 μ L, 0.213 mmol) in dichloromethane (2 mL) followed by silica gel column purification (ethyl acetate/hexane = 1/30 to 1/20 to 1/8 to 1/4) afforded **16m**(65.0 mg, 0. 134 mmol, 75%).

Yellow oil; R_f value 0.24(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3408, 3060, 3032, 2953, 2928, 2856, 1448, 1066, 838, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.16–7.35(m, 13H), 7.15(d, 2H, *J* = 7.0 Hz), 5.63(s, 2H), 4.60(s, 1H, OH), 4.07(s, 2H), 0.78(s, 9H), -0.18(s, 6H); ¹³C NMR(126 MHz, CDCl₃) δ 150.9, 145.2, 134.9, 132.4, 128.9, 128.2, 127.9, 127.6, 127.5, 126.9, 77.6, 53.5, 52.5, 25.6, 18.0, -5.87, -5.89;; HRMS (ESI) calcd for C₂₉H₃₅N₃O₂SiNa [M+Na]⁺ 508.2396, found 508.2392.

(5-Acetoxymethyl-1-benzyl-1*H*-1,2,3-triazol-4-yl)diphenylmethanol (16n):

The reaction with **15n** (127.7 mg, 0.456 mmol), benzylazide (91.0 mg, 0.683 mmol), TMSOTf (99 μ L, 0.547 mmol) in dichloromethane (5 mL) afforded **16n** (41.3 mg, 0.100 mmol, 22%) by silica gel column purification (ethyl acetate/hexane = 1/20 to 1/10 to 1/6).

Yellow oil; R_f value 0.39(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3449, 3060, 3029, 1740, 1230, 1029, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.26–7.36(m, 13H), 7.15(d, 2H, J = 6.0 Hz), 5.60(s, 2H), 4.67(s, 2H), 4.40(s, 1H, OH), 1.70(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 170.4, 152.1, 145.1, 134.7, 129.0, 128.9, 128.4, 127.9, 127.6, 127.4, 127.0, 77.7, 53.6, 52.4, 20.2; HRMS (ESI) calcd for C₂₅H₂₃NaN₃O₃ [M+Na]⁺ 436.1637, found 436.1637.

[5-(3-Benzoyloxyprop-1-)yl-1-benzyl-1*H*-1,2,3-triazol-4-yl]diphenylmethanol (160):



The reaction with **150** (75.2 mg, 0.203 mmol), benzylazide (40.5 mg, 0.305 mmol) and TMSOTf (44 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/5 to 1/3 to 1/2) afforded **160** (95.0 mg, 0.189 mmol, 93%).

Light yellow oil; R_f value 0.43(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3477, 3061, 2954, 2247, 1716, 1450, 1276, 714 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.93(d, 2H, J = 7.0 Hz), 7.59(t, 1H, J = 7.5 Hz), 7.45(dd, 2H, J = 7.5, 7.0 Hz), 7.25–7.29(m, 13H), 7.10(dd, 2H, J = 7.5, 2.5 Hz), 5.48(s, 2H), 4.15(m, 1H, OH), 3.88(t, 2H, J = 6.5 Hz), 2.23(m, 2H), 1.23(m, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 166.2, 149.4, 145.3, 134.9, 133.8, 133.1, 129.9, 129.4, 129.0, 128.4, 128.3, 128.0, 127.72, 127.66, 126.9, 77.8, 63.9, 52.0, 27.5, 20.1; HRMS (ESI) calcd for C₃₂H₂₉N₃O₃ [M+Na]⁺ 526.2107, found 526.2107.

[1-benzyl-5-methoxycarbonyl-1*H*-1,2,3-triazol-4-yl]diphenylmethanol (16p):



The reaction with **15p** (32.9 mg, 0.124 mmol), benzylazide (24.7 mg, 0.185 mmol), TMSOTf (47 μ L, 0.26 mmol) in dichloromethane (1.3 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/15 to 1/10 to 1/5) afforded **16p** (28.2 mg, 0. 071 mmol, 57%).

Colorless oil; R_f value 0.32(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3413, 3056, 1702, 1449, 1265, 735, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.36–7.27(m, 13H), 7.18(d, 2H, *J* = 6.0 Hz), 5.86(s, 2H), 5.69(s, 1H, OH), 3.64(s, 3H); ¹³C NMR(126 MHz, CDCl₃) δ 159.7, 158.7, 144.7, 134.8, 128.9, 128.5, 127.8, 127.5, 127.34, 127.29, 124.8, 77.6, 55.0, 52.9; HRMS (ESI) calcd for C₂₄H₂₁N₃O₃ [M+Na]⁺ 422.1481, found 422.1481.

[1-Benzyl-5-(hept-6-yn-1-yl)-1H-1,2,3-triazol-4-yl]diphenylmethanol (16q):



The reaction with **15q** (71.9 mg, 0.238 mmol), benzylazide (47.5 mg, 0.357 mmol) and TMSOTf (52 μ L, 0.285 mmol) in dichloromethane (2.4 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/2) afforded **16q** (99.5 mg, 0.241 mmol, 93%).

Light yellow oil; R_f value 0.17(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3419, 3302, 2938, 1495, 1448, 1012, 907, 701 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.26–7.37(m, 13H), 7.16(d, 2H, J = 6.5 Hz), 5.47(s, 2H), 4.27(s, 1H, OH), 1.95–2.00(m, 4H), 1.93(t, 1H, J = 3.0 Hz), 1.16(tt, 2H, J = 7.5, 7.5 Hz), 0.94(tt, 2H, J = 7.5, 7.5 Hz), 0.71(m, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 148.9, 145.3, 135.1, 134.7, 129.0, 128.3, 127.9, 127.8, 127.6, 127.0, 84.2, 77.7, 68.3, 52.0, 28.5, 27.7, 27.6, 22.9, 18.0; HRMS (ESI) calcd for C₂₉H₂₉N₃NaO [M+Na]⁺ 458.2208, found 458.2208.

(**1-Benzyl-5-trimethylsilyl-1***H***-1,2,3-triazol-4-yl**)**diphenylmethanol**(**16r**) (CCDC 950504):



The reaction with **15r** (72.5 mg, 0.259 mmol), benzylazide (51.6 mg, 0.388 mmol) and TMSOTf (56 μ L, 0.310 mmol) in dichloromethane (2.7 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10) afforded **16r** (99.5 mg, 0.241 mmol, 93%).

White crystal; R_f value 0.23(ethyl acetate/hexane = 1/4); m.p. 155–156 °C; IR (NaCl, neat) v_{max} 3379, 3060, 3030, 2950, 2898, 1495, 1446, 1249, 847, 755, 699 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.19–7.28(m, 13H, 6.83(d, 2H, *J* = 7.0 Hz), 5.68(s, 2H), 2.96(s, 1H), 0.24(m, 9H); ¹³C NMR(126 MHz, CDCl₃) δ 160.8, 146.6, 136.9, 132.9, 128.7, 127.9, 127.7, 127.6, 127.4, 125.8, 79.1, 53.9, 1.1; HRMS (ESI) calcd for C₂₅H₂₇N₃OSiNa [M+Na]⁺ 436.1821, found 436.1822.

[4-Butyl-1-cinnamyl-1*H*-1,2,3-triazol-4-yl]diphenylmethanol (16z):



The reaction with **15a** (55.1 mg, 0.208 mmol), azide (49.8 mg, 0.313 mmol), TMSOTf (45 μ L, 0.25 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/5) afforded **16z** (78.2 mg, 0. 185 mmol, 89%).

Colorless oil; R_f value 0.18(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3481, 2957, 1448, 762 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.26(m, 15H), 6.43(d, 1H, *J* = 15.0 Hz), 6.29(dt, 1H, *J* = 15.0, 6.0 Hz), 4.99(d, 2H, *J* = 6.0 Hz), 4.29(s, 1H, OH), 2.06(t, 2H, *J* = 6.5 Hz), 0.94(m, 4H), 0.63(t, 3H, *J* = 5.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 148.7, 145.4, 135.5, 134.8, 133.7, 128.6, 128.3, 127.9, 127.5, 126.5, 122.7, 77.7, 50.3, 30.7, 22.7, 22.6, 13.4; HRMS (ESI) calcd for C₂₈H₂₉N₃NaO [M+Na]⁺ 446.2208, found 446.2208.

[1-Adamantyl-5-butyl-1*H***-1,2,3-triazol-4-yl]diphenylmethanol** (16w) (CCDC 950505):



The reaction with **15a** (70.8 mg, 0.268 mmol), azide (71.2 mg, 0.402 mmol), TMSOTf (58 μ L, 0.32 mmol) in dichloromethane (2.7 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10) afforded **16w** (106.7 mg, 0. 242 mmol, 90%).

Colorless solid; R_f value 0.12(ethyl acetate/hexane = 1/5); m.p. 154.2–158.6 °C; IR (NaCl, neat) v_{max} 3509, 2963, 1473, 1000, 761, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.26–7.31(m, 10H), 4.40(s, 1H, OH), 2.37(d, 6H, J = 3.0 Hz), 2.29(t, 2H, J = 6.5 Hz), 2.25(s, 3H), 1.77(t, 6H, J = 13.5 Hz), 0.92–0.97(m, 4H), 0.68(t, 3H, J = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 149.2, 145.6, 134.9, 128.0, 127.8, 127.4, 78.0, 62.8, 35.9, 31.7, 29.7, 24.5, 22.8, 13.3; HRMS (ESI) calcd for C₂₉H₃₅N₃NaO [M+Na]⁺ 464.2678, found 464.2673.

[5-Butyl-1-(2,6-diisopropylphenyl)-1H-1,2,3-triazol-4-yl]diphenylmethanol (16v) :



The reaction with **15a** (43.2 mg, 0.163 mmol), azide (49.9 mg, 0.245 mmol), TMSOTf (35 μ L, 0.20 mmol) in dichloromethane (1.6 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/20 to 1/10) afforded **16v** (64.5 mg, 0.138 mmol, 85%).

Colorless oil; R_f value 0.19(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3504, 2963, 1447, 1000, 761, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.49(t, 1H, J = 7.5 Hz), 7.28–7.36(m, 12H), 4.44(s, 1H, OH), 2.16(qq, 2H, J = 7.0, 6.5 Hz), 1.83(t, 2H, J = 8.0 Hz), 1.19(d, 6H, J = 7.0 Hz), 1.16(d, 6H, J = 6.0 Hz), 0.72(m, 4H), 0.48(t, 3H, J = 6.5
Hz); ¹³C NMR(126 MHz, CDCl₃) δ 148.0, 146.3, 145.4, 136.8, 131.6, 130.8, 127.94, 127.88, 127.6, 123.9, 77.8, 30.2, 28.7, 25.7, 23.1, 22.6, 22.3, 13.2; HRMS (ESI) calcd for C₃₁H₃₈N₃O [M+H]⁺ 468.3015, found 468.3026.

[1-(3-*tert*-butoxylcarbonylaminoprop-1-)yl-5-butyl-1*H*-1,2,3-triazol-4-yl]diphenylm ethanol (16y) :



The reaction with **15a** (53.5 mg, 0.202 mmol), azide (60.8 mg, 0.306 mmol), TMSOTf (91 μ L, 0.51 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/5 to 1/3) afforded**16y** (75.8 mg, 0.163 mmol, 81%).

Colorless oil; R_f value 0.35(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3338, 2960, 1692, 1168, 700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.27–7.31(m, 10H), 4.90(s, 1H, OH), 4.25(s, 1H,NH), 4.22(t, 2H, J = 7.5 Hz), 3.17–3.18(m, 2H), 2.03–2.10(m, 4H), 1.43(s, 9H), 0.91–1.02(m, 4H), 0.70(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 156.0, 148.5, 145.4, 134.7, 127.9, 127.8, 127.5, 79.4, 77.7, 45.1, 37.4, 30.8, 30.2, 28.3, 22.7, 13.5; HRMS (ESI) calcd for C₂₇H₃₆N₄NaO₃ [M+Na]⁺ 487.2685, found 487.2680.

1-Benzyl-4,4,9,9-tetraphenyl-4,9-dihydro-1*H***-naphtho**[**2,3-***d*][**1,2,3**]**triazole** (16s') (CCDC 950506):



The reaction with **15s** (84.3 mg, 0.187 mmol), benzylazide (37.4 mg, 0.281 mmol) and TMSOTf (41 μ L, 0.225 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10) and PTLC afforded **16s'** (50.5 mg, 0.0893 mmol, 48%).

White crystal; m.p. 206.0–207.2 °C; R_f value 0.16(ethyl acetate/hexane = 1/10); IR (NaCl, neat) v_{max} 3060, 3029, 1597, 1495, 1445, 749, 697 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.04–7.20(m, 14H), 6.90–6.98(m, 9H), 6.83(d, 2H, J = 7.5 Hz), 6.74(d, 4H, J

= 8.5 Hz), 4.78(s, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 150.8, 144.8, 144.4, 142.4, 141.5, 137.2, 134.5, 131.4, 130.3, 129.8, 129.5, 128.0, 127.9, 127.7, 127.6, 127.4, 126.9, 126.5, 126.3, 126.1, 54.91, 54.87, 52.2; HRMS (ESI) calcd for C₄₁H₃₂N₃ [M+H]⁺ 566.2596, found 566.2596.

1-(3-Hydroxy-3,3-diphenylprop-1-yn-1-yl)cyclopentanecarbaldehyde(15t')



Yellow oil; R_f value 0.37(ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 3424, 2941, 1727, 1449, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 9.55(s, 1H), 7.60(dd, 4H, *J* = 8.0, 1.0 Hz), 7.33(dd, 4H, *J* = 8.0, 8.0 Hz), 7.26(tt, 2H, *J* = 8.0, 1.0 Hz), 2.89(s, 1H), 2.18(m, 2H), 1.96(m, 2H), 1.80(m, 2H), 1.69(m, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 197.4, 145.0, 128.2, 127.6, 125.8, 87.7, 87.5, 74.4, 53.4, 35.2, 25.1; LRMS (EI) 304(M⁺, 10%), 303(11), 287(37), 275(62), 207(56), 105(100); HRMS (EI) calcd for C₂₁H₂₀O₂ (M⁺) 304.1463, found 304.1464.

1,1-Diphenyl-1-methoxyhept-2-yne(17)

To a stirred solution of sodium hydride (39.4 mg, 0.983 mmol) in THF (3 mL) at 0 °C under nitrogen atmosphere was added **15a** (0.200 g, 0.757 mmol) in THF (1 mL) dropwise and stirred for 30 min at the same temperature. After 30 min, iodomethane (0.141 mL, 2.27 mmol) was then added at same temperature and the mixture was warmed up to room temperature. After 2h, reaction mixture was quenched with water. The mixture was diluted with ethyl acetate and washed with water and brine. Then the collected organic layer was dried over magnesium sulfate and concentration *in vacuo* followed by silica gel column chromatography (ethyl acetate/hexane = 1/5) to give **17** (187.7 mg, 46%).

Colorless oil; R_f value 0.76 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2933, 1488, 1449, 1082, 763, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.56–7.54 (m, 4H), 7.31–7.27 (m, 4H), 7.24–7.20 (m, 2H), 3.33 (s, 3H), 2.38 (t, 2H, J = 7.0 Hz), 1.60 (tt, 2H, J = 7.5, 7.0 Hz), 1.47 (tq, 2H, J = 7.0, 7.5 Hz), 0.94 (t, 3H, J = 7.5 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 143.9, 128.0, 127.3, 126.6, 90.4, 80.9, 79.4, 52.2, 30.8, 22.0, 18.6,

13.6; LRMS (EI) 278 (M⁺,27%), 247 (60), 221 (57), 201 (100); HRMS (EI) calcd for $C_{20}H_{22}O$ (M⁺) 278.1670, found 278.1668.

1-Benzyl-5-butyl-4-(methoxydiphenylmethyl)-1H-1,2,3-triazole(18)



To a mixture of propargyl ether **17** (34.7 mg, 0.125 mmol) and benzyl azide (24.9 mg, 0.187 mmol) in dichloromethane (1.5 mL) under nitrogen atmosphere, TMSOTf (27 μ L, 0.150 mmol) was added at room temperature dropwise. After five minutes, the reaction was quenched with saturated sodium bicarbonate aqueous solution, and was washed with brine. Then the collected organic layer was dried over magnesium sulfate followed by concentration *in vacuo* and silica gel column chromatography (ethyl acetate/hexane = 1/5) to afford **18** (49.8 mg, 97%) as a colorless oil.; R_f value 0.19 (ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 2927, 2855, 1456, 1070, 744, 703 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.43 (d, 4H, *J* = 7.0 Hz),7.19–7.10 (m, 7H), 7.03 (tt, 2H, *J* = 7.5, 7.0 Hz), 6.98 (d, 2H, *J* = 7.0 Hz), 5.33 (s, 2H), 2.91 (s, 3H), 1.96 (t, 2H, *J* = 8.0 Hz), 0.83–0.74 (m, 4H), 0.50 (t, 3H, *J* = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 146.1, 143.9, 136.9, 135.3, 128.9, 128.1, 127.8, 127.6, 126.9, 126.8, 82.6, 51.9, 51.8, 29.8, 22.8, 22.6, 13.4; LRMS (EI) LRMS (EI) 411(M⁺,7%), 381(35), 105(12), 91(100); HRMS (EI) calcd for C₂₇H₂₉N₃O (M⁺) 411.2311, found 411.2314.

1-benzyl-5-butyl-4-((nonyloxy)diphenylmethyl)-1H-1,2,3-triazole (19a)



The reaction of **19a** (85.9 mg, 82%) and **16a** (7.7 mg, 10%) from **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol) and BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/30 to 1/20 to 1/10).

Colorless oil ; R_f value 0.38 (ethyl acetate/hexane = 1/5); R_f value 0.38 (ethyl

acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2927, 2855, 1457, 1070, 704 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.59 (d, 4H, J = 7.0 Hz), 7.33–7.25 (m, 7H), 7.17 (dd, 2H, J = 6.0, 7.5 Hz), 7.11 (d, 2H, J = 6.0 Hz), 5.47 (s, 2H,) 3.09 (t, 2H, J = 7.0 Hz), 2.07 (t, 2H, J = 8.0 Hz), 1.54 (tt, 2H, J = 7.0, 8.0 Hz), 1.28–1.22 (m, 12H), 0.96–0.86 (m, 7H),0.65 (t, 3H, J = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 146.6, 144.4, 136.9, 135.4, 128.8, 128.1, 127.8, 127.5, 126.9, 126.7, 81.8, 63.7, 51.8, 31.9, 29.9, 29.52, 29.46, 29.2, 26.3, 22.9, 22.7, 22.6, 14.1, 13.5; HRMS (ESI) calcd for C₃₅H₄₅N₃ONa [M+Na]⁺ 546.34603, found 546.34558.

1-benzyl-4-(((6-bromohexyl)oxy)diphenylmethyl)-5-butyl-1*H*-1,2,3-triazole(19b)



The reaction of **19b** (81.5 mg, 73%) and **16a** (8.7 mg, 11%) from **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol) and BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/15 to 1/10).

Colorless oil ; R_f value 0.29 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2933, 2868, 1456, 1071, 896, 704 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.58 (d, 4H, *J* = 8.0 Hz), 7.34–7.25 (m, 7H), 7.17 (dd, 2H, *J* = 7.5, 6.5 Hz), 7.12 (d, 2H, *J* = 6.5 Hz), 5.47 (s, 2H), 3.35 (t, 2H, *J* = 7.0 Hz), 3.11 (t, 2H, *J* = 6.5 Hz), 2.04 (t, 2H, *J* = 8.0 Hz), 1.80 (tt, 2H, *J* = 7.0, 6.5 Hz), 1.55 (tt, 2H, *J* = 6.5, 7.0 Hz), 1.35 (m, 4H), 0.96–0.86 (m, 4H), 0.65 (t, 3H, *J* = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 146.4, 144.3, 136.9, 135.3, 128.8, 128.1, 127.8, 127.4, 126.9, 126.7, 81.8, 63.4, 51.8, 33.8, 32.7, 29.9, 29.7, 27.9, 25.5, 22.8, 22.7, 13.4; LRMS (EI) ; HRMS (ESI) calcd for C₃₂H₃₈BrN₃ONa [M+Na]⁺ 582.20959, found 582.20950.

1-benzyl-5-butyl-4-((2-methoxyethoxy)diphenylmethyl)-1*H*-1,2,3-triazole(19c)



The reaction of **19c** (105.7 mg, 76%) and **3a** (14.3 mg, 12%) from **4a** (80.5 mg, 0.196 mmol) and BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/60 to 1/50 to 1/40 to 1/30 to 1/20 to 1/10).

Colorless oil; R_f value 0.18 (ethyl acetate/hexane = 1/4); IR (NaCl, neat) v_{max} 2955, 2929, 1449, 1080, 705 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.60 (d, 4H, J = 8.0 Hz), 7.26–7.34 (m, 7H), 7.18 (dd, 2H, J = 7.5, 7.0 Hz), 7.14 (d, 2H, J = 7.5 Hz), 5.46 (s, 2H), 3.49 (t, 2H, J = 4.0 Hz), 3.30–3.31 (m, 5H), 2.13 (m, 2H), 0.89–1.02 (m, 4H), 0.67 (t, 3H, J = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 146.5, 143.8, 137.0, 135.3, 128.9, 128.1, 127.8, 127.7, 127.0, 126.8, 82.3, 72.0, 62.9, 58.7, 51.8, 30.0, 22.9, 22.7, 13.5; LRMS (EI) 455 (3%, M⁺), 381 (26), 91 (100); HRMS (EI) calcd for C₂₉H₃₃N₃O₂ (M⁺) 455.2573, found 455.2573.

1-Benzyl-5-butyl-4-(diphenyl(prop-2-yn-1-yloxy)methyl)-1H-1,2,3-triazole (19d):



The reaction with **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol), $BF_3 \cdot OEt_2$ (47% in ether, 65 µL, 0.24 mmol) in dichloromethane (2 mL) and additional propargyl alcohol (33.7 mg, 0.600 mmol) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/4) afforded **19d** (64.7 mg, 0.149 mmol, 72%) and **2a** (9.2 mg, 0.023 mmol, 12%).

Colorless oil; R_f value 0.34(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3438, 3293, 2957, 1490, 1449, 1058, 727 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.57(d, 4H, *J* = 7.0 Hz), 7.25–7.30(m, 7H), 7.19(t, 2H, *J* = 7.0 Hz), 7.13(d, 2H, *J* = 6.5 Hz), 5.44(s, 2H), 3.87(d, 2H, *J* = 3.0 Hz), 2.17(t, 1H, *J* = 3.0 Hz), 2.09(t, 2H, *J* = 8.0 Hz), 0.91–0.97(m, 4H), 0.64(t, 3H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 145.6, 143.2, 137.3, 135.1, 128.8, 128.2, 127.9, 127.10, 127.08, 83.0, 80.4, 73.1, 52.7, 51.8, 30.0, 22.9, 22.7, 13.4; HRMS (ESI) calcd for C₂₉H₂₉N₃ONa [M+Na]⁺ 458.22083, found 458.22086.

1-benzyl-5-butyl-4-((pent-4-yn-1-yloxy)diphenylmethyl)-1H-1,2,3-triazole(19e)



The reaction of **19e** (71.8 mg, 77%) and **16a** (8.1 mg, 10%) from **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/4).

Colorless oil ; R_f value 0.29 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3448, 3303, 2956, 2871, 1449, 1071, 705 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.59 (d, 4H, J = 8.0 Hz), 7.33–7.25 (m, 7H), 7.18–7.12 (m, 4H), 5.46 (s, 2H), 3.22 (t, 2H, J = 6.5 Hz), 2.81 (t, 2H, J = 7.5 Hz), 2.03 (t, 2H, J = 8.5 Hz), 1.78–1.73 (m, 3H), 0.95–0.86 (m, 4H), 0.63 (t, 3H, J = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 146.2, 144.2, 137.0, 135.3, 128.9, 128.1, 127.8, 127.4, 126.9, 126.7, 83.8, 81.8, 68.4, 62.0, 51.8, 29.9, 29.0, 22.8, 22.6, 15.5, 13.4; HRMS (ESI) calcd for C₃₁H₃₃N₃ONa [M+Na]⁺ 486.25213, found 486.25226.

1-Benzyl-5-butyl-4-((naphthalen-2-ylthio)diphenylmethyl)-1H-1,2,3-triazole (19f)



The reaction of **19f** (87.8 mg, 81%) and **16a** (5.2 mg, 7%) from **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/5).

Colorless oil; R_f value 0.28 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3054, 2975, 1495, 1455, 728, 698 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.76 (d, 1H, J = 8.0 Hz), 7.57 (d, 1H, J = 7.5 Hz), 7.54–7.53 (m, 2H), 7.48–7.43 (m, 6H), 7.32–7.22 (m, 10H), 7.04 (d, 2H, J = 7.5 Hz), 5.49 (s, 2H), 2.18 (t, 2H, J = 8.5 Hz), 0.97 (tq, 2H, J = 7.5, 8.0 Hz), 0.73 (m, 2H), 0.64 (t, 3H, J = 7.5 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 147.1,

142.9, 135.9, 135.1, 133.9, 133.0, 132.3, 131.5, 131.1, 129.4, 128.7, 128.0, 127.7, 127.6, 127.4, 127.2, 126.9, 126.7, 126.1, 125.9, 64.0, 52.0, 29.0, 23.3, 22.7, 13.3; HRMS (ESI) calcd for C₃₆H₃₃N₃SNa [M+Na]⁺ 562.22929, found 562.22952.

N-((1-benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)diphenylmethyl)-*N*-ethylethanamine (19g):



The reaction with **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) and additional diethylamine (63 μ L, 0.600 mmol) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/5) afforded **19g** (70.6 mg, 0.156 mmol, 78%) and **2a** (8.6 mg, 0.022 mmol, 11%).

Colorless oil; R_f value 0.23(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2960, 1594, 1456, 1025, 705 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.53(d, 4H, J = 8.0 Hz), 7.31–7.35(m, 3H), 7.25–7.29(m, 4H), 7.17(dd, 2H, J = 6.5, 7.0 Hz), 7.07(d, 2H, J = 7.0 Hz), 5.49(s, 2H), 2.45(q, 4H, J = 7.5 Hz), 2.30(t, 2H, J = 8.5 Hz), 1.07(m, 2H), 0.82(m, 2H), 0.74(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 146.6, 144.1, 135.6, 135.4, 129.1, 128.8, 128.0, 127.3, 126.6, 126.0, 74.4, 51.9, 47.2, 29.1, 23.6, 23.0, 16.1, 13.5; HRMS (ESI) calcd for C₃₀H₃₆N₄Na [M+Na]⁺ 475.2838, found 475.2828.

N-((1-Benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)diphenylmethyl)prop-2-en-1-amine (19h)



The reaction of **19h** (68.2 mg, 77%) and **16a** (9.4 mg, 12%) from **15a** (53.8 mg, 0.204 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/4).

Colorless oil; R_f value 0.29 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3320, 2957, 2870, 1491, 1456, 728, 704 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.45 (d, 4H, J = 7.0 Hz), 7.32–7.23 (m, 7H), 7.18 (dd, 2H, J = 7.0, 6.5 Hz), 7.10 (d, 2H, J = 7.0 Hz), 5.91 (ddt, 1H, J = 17.0, 10.5, 5.0 Hz), 5.43 (s, 2H), 5.18 (dd, 1H, J = 17.0, 1.5 Hz), 5.01 (dd, 1H, J = 10.5, 1.5 Hz), 2.92 (d, 2H, J = 5.0 Hz), 2.18 (t, 2H, J = 8.5 Hz), 2.07 (s, 1H, NH), 0.98 (td, 2H, J = 7.5, 7.0 Hz), 0.81–0.75 (m, 2H), 0.64 (t, 3H, J = 7.0 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 148.3, 144.4, 137.0, 135.3, 135.1, 128.8, 128.3, 128.0, 127.7, 126.8, 126.5, 114.9, 66.6, 51.8, 46.6, 29.8, 22.9, 22.7, 13.4; HRMS (ESI) calcd for C₂₉H₃₂N₄Na [M+Na]⁺ 459.2525, found 459.2524.

4-(Azidodiphenylmethyl)-1-benzyl-5-butyl-1*H*-1,2,3-triazole (19i):



The reaction with **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) and additional trimethylsilylazide (80 μ L, 0.600 mmol) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/5) afforded **19i** (72.1 mg, 0.171 mmol, 86%) and **16a** (7.6 mg, 0.019 mmol, 10%).

Colorless oil; R_f value 0.26(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2958, 2102, 1456, 1249,700 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.29–7.37(m, 13H), 7.16(d, 2H, J = 7.0 Hz), 5.49(s, 2H), 2.17(t, 2H, J = 8.5 Hz), 0.99(tq, 2H, J = 7.5, 7.0 Hz), 0.86–0.92(m, 2H), 0.66(t, 3H, J = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 146.1, 141.8, 136.0, 135.0, 128.9, 128.23, 128.15, 128.1, 127.9, 126.9, 72.1, 51.9, 30.2, 22.8, 22.6, 13.4; HRMS (ESI) calcd for C₂₆H₂₆N₆Na [M+Na]⁺ 445.21166, found 445.21159.

3-((1-Benzyl-5-butyl-1*H***-1,2,3-triazol-4-yl)diphenylmethyl)-1***H***-indole (19j) (CCDC 956342):**



The reaction with **15a** (55.1 mg, 0.208 mmol), benzylazide (41.6 mg, 0.313 mmol), $BF_3 \cdot OEt_2$ (47% in ether, 67 µL, 0.25 mmol) in dichloromethane (2 mL) and additional indole (73.3 mg, 0.625 mmol) dissolved in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/5 to 1/2) afforded **19j** (67.7 mg, 0.136 mmol, 65%).

Colorless solid; R_f value 0.23(ethyl acetate / hexane = 1 / 5); m.p. 228.2–229.4 °C; IR (KBr, neat) v_{max} 3222, 2947, 2868, 1492, 1330, 1237, 704 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 8.66(s, 1H), 7.73–7.75(m, 4H), 7.61–7.71(m, 10H), 7.49–7.54(m, 3H), 7.26(m, 2H), 7.18(d, 1H, *J* = 2.5 Hz), 5.91(s, 2H), 2.27(t, 2H, *J* = 8.5 Hz), 1.16(td, 2H, *J* = 7.5, 7.0 Hz), 0.92(t, 3H, *J* = 7.5 Hz), 0.78(m, 2H); ¹³C NMR(126 MHz, CDCl₃) δ 148.3, 145.2, 136.9, 135.9, 135.6, 130.2, 128.8, 128.0, 127.5, 127.3, 126.6, 126.1, 124.6, 122.4, 121.8, 121.3, 119.2, 111.0, 53.8, 51.9, 28.7, 23.6, 22.8, 13.4; LRMS (EI) 496(M⁺, 22%), 354(23), 281(47), 207(54), 129(54), 91(100); HRMS (EI) calcd for C₃₄H₃₂N₄ (M⁺) 496.2627, found 496.2628.

1-Benzyl-5-butyl-4-(1,1-diphenylbut-3-en-1-yl)-1*H*-1,2,3-triazole(19k)



The reaction of **19k** (57.9 mg, 69%) and **16a** (2.1 mg, 3%) from **15a** (53.0 mg, 0.200 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/15 to 1/10 to 1/4).

White solid; R_f value 0.28 (ethyl acetate/hexane = 1/5); m.p. 128.9–130.1 °C; IR (NaCl, neat) v_{max} 3081, 2959, 1466, 907, 700 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.43–7.38(m, 3H), 7.34–7.31 (m, 4H), 7.28–7.24 (m, 6H), 7.20 (t, 2H, *J* = 7.0 Hz), 6.02 (tdd, 1H, *J* = 17.0, 6.5, 7.5 Hz), 5.53 (s, 2H), 4.97–4.92 (m, 2H), 3.63 (d, 2H, *J* = 6.5 Hz), 1.90 (m, 2H), 0.89 (tt, 2H, *J* = 7.5, 7.5 Hz), 0.62 (t, 3H, *J* = 7.5 Hz), 0.51–0.45 (m, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 149.1, 144.6, 136.2, 135.4, 135.3, 128.9, 128.8, 128.1, 127.7, 126.8, 126.2, 117.0, 52.1, 51.8, 47.1, 29.1, 23.1, 22.7, 13.4; HRMS (ESI) calcd for C₂₉H₃₁N₃Na [M+Na]⁺ 444.24157, found 444.23930.

3-(1-benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)-3,3-diphenylpropanal (19l)



The reaction of **191** (45.8 mg, 55%) and **16a** (27.6 mg, 35%) from **15a** (52.4 mg, 0.198 mmol), benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/5 to 1/4).

Colorless oil; R_f value 0.14 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2957, 2858, 1715, 1455, 1023, 700 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 9.84 (t, 1H, *J* = 2.5 Hz), 7.56–7.51 (m, 3H), 7.49–7.41 (m, 6H), 7.35 (d, 2H, *J* = 7.0 Hz), 7.25 (d, 4H, *J* = 8.5 Hz), 5.65 (s, 2H), 3.80 (d, 2H, *J* = 2.5 Hz), 1.96 (t, 2H, *J* = 8.5 Hz), 0.95 (qt, 2H, *J* = 7.5, 7.5 Hz), 0.71 (t, 3H, *J* = 7.5 Hz), 0.59–0.53 (m, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 202.5, 148.0, 143.8, 135.8, 135.1, 129.0, 128.5, 128.4, 128.3, 127.0, 126.99, 55.2, 52.0, 49.7, 29.3, 23.0, 22.7, 13.3; HRMS (ESI) calcd for C₂₈H₂₉N₃ONa [M+Na]⁺ 446.22083, found 446.22063.

ethyl 3-(1-benzyl-5-butyl-1H-1,2,3-triazol-4-yl)-3,3-diphenylpropanoate(19m)



The reaction of **19m** (63.1 mg, 68%) and **16a** (15.3 mg, 20%) from **15a** (52.5 mg, 0.199 mmol) , benzylazide (40.0 mg, 0.300 mmol), BF₃·OEt₂ (47% in ether, 65 μ L, 0.24 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification ethyl acetate/hexane = 1/80 to 1/70 to 1/50 to 1/20 to 1/20 to 1/4).

Colorless oil; R_f value 0.14 (ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2957, 2870, 1742, 1151, 700 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.33–7.28 (m, 3H), 7.25–7.22 (m, 4H), 7.21–7.15 (m, 6H), 7.10 (d, 2H, J = 7.0 Hz), 5.42 (s, 2H), 3.95 (q, 2H, J = 7.0 Hz), 3.78 (s, 2H), 1.84 (t, 2H, J = 8.5 Hz), 1.06 (t, 3H, J = 7.0 Hz), 0.77 (qt, 2H, J = 7.0, 7.5 Hz), 0.51 (t, 3H, J = 7.5 Hz), 0.37–0.30 (m, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 170.8, 148.0, 144.2, 135.4, 135.2, 128.9, 128.7, 128.1, 127.9, 126.8, 126.5,

60.0, 51.9, 50.8, 46.9, 29.0, 23.3, 22.7, 13.9, 13.4; HRMS (ESI) calcd for $C_{30}H_{34}N_3O_2$ [M+H]⁺ 468.26510, found 468.26574.

Bis(1-benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)(phenyl)methanol (21):



The reaction with **20** (55.0 mg, 0.205 mmol), benzylazide (81.9 mg, 0.615 mmol) and TMSOTf (44 μ L, 0.25 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/3 to 2/1) afforded **20** (78.8 mg, 0.147 mmol, 72%).

Colorless oil; R_f value 0.25(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3471, 2957, 1456, 727, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.29–7.34(m, 6H), 7.26(s, 5H), 7.17(d, 4H, J = 6.5 Hz), 5.49(d, 2H, J = 15.5 Hz), 5.43(d, 2H, J = 15.5 Hz), 5.28(s, 1H, OH), 2.35–2.53(m, 4H), 0.99–1.02 (m, 4H), 0.75–0.92 (m, 4H), 0.63(t, 6H); ¹³C NMR(126 MHz, CDCl₃) δ 148.1, 144.6, 135.6, 135.1, 128.9, 128.2, 127.9, 127.7, 127.4, 127.1, 74.2, 51.9, 30.2, 23.0, 22.6, 13.4; HRMS (ESI) calcd for C₃₃H₃₈N₆NaO [M+Na]⁺ 557.3005, found 557.3004.

1-(1-Benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)-1-phenylhept-2-yn-1-ol(22)



The reaction with **20** (77.0 mg, 0.261 mmol), benzylazide (36.5 mg, 0.274 mmol) and TMSOTf (57 μ L, 0.0.313 mmol) in dichloromethane (2 mL) followed by silica gel column chromatography purification (ethyl acetate/hexane = 1/10 to 1/3 to 2/1) afforded **22** (57.9 mg, 55%).

Colorless oil; R_f value 0.29(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 3352, 2957, 1455, 1003, 733, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.83(d, 2H, J = 7.5 Hz), 7.55(m, 6H), 7.35(d, 2H, J = 7.0 Hz), 5.67(d, 1H, J = 15.5 Hz), 5.64 (d, 1H, J = 15.5 Hz), 4.63(s, 1H, OH), 2.59–2.61 (m, 1H), 2.52(td, 2H, J = 1.5, 7.0 Hz), 2.41–2.46(m,

1H), 1.75(tt, 2H, J = 7.5, 7.5 Hz), 1.62(tq, 2H, J = 7.5, 7.5 Hz), 1.24–1.27(m, 3H), 1.10(t, 3H, J = 7.5 Hz), 0.92–1.02(m, 1H), 0.89(t, 3H, J = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 147.8, 143.5, 134.9, 133.7, 128.9, 128.3, 128.1, 128.0, 127.1, 126.5, 88.2, 80.9, 69.7, 52.0, 30.5, 29.9, 22.6, 22.5, 22.0, 18.6, 13.6, 13.5; HRMS (ESI) calcd for C₂₆H₃₁N₃NaO [M+Na]⁺ 424.2365, found 424.2365.

(E)-1-(1-benzyl-5-butyl-1H-1,2,3-triazol-4-yl)-1-phenylhept-1-en-3-one(22')



Colorless oil; R_f value 0.34(ethyl acetate/hexane = 1/5); IR (NaCl, neat) v_{max} 2957, 2871, 1684, 1456, 727, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.32–7.36(m, 6H), 7.22–7.24(m, 2H), 7.14(d, 2H, J = 8.0 Hz), 7.04(s, 1H), 5.45(s, 2H), 2.32(t, 2H, J = 7.5 Hz), 1.84(t, 2H, J = 9.5 Hz), 1.48(tt, 2H, J = 7.5, 7.0 Hz), 1.19(tq, 2H, J = 7.5, 7.5 Hz), 0.83–0.88(m, 4H) , 0.82(t, 3H, J = 7.5 Hz), 0.59(t, 3H, J = 7.0 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 202.0, 144.2, 142.7, 137.4, 137.1, 134.8, 129.0, 128.9, 128.6, 128.4, 128.3, 127.1, 126.5, 52.0, 43.4, 30.8, 26.2, 22.6, 22.4, 22.2, 13.8, 13.3; HRMS (ESI) calcd for C₂₆H₃₁N₃NaO [M+Na]⁺ 424.2365, found 424.2372.

(1-Benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)(5-butyl-1-cinnamyl-1*H*-1,2,3-triazol-4-yl)(phenyl)methanol (23):



To the mixture of **20** (52.2 mg, 0.194 mmol), benzylazide (27.2 mg, 0.204 mmol) in dichloromethane (2.0 mL) was added TMSOTf (42.2 μ L, 0.233 mmol) at -90 °C. After 1 min, cinnamylazide (46.4 mg, 0.291) dissolved in 0.5 mL of dichloromethane was added to the mixture. After five minutes, the reaction was quenched with saturated sodium bicarbonate aqueous solution, and was washed with brine. Drying the organic layer over magnesium sulfate followed by concentration in vacuo and silica gel column

chromatography (ethyl acetate/hexane = 1/20 to 1/10 to 1/3 to 1/2) afforded **23** (58.9 mg, 0.105 mmol, 54%)

Colorless oil; R_f value 0.17(ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 3366, 2956, 2929, 1456, 728, 698 cm⁻¹; ¹H NMR(500 MHz, CDCl₃) δ 7.29–7.36(m, 8H), 7.26–7.28(m, 5H), 7.18(d, 2H, *J* = 7.0 Hz), 6.51(d, 1H, *J* =7.0 Hz), 6.33(dt, 1H, *J* = 15.5, 6.5 Hz), 5.50(d, 2H, *J* = 15.5 Hz), 5.43(d, 2H, *J* = 15.5 Hz), 5.30(s, 1H), 5.03(dd, 2H, *J* = 5.0, 5.0 Hz), 2.59–2.64(m, 1H), 2.47–2.51(m, 2H), 2.38–2.43(m, 1H), 0.79–1.28(m, 8H), 0.69(t, 3H, *J* = 7.5 Hz), 0.63(t, 3H, *J* = 7.5 Hz); ¹³C NMR(126 MHz, CDCl₃) δ 148.1, 144.6, 135.7, 135.5, 135.1, 133.8, 128.9, 128.6, 128.22, 128.18, 127.9, 127.8, 127.4, 127.1, 126.6, 122.8, 74.2, 51.9, 50.3, 30.5, 30.2, 23.1, 23.0, 22.7, 22.6, 13.52, 13.47; HRMS (ESI) calcd for C₃₅H₄₀NaN₆O [M+Na]⁺ 583.3161, found 583.3161.

N-((1-benzyl-5-butyl-1*H*-1,2,3-triazol-4-yl)(5-butyl-1-cinnamyl-1*H*-1,2,3-triazol-4-y l)(phenyl)methyl)prop-2-en-1-amine(24)



To a mixture of **20** (77.6 mg, 0.289 mmol), benzylazide (40.4 mg, 0.304 mmol) in dichloromethane (3.0 mL) was added boron trifluoride diethyl ether complex (101.3 μ L, 0.376 mmol) at -60 °C. After 1 min, cinnamylazide (69.0 mg, 0.434 mmol) dissolved in 0.5 mL of dichloromethane was added to the mixture. After five minutes, allylamine (51.3 μ L, 0.867 mmol) was added at the same temperature, and then warmed up to room temperature. After 30 min, the mixture the reaction was quenched with saturated sodium bicarbonate aqueous solution, and was washed with brine. Then the collected organic layer was dried over magnesium sulfate followed by concentration *in vacuo* and silica gel column chromatography (ethyl acetate/hexane = 1/12 to 1/10 to 1/5 to 1/3) to afford **24** (45.9 mg, 27%).

Colorless oil; R_f value 0.30 (ethyl acetate/hexane = 1/2); IR (NaCl, neat) v_{max} 2957, 2931, 2870, 1456, 1241, 904, 728 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.47 (dd, 2H, J = 7.5, 2.0 Hz), 7.39–7.25 (m, 11H), 7.16 (dd, 2H, J = 6.5, 1.5 Hz), 6.49 (d, 1H, J = 16.0 Hz), 6.33 (td, 1H, J =16.0, 6.5 Hz), 5.92 (ddt, 1H, J = 17.0, 10.0 Hz), 5.50 (s, 2H), 5.21 (dd, 1H, J = 17.0, 2.0 Hz), 5.06–5.02 (m, 3H), 3.01 (d, 2H, J = 4.0 Hz), 2.58–2.29 (m,

4H), 1.21–1.15 (m, 2H), 1.09–1.05 (m, 4H), 0.86–0.78 (m, 5H), 0.70 (t, 3H, J = 7.5 Hz); ¹³C NMR (126 MHz, CDCl₃) δ 147.2, 146.9, 142.9, 136.9, 135.7, 135.6, 135.3, 133.5, 128.8, 128.6, 128.4, 128.2, 128.1, 127.6, 126.9, 126.8, 126.5, 123.0, 114.8, 62.6, 51.8, 50.2, 46.4, 30.2, 29.8, 22.91, 22.87, 22.76, 22.67, 13.61, 13.55; HRMS (ESI) calcd for C₃₈H₄₅NaN₇ [M+Na]⁺ 622.36341, found 622.36220.

General procedure of synthesis of triazole ureas:

A slurry of 30 wt% (based on starting material triazole) of 10% Pd/C in ethanol was added to a stirred solution of 1,4,5-trisubstituted triazoles under nitrogen and the resulting mixture was stirred under an atmosphere of hydrogen gas for 20 h. The reaction mixture was filtered through a plug of celite washing with methanol. The filtrate was evaporated under reduced pressure to give a white solid residue which was used to the next step without further purification. The crude unprotected triazoles (1.2 equiv), pyrrolidine carbonyl chloride **25** (1.0 equiv), and 4-dimethylaminopyridine (0.2 equiv) were dissolved in 5:1 THF/triethylamine (0.1 M based on triazoles), and the mixture was stirred for 10 h at 60 °C. The solvents were removed, and then the crude material was purified by silica gel chromatography to afford triazole ureas.

(4-(Hydroxydiphenylmethyl)-2*H*-1,2,3-triazol-2-yl)(pyrrolidin-1-yl)methanone (26a)



17.9 mg (30%) for two steps from **16k** (70.0 mg, 0.206 mmol) [silica gel chromatography (hexane/ethyl acetate = 20:1 to 10:1 to 5/1 to 3/1 to 2/1 to 1/1 to ethyl acetate)].

White solid; R_f value 0.23 (ethyl acetate/hexane = 1/2); m.p. 156.7–157.7 °C; IR (NaCl, neat) v_{max} 3363, 2917, 1697, 1264, 1058, 699 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.56 (s, 1H), 7.33–7.29 (m, 10H), 3.76 (t, 2H, J = 6.5 Hz), 3.70 (t, 2H, J = 6.5 Hz), 1.96–1.92 (m, 4H); ¹H NMR (500 MHz, CD₂Cl₂) δ 7.61 (s, 1H), 7.35–7.27 (m, 10H), 3.72 (t, 2H, J = 6.0 Hz), 3.63 (t, 2H, J = 6.0 Hz), 1.95–1.90 (m, 4H); ¹H NMR (500 MHz, CD₃OD) δ 7.84 (s, 1H), 7.37–7.23 (m, 10H), 3.73 (t, 2H, J = 6.0 Hz), 3.64 (t, 2H, J = 6.0 Hz), 1.94–1.90 (m, 4H); ¹³C NMR (126 MHz, CDCl₃) δ 155.6, 147.7, 144.9, 135.3, 128.2, 127.8, 127.1, 77.2, 50.1, 48.7, 26.4, 24.0; ¹³C NMR (126 MHz, CD₂Cl₂) δ 155.8, 147.9, 145.5, 135.4, 128.4, 128.1, 127.4, 77.4, 50.4, 48.9, 26.7, 24.4; ¹³C NMR (126 MHz,

CD₃OD) δ 158.0, 149.6, 147.1, 136.7, 128.9, 128.5, 128.4, 78.0, 51.5, 49.8, 27.4, 25.0; HRMS (ESI) calcd for C₂₀H₂₀N₄O₂Na [M+Na]⁺ 371.14839, found 371.14822.

(4-Butyl-5-(hydroxydiphenylmethyl)-2*H*-1,2,3-triazol-2-yl)(pyrrolidin-1-yl)methan one (26b)



121.7mg (0.300mmol, 80%) for two steps from **16a** (220.1 mg, 0.554mmol) [silica gel chromatography (hexane/ethyl acetate = 3:1 to 1:1, dichloromethane/methanol = 40:1)]. White solid; Rf value 0.45 (ethyl acetate/hexane = 1/2); m.p. 168.7–170.4 °C; IR (NaCl, neat) v_{max} 3333, 2958, 2871, 1694, 1448, 1418, 1264, 1171, 941 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.27–7.19 (m, 10H), 3.67 (s, 1H), 3.64–3.58 (m, 4H), 2.20 (t, 2H, *J* = 8.0 Hz), 1.83 (br, 4H), 1.27 (tt, 2H, *J* = 7.5, 8.0 Hz), 1.07 (tt, 2H, *J* = 7.5, 7.5 Hz), 0.66 (t, 3H, *J* = 7.5 Hz); 13C NMR (126 MHz, CDCl₃) δ 152.2, 148.9, 147.9, 144.4, 128.0, 127.8, 127.5, 77.9, 50.1, 48.6, 30.3, 26.5, 25.5, 24.0, 22.4, 13.6; HRMS (ESI) calcd for C₂₄H₂₈N₄O₂Na [M+Na]+ 427.21099, found 427.21052.

(4-Cyclohexyl-5-(hydroxydiphenylmethyl)-2*H*-1,2,3-triazol-2-yl)(pyrrolidin-1-yl)m ethanone (26c)



152.0 mg (0.353 mmol, 67%) for two steps from **16i** (260.0 mg, 0.614 mmol) [silica gel chromatography (hexane/ethyl acetate = 5:1 to 3:1 to 1:1, dichloromethane/methanol = 40:1)].

White solid; R_f value 0.26 (ethyl acetate/hexane = 1/2); m.p. 226.8–227.0 °C; IR (NaCl, neat) v_{max} 3334, 3061, 2925, 2854, 1694, 1423, 1340, 1260 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.34–7.26 (m, 10H), 3.73 (s, 1H), 3.72 (t, 2H, J = 7.0 Hz), 3.68 (t, 2H, J = 7.0Hz), 2.17–2.11 (m, 1H), 1.91 (m, 4H), 1.69–1.53 (m, 3H), 1.46–1.37 (m, 4H),

1.16–1.10 (m, 1H), 0.96–0.89 (m, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 153.1, 151.5, 148.0, 144.5, 128.0, 127.8, 127.7, 78.0, 50.1, 48.6, 35.2, 26.5, 26.3, 25.7, 24.0; HRMS (ESI) calcd for C₂₆H₃₀N₄O₂Na [M+Na]⁺ 453.22664, found 453.22670.

(4-(((tert-Butyldimethylsilyl)oxy)methyl)-5-(hydroxydiphenylmethyl)-2*H*-1,2,3-tria zol-2-yl)(pyrrolidin-1-yl)methanone (26d, CCDC 1014155)



39.1 mg (0.079 mmol, 33%) for two steps from **16m** (139.6 mg, 0.287 mmol) [silica gel chromatography (hexane/ethyl acetate = 5:1 to 3:1 to 1:1)].

White solid; R_f value 0.2 (ethyl acetate/hexane = 1/2); m.p. 146.8–147.9 °C; IR (NaCl, neat) v_{max} 3403, 2928, 2884, 1717, 1410, 1259, 1059, 993 cm⁻¹; ¹H NMR (500 MHz, CD₂Cl₂) δ 7.38–7.36 (m, 4H), 7.33–7.27 (m, 6H), 5.68 (s, 1H), 4.65 (s, 2H), 3.59 (t, 2H, J = 6.5 Hz), 3.54 (t, 2H, J = 6.5 Hz), 1.90–1.84 (m, 4H), 0.82 (s, 9H), -0.02(s, 6H); ¹³C NMR (126 MHz, CD₂Cl₂) δ 153.9, 147.9, 146.5, 145.3, 128.1, 127.7, 127.5, 76.9, 58.4, 50.4, 48.9, 26.7, 25.8, 24.4, 18.4, -5.7; HRMS (ESI) calcd for C₂₇H₃₆N₄O₃SiNa [M+Na]⁺ 515.24544, found 515.24415.

3-(5-(Hydroxydiphenylmethyl)-2-(pyrrolidine-1-carbonyl)-2*H*-1,2,3-triazol-4-yl)pr opyl benzoate (26e, CCDC 1014154)



117.7 mg (0.231 mmol, 40%) for two steps from **160** (349.3 mg, 0.694 mmol) [silica gel chromatography (hexane/ethyl acetate = 5:1 to 3:1 to 2:1, dichloromethane / methanol = 60:1)].

White solid; R_f value 0.2 (ethyl acetate/hexane = 1/2); m.p. 184.7–185.5 °C; IR (NaCl, neat) v_{max} 3403, 2928, 2884, 1717, 1410, 1259, 1059, 993 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.92 (dd, 2H, J = 8.0, 1.0 Hz), 7.53 (td, 1H, J = 7.0, 1.0 Hz), 7.39 (tt, 2H, J = 7.0, 8.0 Hz), 7.30–7.23 (m, 10H), 4.18 (t, 2H, J = 5.5 Hz), 3.63 (s, 1H), 3.64–3.60 (m, 4H), 2.52 (t, 2H, J = 7.5 Hz), 1.97–1.81 (m, 2H), 1.89–1.84 (m, 4H); ¹³C NMR (126 MHz, CDCl₃) δ 166.5, 152.2, 147.81, 147.76, 144.3, 132.9, 130.1, 129.5, 128.3, 128.1, 127.9, 127.4, 77.9, 64.2, 50.1, 48.6, 27.1, 26.5, 24.0, 22.7; HRMS (ESI) calcd for C₃₀H₃₀N₄O₄Na [M+Na]⁺ 533.21647, found 533.21563.

(4-(Hydroxydiphenylmethyl)-5-(trimethylsilyl)-2*H*-1,2,3-triazol-2-yl)(pyrrolidin-1-yl)methanone (26f)



82.1 mg (0.195 mmol, 58%) for two steps from **16r** (164.7 mg, 0.404 mmol) [silica gel chromatography (hexane/ethyl acetate = 5:1 to 3:1 to 2:1 to 1:1)].

White solid; R_f value 0.34 (ethyl acetate/hexane = 1/2); m.p. 168.7–169.4 °C; IR (NaCl, neat) v_{max} 3350, 2959, 1697, 1252, 1048, 928, 761 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.32–7.28 (m, 10H), 3.66 (t, 2H, *J* = 7.0 Hz), 3.58 (t, 2H, *J* = 7.0 Hz), 2.88 (s, 1H), 1.92–1.84 (m, 4H), 0.22 (s, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 160.1, 148.6, 148.2, 145.6, 128.0, 127.7, 127.4, 78.7, 49.9, 48.4, 26.4, 24.1, –0.4; HRMS (ESI) calcd for C₂₃H₂₈N₄O₂SiNa [M+Na]⁺ 443.18792, found 443.18773.

Chapter 5 Supporting Information

5.2 NMR spectra









OH





ΟН





ŌН

MeO

127



ŌН

























QMe QH
































































Z:\ZHANG HUAN\NMR\huan_4_102_1H.1 7.633 7.617 7.555 7.555 7.555 7.380 7.380 7.365 7.365 7.365 7.261 7.255 7.255 7.255 7.255 7.255 7.255 4.152 4.138 4.123 2.985 2.969 2.952 2.952 2.068 2.068 2.068 2.053 2.053 2.038 1.718 V. , 2.00 0.95 0.95 0.91 0.91 ر 1.97 \$ 2.04 100



PPM

ó

2

Z:\ZHANG HUAN\NMR\huan_4_102_C13.1









QMe QH































































Z:\ZHANG HUAN\NMR\huan_6_72_2_Proton-1-1.jdf









2:\ZHANG HUAN\NMR\huan_6_66_1_Proton-1-1.jdf



 DFILE
 huan_6_66_1_Proton-1-1.jdf

 COMNT
 single_pulse

 DATIM
 2013-05-21 22:51:08

 OBNUC
 1H

 EXMOD
 proton.jxp

 OBFRQ
 500.16 MHz

 OBSET
 2.41 KHz

 OBFIN
 6.01 Hz

 POINT
 16384

 FREQU
 9384.38 Hz

 SCANS
 8

 ACQTM
 1.7459 sec

 PD
 5.0000 sec

 PW1
 6.22 usec

 IRNUC
 1H

 CTEMP
 18.2 c

 SLVNT
 CDCL3

 EXREF
 7.26 ppm

 BF
 0.12 Hz

34

Z:\ZHANG HUAN\NMR\huan_6_66_1_Carbon-1-1.jdf


























Ph OTBS $h \rightarrow = -$

Ph









Z:\ZHANG HUAN\NMR\huan_6_96_1b_Proton-1-1.jdf







1 1

Т



ILE	HT5-59-1 prot	con-1-1.jdf
MNT	HT5-59-1 cour	oling TMSacetylene
TIM	2012-12-19 20	33:23
NUC	1H	
MOD	proton.jxp	
FRQ	500.16	MHz
SET	2.41	KHz
FIN	6.01	Hz
INT	16384	
EQU	9384.38	Hz
ANS	16	
QTM	1.7459	sec
)	5.0000	sec
1	6.22	usec
NUC	1H	
EMP	13.8	с
VNT	CDCL3	
REF	7.26	ppm
	0.10	Hz
AIN	40	

DFILE	HT5-59-1 13C-1-1.jdf
COMNT	HT5-59-1 13C NMR
DATIM	2012-12-19 16:38:55
OBNUC	13C
EXMOD	carbon.jxp
OBFRQ	125.77 MHz
OBSET	7.87 KHz
OBFIN	4.21 Hz
POINT	32767
FREQU	39308.18 Hz
SCANS	114
ACQTM	0.8336 sec
PD	2.0000 sec
PW1	3.12 usec
IRNUC	1H
CTEMP	14.2 c
SLVNT	CDCL3
EXREF	77.00 ppm
BF	0.10 Hz
RGAIN	52

I I













































































Z:\ZHANG HUAN\NMR\huan_6_61_1_Proton-2-1.jdf








































,∽^N`N∽Bn

Ph



200

150

100



13C carbon.jxp 125.77 MHz 7.87 KHz 4.21 Hz 32767 39308.18 Hz 591 591 0.8336 sec 2.0000 sec 3.12 usec IRNUC 1H 19.3 c CTEMP SLVNT CDCL3 77.00 ppm 0.12 Hz 58 EXREF BF RGAIN

PPM

8









Z:\ZHANG HUAN\NMR\huan_6_143_1_Proton-1-1.jdf













Z:\ZHANG HUAN\NMR\huan_7_99_1b_Proton-1-1.jdf





















Z:\ZHANG HUAN\NMR\huan_6_115_1_Proton-1-1.jdf













Z:\ZHANG HUAN\NMR\huan_7_126_1_Proton-1-1.jdf















Z:\ZHANG HUAN\NMR\huan_7_127_1_Proton-1-1.jdf

















Provided ¹H and ¹³C NMR Spectra of 26b recorded in CD₃OD.





Chapter 6 Conclusion

The applications of triazoles produced azide-alkyne cycloadditions range from drug discovery, chemical biology, materials science, development of sensors, polymer chemistry, to other molecular science. The copper-catalyzed azide-alkyne cycloaddition (CuAAC) to produce these functional cores is the most famous method, which is arguably the most widely utilized in "click chemistry". Because of its various efficiencies in research area, many groups have recently reported the rapid, mild, and Cu-free triazolations, such as using strained alkyne and substituent activation. However, CuAAC is limited to mostly terminal alkynes and the toxicity of copper salts limits the utility for *in vivo* applications. In addition, most of the reported methods including classic Huisgen reactions require room or high temperature with a long reaction time. With three zwiterionic nitrogens, organic azides not only can be used as 1,3-dipolars for cycloadditions but also electrophiles and nucleophiles.

In chapter 2, based on the previous results for synthesis of cylclic unsaturated imines with allyl cations and organic azides, I investigated the intramolecular [3+2] cycloaddition of benzyl propargyl alcohols with azide under acid conditions to afford triazoles. According to these successful results, intermolecular triazole formations were examined. Through optimization, the internal reactions were also demonstrated with organic azide in the present of TMSOTf, and the desired transformation was preceded in good to excellent yield even at -90 °C. With optimized condition, various fully substituted functional triazoles were successfully produced within 10 min (Scheme 52). The reaction results indicate that: 1) for R^1 and R^2 group, electron deficient aryl was found to be effective as diphenyl group; 2) both terminal and internal alkyne can afford the desired triazole product; 3) strong nucleophilic azides were favorite to this transformation; 4) this cyclization reaction proceed regioselectively to produce triazoles was generated as single isomer; 5) the transformation can be carried out not only at ambient but also low temperature, even at -90 °C; 6) the presence of methylenetriazolium was indicated.



Scheme 52. Intra- and inter- molecular azide-alkyne cycloaddition reactions.

In chapter 3, to develop the efficiency of this carbocation-mediated azide-alkyne cycloaddition reaction, I conducted various types of multicomponent



coupling reactions in one-pot. According to the proposed mechanism, not only a hydroxyl group, but also other functional groups could be introduced by changing quenching methods. Investigation of the origin of the hydroxyl group to make sure the hydroxyl groups source indicated that BF₃·OEt₂ is a suitable acid for the substitution of triazolium intermediates with additional nucleophiles, such as alcohols, amines, thiol, carbon nucleophiles and so on. Moreover, controlling of the amount of the reagents and the reactivity of substrates, four-component coupling reaction was achieved by double [3+2] reaction followed by substitution with nucleophile with dialkyne substrates (Figure 6).

In chapter 4, to further demonstrate the efficiency of the method toward bioactive molecule synthesis, I carried out the synthesis of triazole urea reported as a serine hydrolase inhibitor along with its 5-substituted derivatives



(Figure 7). Serine hydrolases have been used as the targets of clinical drugs to treat

various diseases such as diabetes and Alzheimer's disease. The 1,4-disustituted and 1,4,5 trisubstituted *N*-benzyltriazoles prepared by the developed method were deprotected followed by carbamoylation to afford serine hydrolase inhibitor triazole urea and its 5-substituted derivatives, which were difficult to prepared by traditional methods. Moreover, I also suggested the errata and incorrect assignment of analycal data of the reported inhibitor compound.

I successfully developed the rapid regioselective synthesis of fully substituted 1,2,3-triazoles mediated by propargyl cations formed from propargyl alcohols, accepting both terminal and internal alkynes at low and ambient temperature in good yields. Various types of one-pot multicomponent coupling reactions, including double triazolations and functionalizations of triazoles with additional nucleophiles, were demonstrated. The synthesized triazoles were successfully converted to serine hydrolase inhibitor triazole urea and its derivatives. Our method can provide new preparation method of highly substituted triazoles and exploration of their uses in synthetic organic chemistry and pharmaceutical research.

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LIST OF PUBLICATIONS

学位論文の主たる部分を公表した論文

(1) <u>Huan Zhang</u>, Hiroki Tanimoto, Tsumoru Morimoto, Yasuhiro Nishiyama, Kiyomi Kakiuchi, "Regioselective Rapid Synthesis of Fully-substituted 1,2,3-Triazoles Mediated by Propargyl Cations", *Org. Lett.* **2013**, *15*, 5222–5225.

(2) <u>Huan Zhang</u>, Hiroki Tanimoto, Tsumoru Morimoto, Yasuhiro Nishiyama, Kiyomi Kakiuchi, "Acid-mediated Synthesis of Fully Substituted 1,2,3-Triazoles: Multicomponent Coupling Reactions, Mechanistic Study, Synthesis of Serine Hydrolase Inhibitor and Its Derivatives", *Tetrahedron* **2014**, *70*, 9828–9835.

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